Bangladesh.

RUCL Institutional Repository

http://rulrepository.ru.ac.bd

Department of Botany

MPhil Thesis

1998

Studies on the Nutrient Status and Plankton Succession in Four Ponds of Rajshahi Pisciculture Hatchery

Islam, Mohd. Azizul

University of Rajshahi

http://rulrepository.ru.ac.bd/handle/123456789/925

Copyright to the University of Rajshahi. All rights reserved. Downloaded from RUCL Institutional Repository.

STUDIES ON THE NUTRIENT STATUS AND PLANKTON SUCCESSION IN FOUR PONDS OF RAJSHAHI PISCICULTURE HATCHERY



MOHD. AZIZUL ISLAM B.Sc. (Hons.), M.Sc.

A THESIS SUBMITTED FOR THE DEGREE
OF MASTER OF PHILOSOPHY
TO THE DEPARTMENT OF BOTANY
UNIVERSITY OF RAJSHAHI
BANGLADESH

December, 1998.

PHYCOLOGY &

LIMNOLOGY LAB.

DEPARTMENT OF BOTANY
UNIVERSITY OF RAJSHAHI

DECLARATION

I do hereby declare that whole of the work submitted as a thesis for the degree of Master of Philosophy in Botany of the University of Rajshahi, as the result of my own investigation.

M. Zama

Prof. Dr. M. Zaman Supervisor

MOHD. AZIZUL ISLAM

Candidate

MOHD. Azizer Islam

CERTIFICATE

I do hereby certify that the work embodied in this thesis has not already been submitted in substance for any degree and has not been concurrently submitted in candidature for any degree.

MOHD. AZIZUL ISLAM
Candidate

CERTIFICATE

Certified that the thesis entitled "Studies On The Nutrient Status and Plankton Succession in Four Ponds of Rajshahi Pisciculture Hatchery" is suitable for submission for the Degree of Master of Philosophy of the Rajshahi University, as to the style and content.

Dr. M.Zaman

M. Zama

Professor of Botany Rajshahi University

Supervisor.

ACKNOWLEDGMENTS

The author wishes to express his deepest Sense of gratitude and indebtedness to Prof. Dr. M. Zaman, Dpt. Botany, University of Rajshahi Under whose Valuable guidance, constructive criticism and encouragement, this work was carried out.

The author extends his heartfelt thanks to Prof. Asheque Ahmed, chairman, Dpt. Botany, University of Rajshahi for Kindly providing the laboratory facilities for this work.

The author is indebted to Prof. A. T. M. N. Zaman, Prof Dr. M. S. Alam and Prof. Dr. G. Kabir Dpt. Botany, University, Rajshahi for their helps, co-operation, and constructive criticism and encouragement during the study period.

The author wishes to express his gratitude to Mrs. S. Naz, Lecturer, Dpt. Botany, University of Rajshahi for her help, co-operation, constructive criticism and encouragement for this work.

The author also wishes to express his sincere thanks to research fellows Miss. N. Moshfaqua, A. H. Choudhury, M. A. Hasan, M. Fakruzzaman, of the Phycology and Limnology Laboratory for their active Co-operation, help and encouragement.

The author wishes to express his deepest sense of gratitude and indebtedness to Prof. Dr. M. Altaf Hossain -2, Dpt. Zoology, University of Rajshahi for his active help and encouragement during the course of this work.

The author is also indebted to Prof. Dr. M. A. Mannan and Prof. Dr. M. A. R. Khan, Dpt. Zoology, University of Rajshahi for their suggestion and encouragement.

The author is glad to acknowledge to Prof. Dr. R. J. M. S. Alam, Head of the Dpt. Botany, Rajshahi College for his encouragement during the study.

The author wishes to express his gratitude to M. A. Hossain, lecturer, Rajshahi HSC Teachers Training Institution for active Cooperation and financial help.

The author wishes to express his gratitude to D. M. Mohsin Ali, Ex-manager, Rajshahi fish farm for his kindly providing facilities and some logistic support.

The author extends his gratitude to M. M. Hussain, present manager. Rajshahi fish farm for his active Co-operation.

The author also wishes to express his gratitude to M. S. I. Khan, Deputy Director, Dpt. Fisheries, Rajshahi for his encouragement and interest in the work.

The author is also due to M. N. Islam, M. A. B. Siddique, M. S. Islam, M. Shamshuddin, Rajshahi fish farm and M. S. Choudhury, M. F. Rahman and M. A. Sukur, Dpt. Botany, Rajshahi College for their Co-oparations.

The author is indebted to his wife Mrs. M. J. A. Khanam for her encouragement, active help in preparation of the manuscript during the course of this study.

Sincere gratitude is due to M. S. Islam, lecturer, Dpt. of statistics, University of Rajshahi for some statistical analysis.

The cooparation of Agrazatra Computer is duly acknowledged.

•

CONTENTS

Page No.

DECLARATION	
CERTIFICATE	
CERTIFICATE	
ACKONOWLEDGMENT	
CONTENT	i-v
ABSTRACT	1-2
CHAPTER-1 INTRODUCTION	3-8
Intruduction	4
Review of Literature	9-13
CHAPTER-2; MATERIALS AND METHODS	14-26
Description Of Study Ponds	15
Map of the study area	16
Physical Conditions	17
Air and water temperature	17
Transparancy of water	18
Average depth	18
Chemicical conditions	19
Dissolved oxygen content &% saturation 02 in water	19
Carbonate alkalinity	20

Bicarbonate alkalinity	20
P ^{II}	20
Electric conductivity	20-21
Oxidation- Reduction	21
Oxidation -Reduction Index	21
Total dissolved solids	21
Free CO ₂	21-22
Biochemical Oxygen Demand	22
Chemical Oxygen Demand	22
Total hardness	23
Phosphate	23
Mounting Fluid	23-24
Plankton analysis	24-25
Primary Productivity	25-26
Ammonium-nitrogen	26
CHAPTER-3:OBSERVATIONS AND RESULTS	27-60
Physical Factors	28
Air temperature	28
Average depth	28
Secchi disk depth	29-30
Total dissolved solids	30

Chemical Conditions	30
E. cond.	30
Chemical condition	31
P ^H	31-32
Free CO ₂	32-33
Carbonate alkalinity	33-34
Bicabonate alkalinity	34-35
Total hardness	35-36
Chloride	36-37
Dissolved Oxygen	37-38
% Saturation 02	38-39
Biochemical Oxygen Demand	40-41
Chemical Oxygen Demand	41-42
Ammonium-nitrogen	42
Phosphate	43-44
Redox potential	44-45
Redox index	45
Gross Primary Productivity	
Net Primary Productivity	
Zooplankton	
Phytoplankton	

CHAPTER-4: DISCUSSION	61-97
Temperature	62
Average depth	63-64
Secchi disk depth	64-65
Total dissolved solids	65-66
Electric conductivity	66-67
P ^H	67-69
Free CO ₂	69-71
Cabonate alkalinity	71-72
Bicabonate alkalinity	72-74
Total hardness	74-76
Chloride	76-78
Dissolved Oxygen	78-82
Biochemical Oxygen demand	82-84
Chemical Oxygen demand	
Ammonum-nitrogen	
Phosphate	
Oxidation-reduction and	
Oxidation reduction index	88-89
Zooplankton	
Phytoplankton	

CHAPTER-5:REFERENCES98-133
CHAPTER-6 : Appendix
Physico-chemical condition
Seasonal variation of physico-chemical & bio-cond 138-141
Abundance of Zooplank142-148
S. Abun. of diff. gen. & groups of Z.P149-153
S. abun of diff. of groups of Z.P. & T. abun153-154
T. No. of gen. of diff group of Zooplank & their P.C
Occurrance of groups of zooplankton156
Abun. of Phytoplankton157-165
Seasonal abun. of diff. gen. of Phyto166-169
Seasonal abun. of diff. groups of Phyto
T.no. of gen. of phyto. belonging to diff. groups of algae 172
Yearly abun. of diff. groups of phyto
Seasonal abun. of P.C. of Phyto. & zoo
Com. Seasonal abun. of phyto. & zoo. (p ₁ , p ₂ , p ₃ and p ₄) 173

. !

ABSTRACT

Physico-Chemical and Biologycal investigation of four selected ponds of Rajshahi Pisciculture Hatchery was carried out for twelve months from May ,1994 to April,1995. Fortnightly Sampling was performed in the study ponds. The Physico-chemical parameter and plankton abudance with their seasonal succession was studied. The range of phyico-chemical and Biological conditions of the study ponds were considered separately.

The range of the water temperature varied from 31.8°c, in pond-1, 18.4-32.8°c in pond-2,19.0-32.5°c in pond-3 and 19.1-32.5°c in pond-4 Average depth-0.9-1.2 M. in pond-1, 1.4-1.8 M. in pond-2, 1.7-2.3 M.in pond-3 and 1.4-1.9 M. in pond-4; TDS-353-488.3 mg/l in pond-1, 383-499 mg/l in pond-2, 381.6-450.1 mg/l in pond-3 and 351.3-404.6 mg/l in pond-4; E. cond-543-753 µmoh/cm in pond-1, 422-767µ moh/cm in pond-2, 597-690 μ moh/cm in pond-3, and 541-724 μ moh/cm in pond-4: p^H 7.7-8.7 in pond-1, 7.8-8.9 in pond-2, 8.2-8.9 in pond-3 and 8.1-8.9 in pond-4; Free $C0_2$ -2.8-34.7 mg/l in pond-1, 4.3-32.5 mg/l in pond-2, 1-4.6 mg/l in pond-3 and 0-5.7 mg/l in pond-4; CO_3 -alk. 0-34.5 mg/l in pond-1, 0-15.2 mg/l in pond-2, 1.2-8 mg/l in pond-3 and 1.8-7.7 mg/l in pond-4; HCO₃-alk. 21-403 mg/l in pond-1, 18.7-356.7 mg/l in pond-2, 15.1-445.2 mg/l in pond-3 and 15.2-366.6 mg/l in pond-4; Total hardness-200-467 mg/l in pond-1 ,214.2-609.2 mg/l in pond-2, 238.7-608.8 mg/l in pond-3 and

197.8-469.3 mg/l in pond-4; Chloride 99-143 mg/l in pond-1, 103.6-191.7 mg/l in pond-2, 103.4-201.3 mg/l in pond-3 and 93.4-142 mg/l in pond-4; DO 4-7.6 mg/l in pond-1, 5-8.4 mg/l in pond-2, 5.7-9.6 mg/l in pond-3 and 6.5-10.6 mg/l in pond-4; BOD₅-2.7-12 mg/l in pond-1, 2.5-9.6 mg/l in pond-2, 3.2-13.6 mg/l in pond-3 and 8.6-16.8 mg/l in pond-4; COD-7.2-22 mg/l in pond-1, 6.4-20.7 mg/l in pond-2, 8.7-26.3 mg/l in pond-3 and 17.3-37.2 mg/l in pond-4; Ammonium-nitrogen-0-0.13 mg/l in pond-1, 0-0.2 mg/l in pond-2, 0-0.13 mg/l in pond-3 and 0-0.2 mg/l in pond-4, PO₄.0.1-0.7 mg/l in pond-1, 0.03-0.2 mg/l in pond-2, 0.01-0.3 mg/l in pond-3 and 0.02-0.4 mg/l in pond-4; Eh-0.25-0.35 mv. in pond-1, 0.22-0.35 mv. in pond-2, 0.23-0.31 mv. in pond-3 and 0.23-0.32 mv in pond-4 and rH₂-25.8-27.8 in pond-1, 25.3-27.6 in pond-2, 25-27 in pond-3 and 15.8-27.2 in pond-4.

In all 33 genera of phytoplankton were indentified of which 12 genera belonged to Chlorophyceae, 7 to Cyanophyceae, 10 to Bacillariophyceae, 3 to Euglenophyceae and 1 to Dinophyceae. In all 38 genera of Zooplankton were recorded of which 8 belonged to Copepoda, 7 to Cladocera and 12 to Rotifera. The quantitative and qualitative analysis of physico-chemical factors of waters and abundance of phytoplankton and Zooplankton reveal that all the ponds under study are eutrophic in nature.

INTRODUCTION



INTRODUCTION

Bangladesh lies between 20.30° and 6.45° north latitude and between 88.00° east longitude. It is a humid-tropical, low lying flat country abounding in large number of rivers, haors, beels, flood plain wetlands, khals and artificial ponds . Many of these water bodies are lentic and lotic water systems differing ecosystemologically with respect to their biotic and abiotic components which lead to overall productivity. During the recent few years captive or pond fishery has become popular in the country. In this sector Bangladesh has immense potential for increased food production in her vast inland water bodies especially the ponds. Utilization of this production potential is most essential to meet the protein requirement of the country. The total area of culturable water bodies in the country is estimated to be 150,000 ha. and 50 percent of these are being utilized for aquaculture (Nuruzzaman, 1992). FAO and world Bank indicated proper culture-based fishery and efficient that through management, an increased production of about 300,000 ton/year is possible in Bangladesh.

Ponds and tanks occupy 360,000 acres or 10 percent of total inland water area in Bangladesh. Bangladesh Bureau of Statistics states that the total number of ponds in the country are 12,888,222 covering an area of 362,980 acres. According to SPARSO (1994), the total number is estimated to be 13.76 lacs

covering 1.44 lac hector. However, there is dearth of detailed statistics on pond fishery in the country. Number of culturable ponds and other types of water bodies is high in Rajshahi district compared to other regions of the country. Side by side with the Government managed pisciculture hatcheries, a large number of such organizations are in function under private management in the Rajshahi region.

Rajshahi City Corporation covers an area of about 299.43 sq.km. where the total cultivable land is about 1790.20 hector and the total span of water in terms of ponds is about 320 hector (Naz, 1995). A total number of 818 ponds of various sizes are present in the Rajshahi City Corporation area. Of these 818 ponds 676 ponds are under pisciculture and 142 are derelict (Naz, 1995). The annual production of fish in these ponds is estimated at about 439.15 m.t. per year. Of these total number of 818 ponds, a large number of ponds are within the jurisdiction of the Government. Rajshahi Pisciculture hatchery, situated at the heart of the city, is a Governmental organization having a good number of exotic and indigenous origin and for raising fish hatchlings to be sold to the fish farmers. The hatchery ponds are under intensive cultivation frequently charged heavily with organic and inorganic manures and fish feed indiscriminately. Toxic chemicals including some chlorinated hydrocarbons are used at the time of pond preparation prior to stocking. Such actions are taken to eliminate non-culturable fishes. This has resulted in the

elimination of a large number of inland fresh water non-culturable fishes from the area. Indiscriminate use of organic and inorganic manure along with frequent use of supplementary feed and toxic chemicals have a deleterious effect on the pond ecosystem as a whole. The hatchery ponds were found to be without any macrophytes and visual algal blooms were absent, which warrant thorough investigation.

The applied aspect of limnological studies involves the evaluation of fresh water resources delimited by the quantity and quality of the flora and fauna which a water body can support. Plankton including phytoplankton and the zooplankton regarded as the best index of biological productivity and the nature of the habitat (Welch, 1948; Hutchinson, 1967; Philipose, 1959). In the culturable ponds, the production of plankton, especially the right type of plankton, is essential for successful rearing of fish and higher fish production. Among many problems involving the fish cultivation in ponds, scientific ways of management of pond water for fish augmentation and perpetual use of the culturable water bodies as renewable natural resources are considered very important. Limnological knowledge and their proper application for better fish yield and management of the are pre-requisites for sustainable fish-sectoral development and maintenance of the eutrophic nature of the pond ecosystem. As a matter of fact, fish farming is a practical application of limnology and aquacultural biology. It is therefore,

important to study the interrelationships between the and plankton and also between physico-chemical aspects phytoplankton and zooplankton and the effect of cultural practices on this biota. Although there are a lot of phycological and limnological works carried out elsewhere in the country, Rajshahi region is not represented as yet, and this region still remain a virgin field for such investigations. Now-a days smaller water bodies are regarded as one of the best ecosystems for conducting researches on biomanipulation with a view to boosting up of the aquatic terminal product in terms of fish for human needs. Although, Rajshahi pisciculture hatchery is a suitable organization for such activities of biomanipulation, yet it is known to suffer management of the pond ecosystem improper from limnolgical and ecological point of view. Considering this fact, the present investigation was undertaken.

Objectives of the work.

- 1. To know the physical condition of the habitat including air and water temperature, average depth, limit of visibility or penetration of light in water, TDS and electric conductivity of water of the study waters.
- 2. To know the major chemical conditions including P^H, free CO₂, CO₃, and HCO₃ alkalinities, total hardness, Chloride, DO, % saturation of O₂, BOD, COD, NH₃ N, PO₄, Oxidation-reduction (Eh) and Oxidation-reduction index (rH₂).

- 3. To study the monthly, seasonal and annual abundance and percentage composition of the phyto-and zooplankton and their seasonal succession in the study ponds.
- 4. To study the interrelationships between the physico-chemical variables and between the physico-chemical variables and the phyto and zooplankton.
- 5. To study the primary productivity in the study ponds.
- 6. To evaluate the trophic nature of the study ponds.

Review of Literature

Bangladesh abound with large number of natural bodies of inland water known as rivers, khals, beels, haors, baors, marshes and ponds. Many of these lentic and lotic water systems are permanent and seasonal water bodies.

Bangladesh had been almost a Virgin field for phycological studies till the first half of the present century. A few workers made some contributions in this field during the early thirties. Notable among of them were Benerjee (1936) who worked on the Myxophyceae of water logged conditions of lower Bengal covering the districts Mymensingh, Dhaka, Faridpur and Jessore of Bangladesh and 24-parganas of West Bengal of India. Kundu (1934, 1935 and 1938) worked on the characeae of Rajshahi district of Bangladesh. During the last twenty years, there have been a tremendous progress in the study of the algal taxonomy in Bangladesh. Islam and his students of University of Dhaka have made valuable and original contributions to the field of algal taxonomy and ecology of the inland waters in the country. Among the large number of their works, the notable publications on the algal taxonomy and ecology of the inland waters of Bangladesh are Islam (1960, 1962, 1963, 1965, 1969, 1970, 1972, 1973a, 1973b, 1974a, 1975, 1976, 1980 and 1982), Islam and Ahia (1964), Islam and Sarma (1964a), Islam and Sarma (1964b), Islam and Khatun (1966), Islam and Nahar (1967), Islam and Sarma (1965), Islam and Sarma (1968), Islam and Begum (1970), Islam and Zaman (1974), Islam et al (1978), Islam and Sarma (1976), Islam et al (1978), Islam et al (1978), Islam et al (1978), Islam and Hossain (1978), Islam and Uddin (1973), Islam and Uddin (1978a), Islam and Uddin (1978b), Islam and Haroon (1978), Islam and Aziz (1975), Islam and aziz (1979), Islam and Alam (1980), Islam and Zaman (1981), Islam

and Haroon (1980). In these papers, the authors have described the algal flora of the fresh water habitats of Dhaka and the adjoining districts.

Although, a noteworthy progress have been made on the taxonomy of algal flora in Bangladesh since the early seventies, studies on the hydrobiological and limnological aspects of the inland water are not sufficient. Our knowledge is poor about the effects of various physico-chemical factors on the qualitative and quantitative aspects of aquatic flora and fauna which are of immense value from the biological and ecological points of view. Recently, however, a few workers have made some preliminary hydrobiological and limnological studies in some selected places of the country. Begum (1958) made some preliminary studies on the plankton in a fresh water pond in Dhaka. Islam and Khtun (1966) and Islam and Nahar (1967) worked on the plankton of polluted water. Islam and Begum (1970) made some contributions to the knowledge of phytoplankton of Dhaka district. Das and Bhuiyan (1974) worked on the limnoplankton of Dhaka. Islam et al. (1974) and Islam and Haroon (1975) studied the limnology of the river, Buriganga covering its physico-chemical and biological aspects. Islam and Shaha (1975) studied on the limnology of Ramna lake in Dhaka with some observation on the degree of pollution in this aquatic habitat Mahmood et al. (1976) made some hydrobiological observations on the Karnafully estuary. Islam and Mendes (1976) worked on the limnological aspects of a Jheel in Sher-E-Bangla Nagar in Dhaka. Shafi et al. (1978) made some valuable contributions to the limnology of river, Meghna. Mollah and Haque (1978) made some valuable additions to the knowledge of physico-chemical and biological aspects of some lotic and lemtic habitats in Mymensingh. Islam et al.(1978) studied the limnology of some fresh water ponds in Dhaka. Islam and Paul (1978) made

some hydrobiological studies of the Hakaluki Haor in the Sylhet district of Bangladesh. Islam et al. (1979) and Islam and Chowdhury (1979) worked on the macrophytes, benthic flora and the phytoplankton of the Dhanmondi lake in Dhaka.

Ali et al. (1980) studied the seasonal abundance of plankton in a pond in Dhaka. Islam et. al (1980) made some hydrobiological observation at Naogaon in Rajshahi and recorded a few new algal species and discussed the environmental conditions of the habitat. Rahman et. al (1982) and Mumtazuddin et. al (1982) worked on the limnology of fish ponds in Mymensingh and made some suggestions for the improvement of the overall productivity. Habib et. al (1983), Islam et. al (1983), Habib and Badruddin (1983) and Habib et. al (1984) worked on the physico-chemical characters of water and made some practical suggestions for high productivity in these aquatic bodies of Mymensingh. Ameen et. al (1986) also made some observations on the physico-chemical and biological conditions of some selected fish ponds in Dhaka. Ronald and Azadi (1987) studied the ecology of Halda river in Chittagong and made some estimations of the planktonic organisms and their co'relation with the physico-chemical conditions of water. Begum et. al (1989) and Nesha et. al (1989) worked on the physico-chemical conditions of water and seasonal variation of the plankton in some selected fish ponds in Dhaka with some notes on the productivity of these ponds.

Oppenheimer et. al (1978) and Islam et. al (1978) studied the lomnology of some fresh water ponds in Dhaka. Chowdhury et. al (1989) worked on the seasonal variation of zooplankton in a fish pond of Dhaka in relation to some physico-chemical factors. Mahmood (1986) studied the hydrobiology of Kaptic reservoir. Khondker et. al (1988) and Khondker and Rahim (1991) studied the

primary productivity and water quality of Dhanmondi lake of Dhaka. Khondker et. al (1990) made limnological studied of some ponds in and around Dhaka city. Khondker and Perveen (1992) worked on the limnology of a lake. Naz (1991) studied the eutrophic and hypertrophic nature of fish ponds of Rajshahi University and the adjacent region. Khondker et. al (1993) studied the periphytic and planktonic algae as indicators in Dhanmondi lake. Zaman et. al (1993) made a comparative limnological studies of three ponds in the Jahangirnagar University campus.

Many other scientists including Bhouyain et. al (1981), Mirza et. al (1985), Ali et. al (1980 and 1982) and Ali et. al (1983), Bhouyain and Asmat (1992), Zafar (1964), Zafar (1967), Bhouyain (1983), Khondker (1994 and 1995), Latif et. al (1986), Munawar (1970), Banu et. al (1987), Ameen et. al (1986), Ameen et. al (1987), Ameen et. al (1988), Khan et. al (1990), Zaman et. al (1993) have also made valuable addition to the knowledge of physicochemical and biological aspects of various lentic and lotic water systems including the fish culture ponds in the country during the recent years. Zaman (1991) studied the physico-chemical and biological aspects of Chalan beel and discussed the interrelationships between the physico-chemical parameters and plankton abundance including their periodicity and seasonal variations. Mushfaqua (1995), studied on the physico-chemical conditions and occurrence of the plankton biota in Ramsagar in Dinajpur. Fakruzzaman (1987) studied on the heleo planktion of lower barind region in relation to physico-chemical conditions.

Presently pond fishery is being practiced in the country in large scale for better augmentation of fish product. During the recent years a number of works on pond fishery have been done, but evaluation of modern technique of pisciculture as captive fishery is yet to be made.

Alongside, in India a large numbers of Limnological studies have been made during the last 3 decades. Notable amongst them are Venkateswarlu (1969

and 1986), Philipose (1957, 1959 and 1960), Sreenivasan (1963, 1964 and 1965), Rao (1953 and 1955), Rao (1977) and Rao (1979), Singh (1960), Singh and Ahmed (1990), Das and Srivastava (1956), Akhtar (1970), Kant and Anand (1978), Agrawal et. al (1976), Adebisi (1980), Biswas (1993), Chandler and Weeks (1945), Chakrabarty et. al (1959), Bhatnagar and Sharma (1973), Chakrabarty (1995), Ganapati (1957), Mirsha and Yadev (1978), Goutom (1990), Mirsha and Trivedy (1993), Palharya et. al (1993), Arora (1961), Trivedy (1990), Trivedy and Raj (1992), Paramasivam and Sreenivasan (1981) Ganapati (1940 and 1960), Mishra and Saksena (1992), Patil (1982), Sahai and Sinha (1969), Nasar and Sharma (1980), Pandey (1965), Lakshminarayana (1965), Roy (1955), Sukla (1971) and Rai (1978).

CHAPTER -2

MATERIALS & METHODS



Materials and Methods

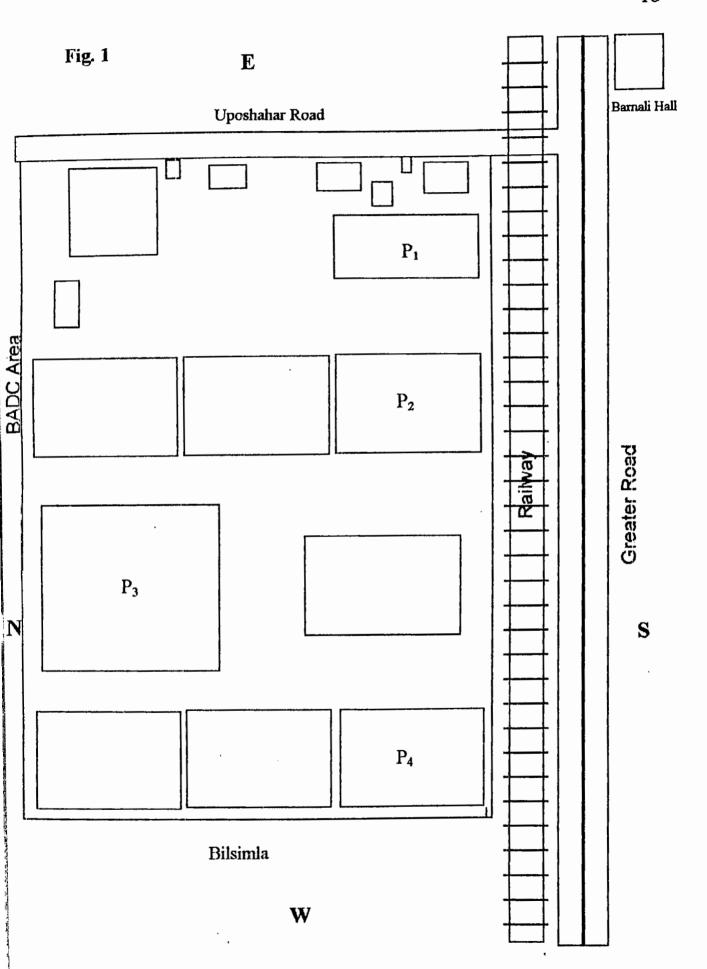
Water and Plankton samples were collected at fortnightly intervals from the four study ponds in Rajshahi Pisciculture hatchery thrice in each sampling date (8 AM; 12 AM and 4 PM).

The sampling dates were fixed between the 14th & 15th and the 29th & 30th days of each month. The data obtained from the regular analysis of the physico-chemical 'conditions were shown in tabular forms. The monthly abundance of phyto-and zooplankton were also presented in separate tables.

Description of the study ponds

The four study ponds are located in the Piscicultural Hatchery in Rajshahi. There are nine ponds in the hatchery under semi-intensive fish cultivation. Of these culture ponds, four were selected for the present investigation. These four ponds are located at a distance of 15 Meters from eachother (Fig.1). All the pond are highly charged. Each year in the month of November all ponds were de-watered and lime, Cow-dung, Poultry-litter, Urea, Triple-super-phosphate, Mustard oil-cake were used as needed. At the end of the year, 1993, 3 kg/decimal and at the end of the year, 1994, 9 kg/decimal lime were used in all the ponds.

Copper-sulphate, Sumithion and Dipterex were used three to five times in each month of the year in all ponds to control the plankton to avoid oxygen deflation. Triple-super-phosphate, Urea, Cow-dung and additional mixed food made by cattle-blood, rice-bran and mustard oil-cake were used in each day in all the ponds at the rate of 2-3% of total weight of stocked fishes.



Pond-1 (P₁): This ponds is situated near the hatchery. This is rectangular in shape. This pond covers an area of (15X45) 675M². No slope is presented in this pond. This pond is highly charged. Hatchery washing materials like Coppersulphate (CuSO₄.5H₂O); sulphuric acid (H₂SO₄); Hydrochloric acid (Hcl); Bleaching powder [Ca[Ocl]cl]; washing soda (Na₂CO₃.1OH₂O); Potassium Permanganate (KMno₄); Methyline blue (C₁₆H₁₈ClN₃S.2H₂O); Formaline (HCHO); are regularly mixed into the water of this pond as Hatchery effluent. The average depth of this pond is 0.75M.

Pond II (P₂): This Pond is situated near the pond (P₁). This Pond is rectangular in shape and covers an area (37X45) 1665M². Slopes of this pond are broken and irregular. This pond is a charged pond. Average depth of this pond is 1.5M.

Pond III (P₃): This pond is located at the middle of all other hatchery ponds. This pond is square in shape and covers an area (64X64) 4096M². The slopes of this are broken and irregular due to feeding activity of Common Carps. This pond is also a charged pond. The average depth of this pond is 1.5M.

Pond IV (P₄): This pond is located near study pond- 3 (P₃). This pond is rectangular in shape and covers an area (87X45) 1665 M². Slopes of this pond are broken due to feeding activity of Common Carps. This is a charged pond. The average depth of this study pond is 1.2 M.

Physical Conditions:

Air and Water Temperature:

A centigrade thermometer with the range of 0°c to 110°c was used to note the air and water temperature simultaneously during the time of sample collection. The temperature of water below the surface was recorded by dipping the thermometer at a depth of 10 to 12 cm. below the surface. To record the bottom temperature, the thermometer was tied with a silk line and descended till it reached the bottom and was kept for some times. Then it was drown upward quickly but Carefully and withdrawn from water Then, the Temperature was recorded immediately.

Transparency of water:

Measurement of limit of visibility i.e. Penetration of light in water was done by a secchi-disk. The secchi-disk was tied with a line and a meter-scale tape. Then it was slowly lowered into the water and depth at which it became invisible was noted. It was then, allowed to drop a little farther and the disk was pulled upward slowly the depth at which it reappeared was noted. The average of these two readings was taken to be the limit of visibility.

The sinking of the disk was always viewed under a sunshed for considerable accuracy in result. The data, thus, obtained, were expressed as secchi-disk depth in cm. Monthly variations of the secchi-disk depth were presented in tabular forms.

Average depth:

Average depth of water in the study ponds were taken regularly at fortnightly intervals during the study period from May 1994 - April 1995. It was accomplished by lowering a graduated nylon tape holding a metal weight at the end for sinking. Sometimes a graduated bamboo pole was also used for this purpose. The data for all the twelve months (two fish culturable years) from May, 1994 to April, 1995, were presented in tabular forms.

Chemical Conditions:

Dissolved oxygen Content and Percentage of Saturation of oxygen in water:

winkler,s method (unmodified) was followed for the estimation of dissolved oxygen content of water below the surface and also at the bottom. A glass stoppered bottle of 250 ml capacity was dipped into water. 10-15 cm. below the surface. The bottle was held under water and the glass-stopper was taken off and replaced again when the bottle was completely filled with water. The water-filled bottle was then take up for analysis in the field. Sample water from the bottom was collected by diving with sample bottle. The water filled bottles were taken up for analysis in the field. Dissolved oxygen was estimated immediately after collection, or the samples were treated with manganous sulphate soln., alkaline iodine soln. (KOH-KI soln.) and acidified with conc. sulphuric acid for later estimation. The treated samples were transferred to the laboratory and the remaining steps of analysis were done. The quantity of dissolved oxygen, thus estimated, was expressed in milligrams per liter of water (mg/l). The percentage of saturation of oxygen in water below the surface and at bottom, was calculated by dividing the titration value in ml. by the solubility value as determined by temperature, following the methods of Montogomary et al (1964), and Murry and Riley (1969). Rowson's nomogram (1944) was also used for a quicker reference to the oxygen saturation values. Percentage of saturation values vary with altitude, atmospheric pressure and temperature. As the study area is situated about 25 meter above the sea level, the saturation correction value was corrected by dividing it by the appropriate correction factors as followed by Mortimer (1956). The data, thus obtained were presented in tables. The monthly variation of dissolved oxygen content were presented in separate tables:

Carbonate alkalinity (CaCO3):

Carbonate alkalinity or phenolphthalein alkalinity was determined by titration of 100 ml. of water sample (from below the surface and at bottom), with N/150 sulphuric acid using phenolphthalein as an indicator (welch, 1948). The resultant data was, expressed in mg/l of CaCo₃ per litre. Monthly variations were computed in tables.

Bicarbonate alkalinity (CaHCO₃):

Bicarbonate alkalinity or methyl orange alkalinity was determined by titration of 100 ml. of water sample (from below the surface and at bottom) with N/50 sulphuric acid using methyl orange as an indicator (welch, 1948). The results were expressed in mg/l of CaHCO₃. Monthly variations were presented in tables:

Hydrogen Ion Concentration (pH):

Water samples were collected by dipping the broad mouthed glass stoppered bottles below the surface and at the bottom to estimate the values of p^H. The p^H values were determined by using a digital p^H meter (Model. Hanna-Instrument). All the data with monthly variations were presented in the tabular forms.

Electric Conductivity:

The value of Electric conductivity of water determined by using a portable conductivity meter (Moole, CM-1k) of range 0-10,000 Mm. moh/cm. All the resultant data were expressed in micro ohms per centimeter (µmoh/cm). Monthly variations were presented in tables. The value of Electric Conductivity was estimated by the formula given FAO (1984).

Oxidation - Reduction or Redox potential (Eh):

The value of oxidation-Reduction (Redox potential) [Eh] was determined indirectly from the following equation based on the p^H value of water samples.

$$Eh = Eo - 0.058 p^{H} + 0.0145 X log PO_{2}$$

Where, Eo= standard electrode

Potential was determined directly as per O₂ values following Gautom (1994). The monthly variations of Redox Potentials were presented in the tabular forms.

Oxidation-Reduction Index (rH2):

The value of Oxidation-Reduction Index (rH_2) was calculated by following the equation similar to that of Eh. $rH_2 = Eh/0.029 + 2P^H$ (Gautom-1994).

The monthly variations of rH₂ were presented in the tabular forms.

Total Dissolved Solids (TDS):

The values of total dissolved solids were determined indirectly from the following formula based on the value of Electric Conductivity of water samples.

Total Dissolved solids = Electric Conductivity X Conv. Factor.

Where, the value of Conv. factor = 0.65. TDS = E. Conductivity X 0.65.

The values of TDS were expressed in milligram per liter (mg/l). Monthly variations of TDS were presented in the tabular forms.

Free CO2:

Free Carbon-dioxide (CO₂) was estimated by titration of water sample (from below the surface and bottom) with N/44 Sodium hydroxide

(NaoH)solution using phenolphthalein as an indicator (welch, 1948) The results were expressed in mg. of free CO₂ per litre (mg./l) of water. Monthly variations were presented in tabular forms.

Biochemical Oxygen Demand (BOD):

The samples of BOD bottles were filled with water and immediately transported to the laboratory and left for incubation in 20°c for five days. Dissolved Oxygen (DO) content of BOD bottle sample was determined after five days by following the Winkler's Method (Unmodified). The value of BOD was obtained by subtracting final dissolved Oxygen (FDO) from initial dissolved oxygen (IDO) values. When the DO content was almost negligible or showing the condition nearing anoxia, the BOD was determined by diluted method (APHA, 1989).

Chemical Oxygen Demand (COD):

It was determined by ferrous ammonium sulphate method (APHA,1989). A 50 ml. of sample was placed in a 500 ml. refluxing flask and 1 gm of HgSO₄ Crystals and a 5 ml. H₂SO₄ (conc). were added with several boiling chips. After some times a 25 ml. of 0.25 N K₂Cr₂O7 soln. was again mixed. The refluxing flask was attached to a condenser through which (condenser) a 70 ml. Sulphuric acid was added again. After mixing well, all the regents of the flask, heat was applied for half an hour. The sample was cooled in room temperature and was titrated (the sample) against ferrous ammonium sulphate with ferroin indicator. A blank was also run (double distilled H₂O) in the same procedure. The result was expressed in mg/l.

Total Hardness:

A 50 ml. sample pretreated with 1 ml. ammonia buffer solution was titrated against EDTA using Eriocrome Black-T as an indicator (Mishra et al. 1992). Samples were collected from below the surface and at bottom. All the resultant data were expressed in mg/l. Monthly variations of total hardness were presented in tabular forms.

Total hardness was calculated by using the formula.

Total hardness =
$$\frac{\text{ml of EDTA used X1000}}{\text{ml of sample}}$$

Phosphate (PO4):

A 50 ml. of sample was taken in a flask followed by the addition of acid ammonium molybodate soln. (2 ml.) and 4-5 drops of Sncl₂ soln. A blue colour appeared. A blank was run following the same procedure for comparison. Both sample and blank were then kept on white paper, followed by the addition of standard phosphate soln. to the blank drop by drop with the help of graduated pipette (1 ml) until the colour of blank matches the sample, at the end point, the ml. of standard phosphate used was noted. Amount of phosphate was calculated by the following formulae;

Phosphate= ml. of solution used x 0.01 x 20.

The result was expressed in mg/l. Monthly variations of phosphate were presented in the table.

Mounting Fluid:

Ten percent glycerin solution in distilled water was used as mounting fluid for the preparation of temporary and semi-permanent slides for

microscopic studies. Cotton blue and lactophenol were used as mounting fluid and as stain, which facilitated a clear view of the materials under the microscope.

Plankton analysis:

The planktonic forms were collected by a plankton-net of No. 20 silk-bolting cloth. A 100 litres of water from the study pond was passed through the plankton-net using a 5 litre capacity bucket. Water was poured round the innerside of the net to wash down the adhering plankter from the meshes. After collection the plankton, materials were transferred into the glass bottle and preserved permanently in Transeau's Solution. The final volume of the planktonic materials was adjusted to 30 ml. These were regarded as the original concentrate.

The composition of the preservative: Distilled water 60 ml, or 6 parts. Absolute alcohol 30 ml, or 3 parts. Foromaldehyde 10 ml, or 1 parts.

To each 100 ml, of above solution a 5 ml. of glycerin was added to prevent the materials from becoming brittle (Transeau, 1951).

The quantitative enumeration of the phyto and zooplankton were carried out with the help of a Sedwick Rafter counting cell and Occular Whiple micrometer and by drop method (welch, 1948). The counting cell was filled with water to test for any leakage and it was emptied and dried properly. The vials containing the concentrate were shaken properly and 1 ml. of concentrate was taken into the Sedwick Rafter counting cell. Numerical counts of all members of phyto-and-zooplankton were made using the whiple micrometer.

To achieve a random sampling, each time 10 fields were examined for each sample and an average of the counts were recorded. The organisms thus counted, were expressed as units per litre (units/l) of the sample, irrespective of whether they were solitary cells, colonies or filaments or part there of. The monthly abundance of phyto-and-zooplankton were presented in tables. The abundance of plankton groups were calculated according to the following formulae (welch-, 1948).

$$N = \frac{(a \times 1000)c}{I}$$

where, N = Number of plankton per litre of original water.

a= average No. of plankton in all counts in the counting unit.

c= volume of original concentrate in ml.

L= volume of water passed through the net.

Primary Productivity:

Primary productivity was estimated by the method suggested by Gaader and Grane (1927) with an incubation period of four hours. At first, the value of initial dissolved oxygen content was determined by titration and it was recorded. Then, two glass stoppered bottles of 250 ml. capacity were filled with water from below the surface. One bottle was covered with black aluminium foil and other white bottle was uncovered. These two bottles were kept together under water parallelly for four hours for incubation. After the mentioned time, The values of dissolved oxygen contents of both the bottles were determined. the value of primary productivity was estimated by the following formulae:

Gross primary productivity =
$$\frac{LB-DB \times F}{Time}$$
Net primary productivity =
$$\frac{LB-IB \times F}{Time}$$

Where, LB = DO in the light bottle, DB= DO in the dark bottle,

IB = Initial DO, GPP = gross primary productivity,

NPP = Net primary productivity,

F= conversion factor. Ratio of molecular weights of carbon and oxygen (0.375). All oxygen values were converted into carbon values.

The values of GPP and NPP were expressed in milligrame carbon per litre per hour (mgC/l/h). The monthly variations of average of GPP and NPP were presented in tables.

Ammonium-nitrogen (NH₃-N):

Ammonia-nitrogen was determined by the Fish Farmer's Water Quality Test-kit, Model, FF-1A, range- 0.1 to 3.0 mg/l. To determine ammonia-nitrogen, 5 ml. of water sample was taken to a vial treated with Nessler reagent (APHA-1989) and was kept for a 15 minutes. After changing of colour of the sample, was matched with the disk of kit to determine the volume NH₃-N. Ammonia-nitrogen was expressed in mg/l. Monthly variations of ammonia-nitrogen were presented in tables.

CHAPTER -3

OBSERVATIONS AND RESULTS



OBSERVATIONS AND RESULTS

A. Physical Factors: Air-Temperature

Air temperature varied considerably over the year and ranged from $17.5 \pm 3.3^{\circ}$ C to $0.38.3 \pm 1.7^{\circ}$ C. Records of air temperature indicated a gradual fall from October (1994) and it reached the minimum in January (1995). It began to rise from the mid February to April. From May to July the air temperature was found to be more or less uniform with slight variation in June and August. The highest temperature was recorded in April (1995).

Average Depth of Water. The average depth of water in the study ponds was found to vary from 0.9 ± 0 m to 2.3 ± 0.1 m. The average depth of water in study pond-1, varied from 0.9 ± 0 m to 1.2 ± 0 m throughout the study period. The maximum depth of water was observed in the months of August and November $(1.2 \pm 0 \text{ m})$ which fell to 0.9 ± 0 m in September and October which rose again from December to March. The average depth maintained an uniformity in the months of May to July (1994). In the month of April (1995) the average depth started to rise again (Tab-1). In study pond -2, the average depth of water varied from 1.4 ± 0 m to 1.8 ± 0 m throughout the study period. The highest depth was recorded in August and April while the lowest value was observed in October to December (1994) and February to March (1994). The study pond is comparatively deep and the average depth was found to show an uniform value during the study months (Tab -2). Study pond -3, is also a deep pond and the highest depth was observed in August (2.3 \pm 0.1 m) while the lowest value was obtained in the months of November to December (1.7 \pm 0.1

m). The average depth of this study pond was found to be fairly uniform in other months of the year (Tab -3). The average depth of water in study pond-4, was found to vary from 1.4 ± 0 m to 2.0 ± 0 m throughout the study period. The highest depth was observed in August (1994) while the lowest depth was recorded in September to October (1994) and January to February (1995). This pond maintained an uniform depth in rest of the study months (Tab -4). All the study ponds are rainfed but underground water is pumped into them whenever needed.

Secchi Disk Depth (SDD). The Secchi disk depth (SDD) of water in the study ponds was found to vary from 0.1 ± 0 m to 0.4 ± 0.2 m. The SDD of water in study pond -1, varied from 0.1 ± 0 m to 0.3 ± 0 m throughout the study period. The maximum SDD of water was observed in the month of January (1995) while it fell to 0.1 ± 0 m in September (1994). The SDD maintained an uniformity in the other months of the year (Tab -1). In study pond -2, the SDD of water varied from 0.1 ± 0 m to 0.4 ± 0.1 m throughout the study period. The highest SDD was recorded in December (1994) while the lowest value was observed in February (1995). In the month of March (1995) the SDD started to rise again. The SDD was found to show an uniform value during the study months (Tab -2). In study pond -3, the SDD was found to vary from 0.1 ± 0 m to 0.3 ± 0.1 m throughout the study period. The highest SDD was observed in the months of July and February $(0.3 \pm 0.1 \text{ m})$ while the lowest value was obtained in January (0.1 \pm 0 m). The SDD of this study pond was found to be fairly uniform in other months of the study period (Tab -3). The SDD of water in study pond -4, was found to vary from 0.1 ± 0 m to 0.3 ± 0.1 m throughout the study period. The highest SDD was observed in September (0.3 \pm 0.1 m) while

the lowest value was recorded in December $(0.1 \pm 0 \text{ m})$. The pond maintained an uniform SDD depth in rest of the study months (Tab -4).

Total Dissolved Solids (TDS). The Total Dissolved Solids (TDS) of water in the study ponds was found to vary from 344.0 mg/l to 499.0 mg/l. The TDS of water in study pond -1, varied from 353.0 mg/l to 489.5 mg/l throughout the study period. The maximum TDS value of water was observed in May (489.5 mg/l) while fell to 353.0 mg/l in September which rose again in June and July and November and February, 1995. The TDS of water maintained an uniformity in the months of August, October, December (1994) and January, March and April 1995(Tab-1). In study pond -2, the TDS value of water varied from 344.0 mg/l to 499.0 mg/l throughout the study period. The highest TDS value was recorded in October (499.0 mg/l) while the lowest value was observed in September (344.0 mg/l). The TDS was found to show an uniform value during the rest of the study months (Tab -2). In study pond -3, the highest TDS of water was observed in January (450.1 mg/l) while the lowest value was obtained in August (381.6 mg/l). The TDS value of this pond was found to be fairly uniform in other months of the year (Tab -3). The TDS of water in study pond -4, was found to vary from 351.3 mg/l to 470.6 mg/l throughout the study period. The highest TDS value was observed in December (470.6 mg/l) while the lowest value was recorded in April (351.3 mg/l). The pond maintained an uniform TDS value in rest of the study months. (Tab -4).

Electric Conductivity. The Electric conductivity of water in the study ponds was found to vary from 529.0 \pm 32.0 μ moh/cm to 577.5 \pm 38.5 μ moh/cm. The Electric conductivity of water in study pond -1, varied from 543 \pm 46 μ moh/cm to 753 \pm 9 μ moh/cm throughout the study period. The maximum

Electric conductivity of water was observed in May (753 \pm 9 μ moh/cm) while fell to $543 \pm 46 \mu$ moh/cm in September which rose again in June to August and October to November (1994) and February (1995). The Electric conductivity maintained an uniformity in the months of December (1994) to January and March to April, 1995 (Tab-1). In study pond -2, the Electric conductivity of water varied from 529.0 \pm 32.0 μ moh/cm to 767.0 \pm 37.0 μ moh/cm throughout the study period. The highest Electric conductivity was recorded in October (767.0 ± 37.0 μ moh/cm) while the lowest value was observed in September $(529.0 \pm 32.0 \,\mu \,\text{moh/cm})$. The Electric conductivity was found to show an uniform value during the rest of the study months (Tab -2). In study pond -3, the highest Electric conductivity was observed in January (693.0 ± 47.0 μ moh/cm) while the lowest value was obtained in August (587.0 \pm 2.0 μ moh/cm). The Electric conductivity of this study pond was found to be fairly uniform in other months of the year (Tab -3). The Electric conductivity in study pond -4, was found to vary from 541.0 ± 45.0 to 724.0 ± 61.0 μ moh/cm. throughout the study period. The highest Electric conductivity was observed in December (724.0 ± 61.0 μ moh/cm.) while the lowest value was recorded in April (541.0 \pm 45.0 μ moh/cm.). The pond maintained an uniform Electric conductivity in rest of the study months (tab -4).

Chemical Conditions.

Hydrogen Ion Concentration (p^H). The p^H of water in the study ponds was found to vary from 7.7 ± 0.1 to 8.9 ± 0.1 . The p^H of water in the study pond -1, varied from 7.7 ± 0.1 to 8.7 ± 0.1 throughout the study period. The maximum p^H value of water was observed in the months of October and March to April 1995 (8.7 ± 0.1) which fell to 7.7 ± 0.1 in May (1994) It rose again in

November (1994) and maintained uniformity upto February 1995 (Tab -1). In study pond -2, the p^H of water varied from 7.8 ± 0.1 to 8.9 ± 0.1 throughout the study period. The highest pH value was recorded in February to March 1995 (8.9 ± 0.1) while the lowest value of p^H was observed in May 1994 (7.8 \pm 0.1). The pH of water rose again in October and January (1995). The pH was found show an uniform value during study months (Tab -2). In study pond -3, the pH of water was found to vary form 8.2 ± 0.1 to 8.9 ± 0.1 throughout the study period. The highest value of p^H was observed in April (8.9 \pm 0.1) while the lowest value was obtained in May (8.2 \pm 0.1). The p^H started to rise again from September 1994 to March, 1995. The pH of this study pond was found to be fairly uniform in other months of the year (Tab -3). The pH of water in study pond -4, was also found to vary from 8.2 ± 0.1 to 8.9 ± 0.1 throughout the study period. The highest value was observed in February and April 1995 (8.9 ± 0.1) while the lowest value was recorded in may (8.4 ± 0.1) . The p^H started to rise again and maintained high value in September to October and December (1994) to January (1995). The pond maintained an uniform p^H value in rest of the study months (Tab -4).

Free Carbondioxide (CO₂). The free CO₂ of water in all the study ponds was found to vary from zero (0.0 mg/l) to 34.7 ± 6.9 mg/l. The free CO₂ of water in study pond -1, varied from 2.8 ± 6.2 mg/l to 34.7 ± 6.4 mg/l throughout the study period. The maximum value was observed in the months of May, September, October, December and February (34.7 ± 6.4 mg/l) to April which fell to 2.8 ± 6.2 mg/l in August. The free CO₂ content maintained an uniform value in the months of June, July, November (1994) and January 1995 (Tab -1).

In study pond -2, the free CO_2 of water varied from 4.3 ± 9.5 mg/l to 32.5 ± 6.7 mg/l throughout the study period. The highest value of free CO_2 was recorded in May, September, December $(32.5 \pm 6.7$ mg/l) and February and March while the lowest value was observed in April $(4.3 \pm 9.5$ mg/l). The free CO_2 was found to show an uniform value during rest of the study months (Tab -2).

In study pond -3, the highest free CO_2 was observed in March (14.6 \pm 15.2 mg/l) while zero value was obtained in the months of June, December and January (0 mg/l). In the months of September and February, the free CO_2 started to rise again. The free CO_2 of this study pond was found to be fairly uniform in other months of the year (Tab -3). The free CO_2 of water in study pond -4, was found to vary from zero (0.0 mg/l) to 5.4 ± 7.7 mg/l throughout the study period. The highest value of free CO_2 was recorded in January (5.4 \pm 7.7 mg/l) while the absence of free CO_2 was observed in the months of June to September, December and February to April. The pond maintained an uniform free CO_2 in rest of the study months (Tab -4).

Carbonate alkalinity as (CaCO₃). The carbonate alkalinity of water in the study ponds was found to vary from zero (0.0 mg/l) to $17.2 \pm 8.4 \text{ mg/l}$. The CO₃ alkalinity of water in the study pond -1, varied from zero (0.0 mg/l) to $17.2 \pm 8.4 \text{ mg/l}$ throughout the study period. The maximum CO₃ alkalinity of water was observed in the month of August $(17.2 \pm 8.4 \text{ mg/l})$ which fell to zero (value) in the months of May to July, September to December (1994) and February to April (1995). In the month of January, the CO₃ alkalinity started to mark a rise again (Tab -1). In study pond -2, the CO₃ alkalinity of water varied from zero (0.0 mg/l) to $15.2 \pm 8.7 \text{ mg/l}$ throughout the study period. The highest value was recorded in April $(15.2 \pm 8.7 \text{ mg/l})$ and the lowest in the months of May, June,

July and December 1994 (0.0 mg/l). The CO₃ alkalinity value started to rise again and from the month of January (1995), the pond was found to show an uniform value of carbonate alkalinity in the rest of the study months (Tab -2). The carbonate alkalinity of water in study pond -3, was found to vary from 1.2 ± 1.6 mg/l to 8.0 ± 4.7 mg/l throughout the study period. The highest value of carbonate alkalinity was observed in January (8.0 ± 4.7 mg/l) while the lowest value was obtained in August (1.2 ± 1.6 mg/l). From the months of May, September and November to December (1994), the carbonate alkalinity started to rise again. The rest of the months, the CO3 alkalinity value was found to be uniform (Tab -3). The carbonate alkalinity of water in study pond -4, was found to range from 1.8 \pm 1.9 mg/l to 7.7 \pm 1.8 mg/l throughout the study period. The highest value of carbonate alkalinity was observed in February, 1995 (7.7 \pm 1.8 mg/l) while the lowest value was recorded in October (1.8 \pm 1.9 mg/l). The carbonate alkalinity of this study pond was found to rise again from May and highest values were observed in August, September, and from December (1994) to April 1995 (Tab -4).

Biocarbonate alkalinity as (CaHCO₃). The biocarbonate alkalinity (HCO₃) of water in all the study ponds was found to range from 15.1 ± 4.5 mg/l to 445.2 ± 60.5 mg/l. The HCO₃ alkalinity of water in study pond -1, varied from 21.0 ± 1.7 mg/l to 403.0 ± 56.0 mg/l throughout the study period. The maximum HCO₃ alkalinity of water was observed in the month of April (403.0 ± 56.0 mg/l) which fell to $(21.0 \pm 1.7$ mg/l) in July. It rose again from August (1994) and maintained a high value upto April (1995). The biocarbonate alkalinity maintained an uniformity in the months of May to July, 1994 (Tab -1). In study pond -2, the HCO₃ alkalinity of water ranged from 18.7 ± 2.5 mg/l to 356.7 ± 122.6 mg/l throughout the study period. The highest bicarbonate

alkalinity was recorded in August (356.7 ± 122.6 mg/l) while the lowest value was observed in July (18.7 ± 2.5 mg/l). The biocarbonate alkalinity maintained almost an uniform value in May and June (1994). The HCO₃ alkalinity value started to rise again from August (1994) and maintained a high value till April -1995 (Tab -2). In study pond -3, the HCO₃ biocarbonate alkalinity values varied from 15.1 \pm 4.5 mg/l to 445.2 \pm 60.5 mg/l. The highest value of carbonate alkalinity was observed in April (445.2 ± 60.5 mg/l) while the lowest value was obtained in June (15.1 \pm 4.5 mg/l). The HCO₃ alkalinity value of this pond started to rise again from September (1994) to April (1995). The HCO₃ alkalinity of this pond was found to be fairly uniform in other months of the year (Tab -3). The HCO₃ alkalinity of water in study pond -4, had a range from 15.2 ± 3.2 mg/l to 336.6 ± 35.2 mg/l throughout the study period. The highest value was observed in April (336.6 \pm 35.2 mg/l) while the lowest in July (15.2 \pm 3.2 mg/l). The HCO₃ alkalinity contents of this pond started to rise again from August (1994) and maintained higher values till April (1995). During the rest of the study months the HCO₃ alkalinity values were found to be almost uniform.

Total Hardness (TH). The total hardness of water in all study ponds had ranged from 197.8 ± 28.0 mg/l to 608.8 ± 44.6 mg/l. The total hardness of water in study pond -1, varied from 200.0 ± 18.0 mg/l to 538.0 ± 47.0 mg/l throughout the study period. The maximum total hardness of water was observed in April (538.0 ± 47.0 mg/l) which fell to (200.0 ± 18.0 mg/l) in the months of May, August (200.0 ± 18.0 mg/l) and September (1994). It started to rise again from October (1994) and maintained upto April (1995). Uniform TH values were observed in other months (Tab -1). In study pond -2, the total hardness of water varied from 214.2 ± 43.1 mg/l to 609.2 ± 56.3 mg/l throughout the study period. The highest value was recorded in April (609.2 ± 56.3 mg/l) while the lowest value was observed in September (214.2 ± 43.1 mg/l). The total hardness of

water was found to rise again from October (1994) and maintained higher values till April (1995). TH values were found to be uniform in rest of the study months (Tab-2). In study pond-3, the total hardness of water was found to vary from 228.7 ± 60.0 mg/l to 608.8 ± 44.6 mg/l during the study period. The highest value of total hardness was observe in April (608.8 ± 44.6 mg/l) while the lowest value was obtained in the months of July (228.7 ± 60.0 mg/l) and September (1994). The TH values started to rise again from the months of November 1994 (Tab-3). The total hardness of water in study pond-4, was found to vary from 197.8 ± 28.0 mg/l to 469.3 ± 110.0 mg/l throughout the study period. The highest value of total hardness was observed in April (469.3 ± 110.0 mg/l) while the lowest value was recorded in July (197.8 ± 28.0 mg/l). The total hardness of this study pond was found to rise from February and maintained a higher value till April (1995). The pond maintained an uniform total hardness in the rest of the study months (Tab-4).

Chloride (cl). The Chloride of water in all the study ponds was found to vary from 87.0 to 201.3 mg/l. The Chloride content of water in study pond-1, varied from 87.0 to 143.0 mg/l. The highest value of chloride was observed in the months of June, August (143.0 mg/l) 1994 and March to April (1995) which fell to 87.0 mg/l in December (1994). The Chloride values maintained an uniformity in the months of May, July, September, October (1994) and January to February (1995). In the month of November, the Chloride started to rise again (Tab-1). In study pond-2, the Chloride values of water varied from 103.6 mg/l to 191.7 mg/l throughout the study period. The highest Chloride was recorded in July (191.7 mg/l) while the lowest value was observed in March (103.6 mg/l). The Chloride COntent started to rise again in May, August, September and

November (1994). The Chloride values were found to show an uniform value in the rest of the study months (Tab-2). In study pond-3, the Chloride of water was found to vary from 103.4 mg/l to 201.3 mg/l during the study period. The highest value of Chloride was obtained in Juanuary (201.3 mg/l) and the lowest in July (103.4 mg/l). The Chloride content maintained an uniformity in the months of May to June and August to November (1994). In the month of December (1994) and from February to April (1995), the Chloride marked a rise again (Tab-3). The Chloride content of water in study pond-4, was found to range from 98.6 mg/l to 142.0 mg/l during the period of study. The highest value of Chloride was observed in October (142.0 mg/l) while the lowest value was recorded in May and January (98.6 mg/l). The pond maintained an uniform Chloride content in the rest of the study months (Tab-4).

Dissolve Oxygen (DO). The DO of water in the study ponds was found to vary from 4.0 ± 1.1 mg/L to 10.6 ± 3.5 mg/lthroughout the study period.

Pond -1: The DO content of water at this spot varied from 4.0 ± 1.1 mg/l to 7.6 ± 1.3 mg/l throughout the study period. The maximum DO value was observed in January (7.6 ± 1.3 mg/l) which fell to (4.0 ± 1.1 mg/l) in February and rose again in the months of May and July to October (1994). The DO maintained an uniform value in the months of June and November to December (1994) and from March to April -1995 (Tab-1).

Pond -2: At this study spot the DO content of water was found to vary from 5.0 ± 1.5 mg/l to 8.1 ± 1.1 mg/l throughout the study period. The highest value of DO was recorded in the months from August to October and January to February and April (8.4 ± 1.1 mg/l) while the lowest value was observed in

December (5.0 \pm mg/l). The DO was found to show an uniform moderate value during the rest of the study months (Tab-2).

Pond -3: In study pond the DO content of water was found to vary from 5.7 ± 2.8 mg/l to 9.6 ± 1.4 mg/l throughout the study period. The highest value of DO was observed in September $(9.6 \pm 1.4$ mg/l) and December while the lowest value was obtained in the months of November and February $(5.7 \pm 1.8$ mg/l). In the months of January and April (1995) the value of DO started to rise again. The DO of this study pond was found to be moderately uniform in other months of the year (Tab-3).

Pond -4: The DO content of water in this study pond ranged from 6.3 ± 0.5 mg/l to 10.6 ± 3.5 mg/l throughout the study period. The highest value of DO was observed in September $(10.6 \pm 3.5 \text{ mg/l})$ while the lowest value was recorded in June $(6.3 \pm 0.5 \text{ mg/l})$. In the months of May, October and December (1994) to April (1995) the DO values registered arise gain. The pond maintained an uniform value of DO in rest of the study months (Tab-4).

Percentage Saturation of Oxygen. The percentage saturation of Dissolved Oxygen of water in the study ponds was found to very from 45% to 140% during the study period.

Pond -1: The percentage saturation of DO of water in this study pond, varied from 45% to 92% throughout the study period. The maximum percentage saturation of DO was observed in the month of August to September(92%) which fell to 45% in February and rose again in the months of May, July, October (1994) and January (1995). The percentage saturation of DO maintained an uniformity in the month of June, November to December (1994).

Biochemical Oxygen Demand (BOD₅). The Biochemical Oxygen Demand (BOD₅) of water in the study ponds had ranged from 2.3 mg/l to 16.8 mg/l during the period of investigation.

Pond -1: The BOD₅ in study pond-1, varied from 2.7 mg/l to 12.0 mg/l throughout the study period. The maximum BOD₅ value was observed in the months of August, 94 12.0 mg/l which fell to 2.7 mg/l in September to October (1994) and from February to April (1995) which rose again in the months of May to June (7.6 mg/l) and January (1995). The BOD₅ maintained an uniformity in the months of July and November to December 1994 (Tab-1).

Pond -2: In this study pond, the BOD₅ of water varied from 2.3 mg/l to 9.6 mg/l throughout the study period. The highest value of BOD₅ was recorded in July, August (9.6 mg/l) and October (1994) while the lowest value was observed in June 1994 (2.3 mg/l) and February to March (1995). The BOD₅ was found to show an uniform value during the study months (Tab-2).

Pond -3: At this study spot the BOD₅ value was found to range from 3.2 mg/l to 13.6 mg/l throughout the study period. The highest value was observed in June (13.6 mg/l), July, October, March and April, while the lowest value was obtained in the months of January and February (3.2 mg/l). In the months of May and August 1994, the BOD₅ value was found to rise again. The BOD₅ of this study pond was found to be almost uniform in other months of the year (Tab-3).

Pond -4: The BOD₅ of water in study pond-4, was found to very from 8.6 mg/l to 16.8 mg/l throughout the study period. The highest value of BOD₅ was observed in June and November (16.8 mg/l) while the lowest value was recorded

in February and March (8.6 mg/l). In the months of July and August (1994), January and April (1995), the BOD₅ value was found to rise again. The pond maintained a moderately uniform value in rest of the study months (Tab-4).

Chemical Oxygen Demand (COD). The Chemical Oxygen Demand of water in the study ponds was found to range from 6.7 mg/l to 37.2 mg/l.

Pond -1: The COD of water in this study pond, was found to very from 7.2 mg/l to 22.5 mg/l throughout the study period. The maximum value of COD of water was observed in the months of May, June, August, 1994 (22.4 mg/l) and January (1995) which fell to 7.2 mg/l in October, 1994 and April, 1995. The value registered a rise again in July and November and December (1994). In the rest of the months the COD value were found to be fairly uniform (Tab-1).

Pond -2: In this study pond the COD of water varied from 6,7 mg/l to 20.7 mg/l throughout the study period. The highest value of COD was reCOrded in August (20.7 mg/l) while the lowest value was observed in June (6.7 mg/l). In the months of July and October 1994, the COD was found to rise again. The COD values were almost uniform during other months of the study period (Tab-2).

Pond -3: At this spot, the COD values varied from 8.7 mg/l to 28.7 mg/l during the study period. The highest value was observed in the months of June (28.7 mg/l), July and October in 1994 and March 1995 while the lowest value was obtained in January and February 1995 (8.7 mg/l). In the months of May, August, December 1994 and April 1995, the COD value was found to be fairly uniform (Tab-3).

Pond -4: The COD of water in this study pond, had a range from 17.3 mg/l to 37.2 mg/l during study period. The highest value of COD was observed in June (37.2 mg/l) while the lowest was recorded in March (17.3 mg/). The COD was found to rise again in the months of August, October and November 1994. Fairly uniform COD values were noted in rest of the study months (Tab-4).

Ammonium Nitrogen (NH₃-N). The ammonium nitrogen of water in the study ponds varied from zero value to 0.13 mg/l.

Pond -1: The ammonium nitrogen of water in this study pond, varied from a zero values to 0.13 mg/l throughout the study period. The maximum value of NH₃ -N was observed in January (0.13 mg/l) which fell to zero (0.0 mg/l) in April. It rose again in November 1994 (0.11 mg/l). The ammonium nitrogen maintained an uniformity in its content in other months of the study period (Tab-1).

Pond -2: At this spot, the NH₃-N content of water varied from a zero value to 0.12 mg/l during the study period. The highest value was recorded in December $(0.12 \pm 0 \text{ mg/l})$ while the lowest was observed in April (0.0 mg/l). It rose again in November 1994. The ammonium nitrogen was found to show moderately uniform values in rest of the study months (Tab-2).

Pond -3: In the pond-3, the NH₃-N of water was found to vary from a zero value to 0.13 ± 0 mg/l throughout the study months. The highest value was noted in December $(0.13 \pm 0$ mg/l) while the lowest value was obtained in April (0.0 mg/l). Ammonium nitrogen content started to rise again from November. The NH₃-N of this study pond was found to be fairly uniform in other months of the year (Tab-3).

Pond -4: The ammonium nitrogen of water in this study pond, was found to vary from 0.0 mg/l to 0.12 ± 0 . mg/l during the study period. The highest value of NH₃-N was observed in February (0.12 \pm 0. mg/l) while the lowest value was recorded in April (0.0 mg/l) which started to rise again in January 1995. The NH₃-N values were almost uniform in the rest of the study months (Tab-4).

Phosphate (PO₄). The phosphate content of water in the study ponds was found to range from 0.01 mg/l to 0.7 mg/l during the period of investigation.

Pond -1: The phosphate of water in this study pond, varied from 0.1 mg/l to 0.7 mg/l throughout the study period. The maximum PO₄ value was observed in January and March (0.7 mg/l) which fell to 0.1 mg/l in May, July, and September. The PO₄ value was almost uniform in August, October (1994), February and April 1995 (Tab-1).

Pond -2: In study pond-2, the phosphate content of water varied from 0.02 mg/l to 0.1 mg/l throughout the study months. The highest value was recorded in the months from May to September and from November to January and also from March to April (0.1 mg/l) While the lowest value was observed in October (0.02 mg/l). The phosphate was found to rise again in the month of February 1995 (tab-2).

Pond -3: In this study spot, the phosphate content of water was found to vary from 0.01 mg/l to 0.3 mg/l throughout the study period. The highest phosphate value was observed in June and December 1994 (0.3 mg/l) While the lowest values was obtained in August, February (0.01 mg/l) and March (1995). The phosphate contents of this study pond was found to be fairly uniform in

other months of the year. In the month of November (1994), The phosphate content started to rise again (Tab-3).

Pond -4: The phosphate content of water in this study pond, was found to range from 0.02 mg/l to 1.0 mg/l throughout the study period. The highest value of phosphate content was observed from May to August (1.0 mg/l) while the lowest value was recorded in March (0.02 mg/l). During October and November 1994, February and April 1995 the PO₄ value was almost uniform (0.2 mg/l). The pond maintained an uniform value of phosphate in rest of the study months (Tab-4).

Redox Potential (Eh). The Redox potential (Eh) of water in the study ponds varied from 0.22 mv to 0.36 mv during the study months.

Pond -1: The Redox Potential (Eh) of water in study pond-1, varied from 0.25 mv. to 0.36 mv. throughout the study period. The maximum Eh value was observed in May (0.36 mv) which fell to 0.25 mv in March and April. The Redox Potential values (Eh) maintained an uniformity in the months from June 1994 to March 1995 (Tab-1).

Pond -2: At this study spot, the Redox Potential (Eh) of water varied from 0.22 mv to 0.36 mv during the study time. The highest Eh value was recorded in May (0.36 mv) while the lowest Eh value was noted in the months from January to April 1995 (0.22 mv.). The Eh value was found to show an uniform level during the rest of the study months (Tab-2).

Pond -3: In this study pond, the Redox potential (Eh) of water varied from 0.23 mv. to 0.31 mv. throughtout the study period. The highest value was observed in May (0.31mv) while the lowest value was obtained in the months of February and April 1995 (0.23 mv), The Redox Potential (Eh) of this study pond was found to be fairly uniform in other months of the year .(Tab-3).

Pond -4: The Redox Potential (Eh) of water in this study pond, varied from 0.23 mv to 0.32 mv during the study period. The highest value was observed in July (0.32 mv) while the lowest value was recorded in the months of February and April (0.23 mv). The pond maintained an uniform Eh value in rest of the study months (Tab-4)

Redox Index (rH₂). The Redox Index (rH₂) of water in the study pond was found to very from 25.02 to 27.81 during the study period.

Pond -1: The Redox Index (rH₂) of water in study pond-1, varied from 25.82 to 27.81 throughout the study period. The maximum rH₂ value was observed in May (27.81) which fell to 25.82 in the months from September 1994 to April 1995. The rH₂ value was almost uniform in the period from June to August 1994 (Tab-1).

Pond -2: At this study site, the Redox Index value (rH₂) of water varied from 25.39 to 27.67 throughout the study period. The highest rH₂ value was recorded in May (27.67) while the lowest value was observed in the months from October 1994 to April 1995 (26.02). The rH₂ value was found to show an uniform level during the rest of the study months (Tab-2).

Pond -3: In this study pond, the rH₂ value was found to vary from 25.02 to 27.09 throughout the study period. The highest rH₂ value was observed in May and July to August (27.09) while the lowest value was obtained in the month of January (25.02). The rH₂ value of this study pond was found to be fairly uniform in other months of the year (Tab-3).

Pond -4: Redox Index value (rH₂) of water in study pond-4, was found to vary from 25.73 to 27.23 throughout the study period. The highest rH₂ value was noted in May, July (27.23) and August while the minimum value was recorded in the months from September 1994 to April 1995(Tab-4).

Gross Primary Productivity (GPP). The Gross Primary Productivity (GPP) of water from in study ponds was found to vary from 0.1 mg C/l/h to 0.4 mg C/l/h during the period of investigation.

Pond -1: The Gross Primary Productivity (GPP) of water in this study pond, ranged from 0.2 mg C/l/h to 0.4 mg C/l/h throughout the study period. The maximum value of GPP was observed in the months of January to March (0.4 mg C/l/h) which fell to 0.2 mg C/l/h in May and continued in June, September, November and December 1994. The GPP maintained uniformity in the months of July to August, October 1994, February and April 1995 (Tab-1).

Pond -2: In this study pond, the GPP of water varied from 0.1 mg C/l/h to 0.4 mg C/l/h throughout the study period. The highest value was recorded in July (0.4 mg C/l/h) and August 1994 and January 1995, while the lowest value was observed in December 1994 (0.1 mg C/l/h). This value was almost uniform in rest of the study months (Tab-2).

Pond -3: In this study pond, the GPP value varied from 0.2 mg C/l/h to 0.4 mg C/l/h throughout the study period. The highest value was observed in the months of May (0.4 mg C/l/h) and July while the lowest value was obtained in the months of November (0.2 mg C/l/h) and April. The GPP of this study pond was found to be fairly uniform in other months of the year (Tab-3).

Pond -4: The GPP values at this study pond, varied from 0.2 mg C/l/h to 0.4 mg C/l/h throughout the study period. The highest value was observed in the months from July to September 1994 (0.4 mg C/l/h) while the lowest value was recorded in the months from February to March 1995 (0.2 mg C/l/h). The pond registered an uniform value of GPP in rest of the study months (Tab-4).

Net Primary Productivity (NPP). The Net Primary Productivity (NPP) of water in the study ponds was found to vary from 0.20 mg C/l/h to(0.21 mg C/l/h.

Pond -1: The NPP values in study pond -1, varied from 0.14 mg C/l/h to(0.14 mg C/l/h throughout the study period. The maximum value was observed in the months from February to March 1995 (0.14 mg C/l/h) which fell to (-0.14 mg C/l/h) in the months of June and from November to December 1994. It rose again in May, January and April 1995 (Tab-1).

Pond -2: In this study pond, the NPP value varied from 0.16 mg C/l/h to 0.21 mg C/l/h throughout the study period. The highest value was recorded in November (0.21 mg C/l/h) while the lowest value was observed in December (-0.16 mg C/l/h). The NPP was found to show an uniform value during the other study months (Tab-2).

Pond -3: In this study pond, the NPP was found to vary from -0.20 mg C/l/h to 0.11 mg C/l/h throughout the study period. The highest value was observed in October (0.11 mg C/l/h) and the lowest value was obtained in February (-0.20 mg C/l/h). In July and November, December 1994 and January and March 1995, the NPP values was found nil again. The NPP of this study pond was found to be fairly uniform during rest of the study months (Tab-3).

Pond -4: The NPP in this study pond varied from -0.17 mg C/l/h to 0.13 mg C/l/h during the study period. The highest value was observed in May and December (0.13 mg C/l/h) while the lowest value was recorded in the months from February to March 0.17 mg C/l/h Uniform value of NPP in rest of the study months was noted at this site (Tab-4).

ZOOPLANKTON

The Zooplankton were represented by the members belonging to Copepoda, Clodocera and Rotifera. In all 38 genera of Zooplankton were recorded from the four study ponds of which 10 belonged to the Copepoda, 9 to Clodocera and 19 to Rotifera. Number of genera of Copepoda, Clodocera and the Rotifera varied from pond to pond during the period of study. Monthly and seasonal abundance of different genera of the Zooplankton groups were presented in tables P-1, -2, -3, -4; Zp. 1a, Zp. 2a, 3a, 3b, 5a, 5b. The Zooplankton were treated separately for each study pond.

Pond -1: In all 27 genera of Zooplankton were observed in this study pond of which 8 genera belonged to Copepoda, 7 to Clodocera and 12 to Rotifera. As regards the number of the genera of Zooplankton under these three group, a low diversity is a phenomenon noted during the study period. In terms of the number of genera the Copepods consisted of 29.6%, Clodocera of 25.9% and Rotifers of 44.4% of the total Zooplankton. The genera of Copepoda were Merocyclops, Mesocyclops, Diaptomus, Cleptocamptus. Cyclops. Paradiaptomus, Phyllodiaptomus and Neodiaptomus. Cyclops was found to occur in higher abundance followed by Diaptomus, Mesocyclops and others. by Alona, Bosmina, Daphnia, The represented Cadocerans were Diaphanosoma, Macrothrix, Moina and Sida. The Cladocerans were dominated by Daphnia with highest abundance followed by Diaphanosoma, Moina and others. The Rotifers were Asplanchna, Brachionus, Chromogaster, Filinia, Harringia, Keratella, Manfredium, Notholca, Philodina, Pompholyx, Rotaria and Trichocera. As regards the monthly and seasonal abundance, Brachionus

was found to be the dominant genus followed by *Keratella* and others. As regards the monthly occurrence, all genera except *Cyclops* of Copepoda were found to occur sporadically during the study period. Of the Clodocerans, *Daphnia* was of regular occurrence while others occurred sporadically. The Rotifers were of sporadic occurrence Table-1; Zp, 1a, 2a, Data on Table-1, Zp-1a, 2a, 3a, 3b 5a, 5b indicates that the genera of Zooplankton groups occurred with varying abundance during different seasons of the year. The Copopeds consisted of 12.64%; 25.69%, 45.43% and 17.98% of the total Zooplankton in the summer, monsoon, post-monsoon and winter seasons respectively. Similarly the Cladocerans consisted of 55.31%, 50.22%, 31.07% and 17.12% of the total Zooplankton population in the four seasons. The Rotifers consisted of 32.13%, 23.87%, 23.47% and 64.89% of the total Zooplankton population in the summer, monsoon, post-monsoon and winter seasons respectively.

Pond -2: A total of 31 genera of Zooplankton were recorded from this study pond during the period of study. The Zooplankton comprised of 10 genera of Copepoda, 8 genera of Clodocera and 13 genera of Rotifera. The Rotifers dominated followed the Copepods and Clodecerans (41.9%, 32.2% and 25.8% respectively). The Cladocerans were represented by Cleptocamptus, Cyclops, Merocyclops, Mesocyclops, Diaptomus, Heliodiaptomus, Paradiaptomus, Phylldiaptomus Naodiaptomus and Neodiaptomus. Cyclops was found to occur in higher abundance (21615 to 138025 units/l) followed by Diaptomus (15770 to 30625 units/l) and others. All genera except Cyclops and Diaptomus were of sporadic occurrence throughout the year. The members of Cladocera were Alona, Bosmina, Ceriodaphnia, Daphnia, Diaphanosoma Moina, Polyophemus and Sida. The Cladocerans were dominated by Daphnia

(23195 to 59915 units/l) followed by Diaphanosoma (8000 to 21074 units/l) and others. Daphnia, Diaphanosoma and Moina occurred throughout the study months while other genera of the group were of sporadic occurrence. The Rotifers were represented by Asplanchna, Brachionus, Chromogaster, Cupelopagia, Filinia, Dorystoma, Harringia, Keratella. Pompholyx, Scaridium, Trichocera and Tripleuclamis. Brachionus occurred in abundance (40260 to 79189 units/l) followed by Keratella (18360 to 193128 units/l) and others. All genera of the Rotifers were found to occur sporadically throughout the study months. Table 2, 2a, 3a, 3b, showing the seasonal abundance of individual genus and group of Zooplankton, indicate that the Copepods constituted of 46.6%, 15.15%, 46.8% and 23.4% of the total seasonal Zooplankton population in summer, monsoon, post-monsoon and winter respectively. Alongside, the cladocerans consisted of 23.7%, 30.5%, 23.9% and 20.7% of the total seasonal Zooplankton population in the four seasons respectively. The rotifers also consisted of 29.5%, 54.4%, 25.8% and 55.8% of the total seasonal Zooplankton in Summer, Monsoon, Post-Monsoon and Winter season respectively. Table-2b indicate that the rotifer with highest abundance (68376 to 246834 units/l) dominated the Zooplankton population with Copepods, rank second (38630 to 175135 units/l) followed by the Cladocerans (63512 to 91773 units/l). The highest total Zooplankton abundance observed in the Winter (442213 units/l) followed by Summer (375454 units/l), Post-monsoon (264824 nits/l) and Monsoon (255918 units/l).

Pond-3: In all 18 genera of Zooplankton were recorded from this study ponds, of which 4 genera belonged to Copepoda, 6 to Clodocera and 8 to Rotifera. Compared to study pond-1 and pond-2. This habitat shows a poor

plankton diversity in respect to the number of genera. The Copepods were represented by Cleptocamptus, Cyclops, Mesocyclops and Diaptomus. Cyclops dominated the Copepods with higher abundance (30578 - 46387 units/l) followed by Diaptomus (3548 - 13785 units/l) and others. Cyclops was found to occur in all study months while the other three genera were of highly irregular occurrence. The Cladocerans were represented by Alona, Bosmina, Diaphnia, Diaphanosoma, Moina and Sida of which only Daphnia with higher abundance (25500 - 45065 units/l) was of regular occurrence. Diaphanosoam ranked second in abundance (8346 - 44646 units/1) was of sporadic occurrence. The Rotifer genera were Brachionus, Chromogaster, Filinia, Keratell, Notholca, Polyarthra, Rotaria and Trichocera. All the eight genera of Rotifera occurred sporadically throughout the study. Keratella was in high abundance (24922 - 79513 units/l) followed by Brachionus (24548 - 52672 units/l). The total plankton abundance in different season indicate (Table-3a) that the Summer and Monsoon population abundance were uniformly high while the Post-monsoon and Winter population were comparatively lower but with almost uniform abundance values.

Pond -4: During the entire study period, this study pond was devoid of Copepoda and Cladocera. In all 12 genera of Rotifera were found to occur regularly throughout the study months. The occurring genera were Ascomorphella, Brachionus, Keratella, Notholca, Philodina, Platyas, Polyarthra, Pompholyx and Trichocera. Brachionus was found to be in higher abundance (13789-99720 units/l) followed by Harringia (12879-63184 units/l), Trichocera (I0I54-54251 unit/l), Keratella (3971-48447 units/l) Polyarthra (16950-33078 units/l) and others. Summer abundance of Zooplankton was higher (424278 unit/l) followed by Post-monsoon (233468 unit/l), Monsoon (122378 unit/l) and Winters (I07953 unit/l). (Table-4, 4a).

PHYTOPLANKTON

Monthly, seasonal and yearly abundance of different groups of phytoplankton have been studied. Percentage composition of the Phytoplankton groups have also been treated. The data were presented in tables 1, 1a, 1b, 2, 2a, 2b, 3, 3a, 3b, 4, 4a, 4b, 5, and 6 and groups were plotted with these data. Phytoplankton biota of the four study ponds are considered separately.

Pond-1: A low diversity of phytoplankton in terms of the number of genera have been observed in this study spots. In all 33 genera of phytoplankton have been recorded from this pond of which 12 genera belonged to Chlorophyceae, 7 to Cyanophyceae, 10 to Bacillariophyceae, to Euglenophyceae and 1 to Dinophyceae. As regards the number of genera of the Phytoplankton, Chlorophyceae comprised of 36.7% of the total Phytoplankton genera followed by Bacillariophyceae (30.3%), Cyanophyceae (21.2%), Euglenophyceae (9%) and Dianophyceae (3%). The Euglenoids were found to be in higher yearly abundance (24868 unit/l & 36.37%) followed by the Chlorophytes (16242 unit/l & 23.75%), Cyanophytes (16159 uni/l & 23.63%), Bacillariophytes (10557 unit/l & 15.44%) and the Dinophytes (537 unit/l & 0.78%). The seasonal abundance of the Phytoplankton groups have been depicted in table 1a, 1b. The data shows that the Euglenoids ranked high in the summer (22835u/l) followed by Chlorophyceae (17979 u/l), Cyanophyceae (16411 u/l), Dinophyceae (1038 u/l) and Bacillariophyceae (9781 u/l). During the monsoon and post-monsoon period similar figures were obtained with slight variation while in the winter the Cyanophytes appeared in higher abundance (18004 u/l) followed by Euglenophyceae (16415 u/l), Chlorophyceae (13118 u/l), Bacillariophyceae (9494 u/l) and Dinophyceae with almost negligible

appearance. The total abundance in the summer, monsoon, post-monsoon and the winter were 66044, 82225, 66123 and 57161 units/l respectively. The 12 genera of Chlorophyceae were Chlorella, Selenastrum, Ankistrodesmus, Crucigenia, Scenedesmus, Pediastrum, Actinastrum, Closterium, Cosmarium, Staurastrum, Ulothrix and Spiragyra. The last two genera were found as surface floating scum. The Cyanophyceae were represented by Microcystis, Spirulina, Oscillatoria, Anabaena, Anabaenopsis and Cylindrospermum. The Bacillariophytes genera were Brebissonia, Fragilaria, Eunotia, Gyrosigma, Navicula, Pinnularia, Gomphonema, Tabellaria, Cymbella and Nitzchia. Three genera of the Euglenoids were - Euglena, Phacus and Trachelomonas while the lone Dinophyte was the - Ceratium. As regards the yearly mean abundance, Scenedesmus ranked highest amongst the Chlorophytes followed by Ankistrodesmus and Chlorella. Amongst the Cyanophytes Microcystis was in high abundance followed by Ocillatoria while amongst the Bacillariophytes Navicula and Pinnularia were in higher abundance. Of the Euglenoids, Euglena was in higher abundance(23980 u/l) which is highest amongst all the Phytoplankton groups. Amongst the Chlorophytes, Staurastrum, Ulothrix and Spirogyra were found to appear sporadically during the study period. Spirulina, Anabaenopsis and Cylindrospermum were registered to be of sporadic appearance. Of the Bacillariophytes, Brebissonia, Fragilaria, Gomphonema, Tabellaria Cymbella and Nitzchia were found to occur sporadically during the study period. Phacus and Trachelomonas of Euglenophyceae and Ceratium of Dinophyceae were also found to occur sporadically throughput the study year. Seasonal occurrence of 12 genera of Chlorophyceae (Table-1a, 1b) shows that all genera were recorded in four seasons except spirogyra being absent in the

post-monsoon. All the 7 genera of Cyanophyceae were recorded in the winter with the exception of *Spirulina* and *Cylindrospermum* in the summer, *Anabaenopsis* and *Cylindrospermum* in the monsoon and Spirulina in the post-monsoon. Of the 10 genera of the Bacillariophyceae 9 genera except *Brebissonia* were recorded in summer while Tabellaria was absent in the monsoon. *Cymbella* was not observed in the post-monsoon while *Gomphonema* and *Cymbella* were absent in the winter season. All 3 genera of Euglenophyceae were recorded in all the four seasons except *phacus* and *Trachelomonas* being absent in the post-monsoon period. *Ceratium* of Dinophyceae was recorded in all the four seasons with varying abundance.

Pond-2: The total number of Phytoplankton genera was 26 of which 9 belonged to Chlorophyceae, 7 to Cyanophyceae, 6 to Bacillariophyceae, 3 to Euglenophyceae and 1 to Dinophyceae. As to the number of genera is concerned, the Chlorophyceae appears with 34.6% of the total genera followed by Cyanophyceae (26.9%), Bacillariophyceae (23%), Euglenophyceae (11.5%) and the Dinophyceae (3.8%). The yearly abundance data (Tab 2, 2a, 2b) show that the Chlorophyceae ranked high (35858 u/l), followed by Bacillariophyceae (28089 u/l), Euglenophyceae (24757 u/l), Cyanophyceae (24086 u/l) and Dinophyceae (3444 u/l) and constituted of 30.84%, 24.17%, 21.29%, 20.72% and 2.96 % of the total Phytoplankton respectively. The seasonal abundance data (Tab 2a, 2b) indicate that the Chlorophytes appeared with higher abundance in the post-monsoon (42177 u/l) and winter (41863 u/l). The Cyanophytes showed higher abundance in the monsoon (24524 u/l) and winter (27754 u/l) period while the Bacillariophytes appeared to with higher abundance in the post-monsoon (49802 u/l). The Euglenophytes showed lower abundance

in the Winter while they appeared with moderately high abundance in the other seasons. The Dinophytes were absent during the Summer and Monsoon and found to present in the Post-monsoon and Winter with higher abundance in the Winter (11085 u/l). The 9 genera of Chlorophyceae were Ankistrodesmus, Scenedesmus, Pediastrum. Crucigenia, Gloeotaenium Actinastrum Closterium, Cosmarium and Micrasterias. As regards to the yearly mean abundance, Actinastrum highest followed ranked Closterium, by Ankistrodesmus, Scenedesmus, Crucigenia and others. The Cyanophytes were represented by seven genera which were Microcystis, Anabaenopsis, Anabaena, Spirulina, Oscillatoria, Lyngbya and Cylindrospermum. Oscillatoria was found to be in higher abundance followed by Microcystis, Anabaena and others. Bacillariophyceae was represented by Fragilaria, Brebissonia, Gomphonema, Navicula, Pinnularia and Cymbella. Navicula was in higher annual abundance followed by Fragilaria, Pinnularia and others. The Euglenoids were represented by three genera- Euglena, Phacus and Trachelomonas, of which Euglena was always found to be in highest abundance amongst the other phytoplankton groups. Ceratium the only genus of Dinophyceae was spotted during the period from October to February with lower abundance. The total seasonal abundance of the Phytopplankton biota shows that (Table 2a, 2b) highest abundance was achieved in the post-monsoon (145523 u/l), followed by the Winter (116341 u/l), Summer (107883 u/l) and Monsoon (92858 u/l). Of the chloroplyceae Ankistrodesmus, Gloeotaenium, Actinastrum and Micrasterias registered sporadic appearance amongst green algae, Microcystis, Anabaenopsis Spirulina, Lingya Blue-Cylindrospermum showed sporadic appearance during the study period.

Brebissonia, Gomophonema and Cymbella were also found to be of sporadic appearance. Phacus and Trachelomonas of Euglenophyceae and ceratium of Dinophyceae were also recorded with sporadic occurrence. The seasonal occurrence of different Phytoplankton genera belonging to different groups were depicted in Table 2a, 2b. The data show that all 9 genera of the Chlorophytes were recorded in Summer and Post-monsoon, while Gloeotaenium and Micrasterias were absent in the Monsoon and Ankistrodesmus and Micrasterias were not recorded in the Winter. All the 7 genera of the Cyanophytes were recorded during the Summer, Monsoon and Winter, while Anabaenopsis, Spirulina and Cylindrospermum were not recorded in the Post-monsoon period 6 genera of the Bacillariophyceae were recorded during all the four seasons except Gomphonema being about in the Monsoon. The 3 genera of Euglenophyceae were found to occur in all the seasons while only Trachelomonas was absent in the Monsoon. Ceratium of Dinophyceae was spotted in the Post-monsoon and Winter season only.

Pond -3: In all 30 genera of Phytoplankton were recorded from this study pond, of which 11 genera belonged to the Chlorophyceae, 7 to Cyanophyceae, 10 to Bacillariophyceae and 2 to Euglenophyceae. No Dinophyte was recorded from this pond during the period of study. As regards the total number of genera the Phytoplankton biota, Chlorophyceae comprised of 36.7% followed by Bacillariophyceae (33.3%), Cyanophyceae (23.3%) and the Euglenophyceae (6.6%). The yearly abundance of different groups of Phytoplankton shows that Cyanophytes ranked highest (39347 u/l) followed by Bacillariophyceae (27290 u/l), Euglenophyceae (14639 u/l) and Chlorophyceae (13098 u/l) with a percentage composition of 41.7%, 28.9%, 15.5% and 13.87%

of the total Phytoplankton respectively (Table 3a, 3b). The seasonal abundance of different groups of Phytoplankton were depicted in Table 3a, 3b. The data show that the Chlorophytes were in moderately high abundance during the Summer (19703 u/l) and Monsoon (17932 u/l). The Chlorophytes registered lower abundance during the Post-monsoon and Winter periods. Cyanophytes registered higher abundance during the Summer (46382 u/l) and Post-monsoon (41194 u/l) with moderately high abundance in Monsoon (33529 u/l) and Winter (35946 u/l) also. The Bacillariophytes showed highest abundance during the Winter (37208 u/l) and during the Summer, Monsoon and the Post-monsoon they registered almost uniform abundance values. The Euglenoids showed highest abundance during the Post-monsoon (28928 u/l) and Summer (23781 u/l) with low abundance in the other seasons. The Cyanophytes appeared to be with highest seasonal abundance followed Bacillariophyceae, Euglenophyceae and Chlorophyceae during the study period. The highest total seasonal phytoplankton was recorded in Summer (114396 u/l) followed by Post-monsoon (105668 u/l), Winter (83704 u/l) and the Monsoon (72746 u/l). The Chlorophyceae was represented by Eudorina, Scenedesmus, Pediastrum, Chlorella, Tetraedron, Selenastrum, Ankistrodesmus, Crucigenia, Staurastrum, Euastrum and Cosmarium, Scenedesmus was found to be in higher abundance amongst the Chlorophyceae followed by other genera of the group. It was noted that all the 11 genera of this group were found to occur sporadically in this pond during the entire study period. The yearly mean (Table 3a, 3b) abundance of these genera showed lower individual abundance values. The Cyanophytes were represented by Microcystis, Spirulina. Oscillatoria, Anabaenopsis, Nodularia, Arthrospira and Cylindrospermum. Oscillatoria was recorded with highest abundance (29249 u/l) followed by

Microcystis Anabaenopsis and others. All the Cyanophycean genera except Oscillatoria were recorded with sporadic appearance during the period of study. The Bacillariophytes were represented by Navicula, Fragilaria, Synedra, Gyrosigma, Eunotia, Achnanthes, Cymbella, Nitzschia and Tabellaria. Of these Fragilaria apeared to be in higher abundance followed by Pinnularia and Navicula, Fragilaria and Pinnularia were recorded every month during the study period while the rest of the genera were found to be of sporadic appearance. The Eugenoids had only Euglena and phacus in record, Euglena ranking high with respect annual abundance. Both the genera were of sporadic occurrence. The seasonal occurrence of all the 11 genera of Chlorophyceae (Table - 3a, 3b) shows that all genera were recorded in the Summer with varying abundance values but Eudorina, Tetraedron, Selenastrum and Staurastrum were not recorded during the Monsoon season. During the Post-monsoon only four genera i.e. Scenedesmus, Ankistrodesmus, Crucugenia and Cosmarium were recorded while Eudorina, Chlorella, Tetraedron, Selenastrum and Staurastrum were found to occur in the winter. Of the 7 genera of the Cyanophytes, all were recorded in the Summer while Anabaenopsis, Nodularia and Arthrospira were absent during the Monsoon. Spirulina was found to be absent in the Post-monsoon, while Nodularia and Cylindrospermum were not recorded in the Winter. All 10 genera of Bacillariophyceae were recorded in the Winter while Synedra was not recorded in the Summer. During the Monsoon Synedra, Cymbella and Tabellaria were absent. In the Post-monsoon Gyrosigma, Achnanthes and Nitzchia were not recorded. Of the Euglenoids. Phacus occurred in all the four seasons but Euglena was absent only in Monsoon

Pond-4: In all 30 genera of Phytoplankton were recorded from this study spot, of which 11 genera belonged to Chlorophyceae, 8 to Cyanophyceae, 9 to Bacillariophyceae and 2 to Euglenophyceae. No member of Dinophyceae was recorded from this pond. As regards the total number of genera, Chlorophyceae constituted of 36.7% followed by Bacillariophyceae (30%), Cyanophyceae (26.6%) and Euglenophyceae (6.6%). As regards the yearly abundance and percentage of differnt groups of phytoplankton, Cyanophyceae ranked high above others (34334u/l) others followed by Bacillariophyceae (24365 u/l), Euglenophyceae 21124 u/l) and the Chlorophyceae (16122 u/l), and percentage of composition being 35.78%, 25.39%, 20.01% and 16.8% respectively (Table 4, 4a, 4b). Data on seasonal abundance of different groups of phytoplankton is presented in table 4a, 4b, which indicate that the Chlorophytes were of high abundance during the Winter (26603 u/l) with moderately high abundance in the Post-monsoon (20218 u/l) while in the Summer and Monsoon, they were found to be with lower abundance values. The Cyanophytes showed uniformly high abundance ranging from 32008 to 38698 u/l in all the seasons. The Bacillariophytes recorded highest abundance (55590 u/l) in the Winter which lower abundance was recorded in other seasons. The Euglenoids appeared with high abundance (28925 u/l) in the Monsoon while they maintained a moderate value in other seasons. The total seasonal abundance was recorded in the Winter (134057 u/l) followed by Post-monsoon (85478 u/l), Summer (85473 u/l) and Monsoon (77776 u/l). The Chlorophytes were represented by Chlorella, Actinastrum, Crucigenia, Ankistrodesmus, Selenastrum, Scenedesmus, Pediastrum, Staurastrum, losterium, cosmarium, and Micrasterias. Ankistrodesmus, and Crucigenia were found to occur throughout the study year but all other genera Occurred sporadically. Ankistrodesmus showed higher annual abundance followed by Chlorella, Crucigenia, Selenastrum and others of this group. The Cyanophytes were represented by Microcystis, Spirulina, Oscillatoria, Lyngbya, Anabaena, Gloeotrichia, Cylindrospermum and Nodularia. All these genera except

Oscillatoria occurred sporadically throughout the study period. Oscillatoria was found to be in higher annual abundance (9967 u/l) followed by Micracystis, Anabaena and others. Members of Bacillariophyceae occurring during the period of study were Rhopalodia, Navicula, Pinnularia, Synedra, Gyrosigma, Eunotia, Rhizosolenia, Melosira and Achnanthes. Synedra although appearing sporadically, showed higher abundance annually followed by Pinnularia, Navicula, Eunotia and others. All the genera of this group except Pinnularia were found by occur sporadically throughout the study period. Euglenophyceae was represented by sporadic appearance of Euglena and phacus. Euglena occurred with high annual abundance (20688 u/l) while phacus was of negligible occurrence. The seasonal occurrence of Chlorophytes shows (Table-4a, 4b) that of the 11 genera Chlorella was not recorded in the Summer while Actinastrum. Selenastrum Pediastrum. Closterium. Cosmarium Micrasterias were abent in the Monsoon. In the Post-monsoon Staurastrum, Closterium and Micrasterias were not recorded while all the 11 genera were observed in the Winter. Amongst the 8 Cyanphyte genera Spirulina was not recorded during the Summer while gloeotrichia and Nodularia did not occur in the Monsoon. In the Post monsoon Cylindrospermum and Noclularia were not spotted while Anabaena was found to be absent in the Winter except other marked by the presence of all 9 genera of genera. Summer was Bacillariophyceac, but Rhopalodia, Navicula, Rhizosolenia and Achnanthes were absent in the Monsoon period. During the Post- monsoon all genera except Rhizosolenia and Achnanthes were present while in the Winter only Achnanthes was not spotted. The two Euglenoid genera Euglena and phacus were observed in all seasons except Phacus disappearing only in the Monsoon. To see the interrelationship between the physico-chemical parameters and that between the phyto- and zooplankton, the correlation quotient (r) has been calculated and the results were presented in table 1r, 2r, 3r and 4r.

CHAPTER -4

DISCUSSION



DISCUSSION

Studies of the physico-chemical characteristics of the four study ponds reveal that the physico-chemical variables showed monthly and seasonal changes. Sometimes spatial or sudden changes of the variables were also noticed in the study ponds.

Temperature

Air and water temperature plays significant role in physico-chemical, metabolic and physiological behaviour of aquatic bodies. Changes in temperature affects many physiological processes including density of water and thermal stratification (Lund and Talling, 1957). Temperature may be less important as a factor in unpolluted water body, because of the wide range of temperature tolerance of aquatic organisms. But in a polluted pond or river system, temperature changes may result in increased metabolism of aquatic organisms which may have a profound effect on DO and BOD affecting the aquatic biota including the fishes (Land and Talling, 1957, Palharya et. al., 1993).

Solubility of oxygen is dependant upon the prevailing water temperature. With the increase in water temperature, the solubility of oxygen is reduced causing deoxygenation. Temperature fluctuation may also be caused by the activities of micro-organisms (Willen, 1955; Hutchinson, 1957; Welch, 1948, Trivedy, 1990).

During the period of study in the four ponds, the air temperature variation was almost similar as the study ponds were situated in the same place and spread over a radious of 250 meters. The diurnal and monthly fluctuations were found to be almost similar in the study ponds with some deviations depending upon the average depth of water in the respective ponds. Seasonal changes were apparent in

the study ponds. Maximum values of water temperature were observed in the Summer season and minimum in the Winter, corresponding with the atmospheric temperature. During the Monsoon and the Post-monsoon period, the air as well as water temperature were found to be similar with slight fluctuation. During the Summer the water temperature was higher because of low average depth, clean atmosphere and greater sunshine hours. Range of fluctuation of water temperature was minimum probably because of rainfall, frequent clouds and minimum sunshine hours in the rainy months. Similar views were also held by V yas (1968) and Palharya et. al.(1993). In the present study the seasonal temperature of water ranged from 29-31.8°C in Summer, 28-28.9°C in Monsoon, 24.1-30.4°C in Post-monsoon and 19.3-23.9°C in the Winter in pond-1., while it ranged from 31.8-32.8°C, 29.7 - 32.1°C, 23.1 - 31.0°C and 18.4 - 24 °C in the Summer, Monsoon, Post-monsoon and Winter respectively in pond-2. The temperature in pond -3 ranged from 29.3 - 32.5 °C in Summer, 29.8 - 31.9 °C in Monsoon, 22.9 - 29.2 in Post-monsoon and 19-24.3 °C in the Winter, while in pond -4, it ranged from 29.4 - 32.5 °C, 29.7 - 31.6 °C, 23 -29.6 ^oC and 19.1 - 24 ^oC in Summer, Monsoon, Post-monsoon and Winter respectively. Similar results were obtained by Srivastava (1956) Khanna (1993), Badal & Singh (1981), Moitra and Bhattacharya (1965), Das & Pandey (1978) and many other scientist.

Average Depth

Average depth of water in study ponds showed slight variations, as the water level exhibited minimum fluctuation. All the ponds are fed with the water of deep tube well pump at regular intervals to keep the level upto desired mark. Pond-1 is a shallow habitat while pond -3 is the deepest amongst the study

ponds. The data in tables 1, 2, 3 and 4 indicate that the water level did not show any appreciable decline throughout the year except a rise in the monsoon period. The seasonal ranges of average depth in pond-1 were 1-1.1, 1-1.2, 0.9 - 1.2 and 1-1.1 metres in Summer, Monsoon, Post-monsoon and Winter respectively, while in pond -2, the average depth varied from 1.4-1.8, 1.5 - 1.8, 1.4 - 1.5 and 1.4 - 1.6 metres in the four seasons respectively. In pond -3, the average depth was noticed to range from 1.8 - 2.2, 2.1 - 2.3, 1.7 - 2.0 and 1.7 - 1.9 metres in Summer, Monsoon, Post-monsoon and Winter respectively. Similarly in Pond -4, it ranged from 1.6 - 1.9, 1.6 - 2, 1.4 - 1.5 and 1.4 and 1.7 metres in the four seasons respectively. In water bodies fed regularly with pumped water monthly and seasonal changes are seldom observed with noteworthy uphill downhill peaks. The present study ponds have been found to show such phenomenon. Thermal stratification in the study ponds could not be observed as the water in the pond are occasionally agitated by aerator or hard lime across resulting in the thorough mixing of the water. This resulted in a temperature equilibrium in the study ponds at all depths.

Secchi disk depth

Secchi disk depth of the study ponds also exhibited striking similarity in having similar values throughout the year, Although monthly variations were noted, seasonal variations were found to be of very short range. In all the study ponds very low SDD values were noted because of turbidity caused by regular changing of the pond water with various organic and inorganic materials used for fertilization of these culture ponds. Regular and frequent agitation of the water might have also contributed to this high turbidity or low value of SDD or light penetration into the study waters. The suspended particles, soil, silt

particles, decomposed or undecomposed organic matter, total dissolved solids as well as abundance of the microflora and micro-founa, are regarded as main source of turbidity in water which interfere with the penetration of light. (Gautom, 1990). These reasons may be attributed to the low values of light penetration in the present study pond. The seasonal ranges of the SDD values were very narrow in all the study ponds as depicted in (Tables 1, 2, 3 and 4).

Total dissolved solid

Total dissolved solid or TDS are caused by suspended particles inside the water body influencing its turbidity and light penetration. (Gautom, 1990, Tripathy and Pandey, 1990, Palharya et. al. 1993, Khanna, 1993). Palnkton also show a striking parallelism with suspended solids. Consequently the higher abundanced of total plankton in the study ponds may be one of the reasons for higher TDS values. Similar views were also expressed by Tripathy and Pandey, 1993 and Singh, 1995. The fluctuation of TDS values in the four study ponds may be attributed to the above reasons although higher TDS values are often encountered in the rainy season which may be due to the run off water reaching the ponds. As the study pond water is regularly and frequently agitated for oxygenation, there is less chance left for the suspended particles to settle down on the bottom, thereby leaving the water permanently turbid. Lower values of phytoplankton and lower rate of primary productivity may be caused due to the permanent turbidity in the study waters. Similar views were expressed by David, 1965, Motwani et. al. 1956, Verma and Sukla, 1969, Sharma, 1981 and Tripathy and Pandey, 1993. Ganzalves and Joshi (1946) observed that total solids varied in proportion to the temperature and rarely varied inversely to the water level. Seasonal fluctuation of TDS values in the four study ponds were

depicted in Tables 1, 2, 3 and 4. In Pond -1, the seasonal TDS values ranged from 420.2 - 489.5, 454.7 - 494, 353 - 488.3 and 440.4 - 484.3 mg/l in the Summer, Monsoon, Post-monsoon and the Winter respectively; while in pond-2, the values were found to range from 404.3 - 436.2, 383 - 455, 435 - 499 and 406.3 - 482 mg/l, in the four seasons respectively. In pond -3 the TDS values ranged from 397.2 - 445.9, 381.6 - 419.3, 387.7 - 448.5 and 441.7 - 450.1 mg/l in the Summer, Monsoon, Post-monsoon and the Winter respectively; while in pond -4, the values ranged from 351.3 - 378.6, 354.9 - 403.3, 370.8 - 404.6 and 397.5 - 470.6 mg/l in the four season respectively. It appears from the seasonal TDS values in the study ponds that a common factor complex may be at play for their striking similarly.

Electric conductivity

The Electric conductivity of water depends upon the concentration of ions and its nutrient status. Conductivity of the study ponds is a characteristics associated with the dissolved materials or solutes concentration present in the study waters. During the present investigation the conductivity values were always high (543 - 753 μ moh/cm in pond -1, 422 - 767 μ moh/cm in pond -2, 585 - 693 μ moh/cm in pond -3 and 541 - 724 μ moh/cm in pond -4). The conductivity values in all the study ponds showed a regular pattern of monthly and seasonal fluctuation during the study time. The higher values of electric conductivity indicate that all the study ponds had higher concentration of dissolved materials or solutes. Similar views were expressed by Gautom (1990), Trivedy et. al. (1990) and many other scientists. During the present study the higher conductivity values may be due to addition of considerable quantity of minerals and charging materials throughout the whole year. Trivedy (1990)

expressed similar views. The seasonal fluctuation of the conductivity values in pond -1, ranged from 647 - 753, 700 - 751, 543 - 752 and 678 - 745 μ moh/cm. in Summer, Monsoon, Post-monsoon and Winter respectively, while in pond -2, the values were 422 - 671, 443 - 700, 529 - 767 and 677.5 - 741 μ moh/cm in the four seasons respectively. Similarly, the E. C. values ranged in pond -3, from 611 - 686, 587 - 645, 597 - 690 and 680 - 693 μ moh/cm and in pond -4, seasonal ranges were 541 - 583, 546 - 621, 571 - 623 and 612 - 724 μ moh/cm in Summer, Monsoon, Post-monsoon and Winter seasons respectively. In pond -4, highest conductivity values were recorded in the Winter while those in other seasons were found to be uniform but high. The present investigation testifies that the study ponds remain uniformly rich in higher concentration of dissolved material or solute throughout the whole study year indicating their abundant nutrient status.

Hydrogen ion Concentration (p^H)

The p^H or hydrogen ion concentration is one of the most important variables that controls the aquatic environment. All chemical and biological reactions are dependent upon the hydrogen ion concentration of the reaction system (Ghosh & Sharma, 1987; Gautom, 1990; Trivedy, 1990; Tripathy & Pandey, 1993, Palharya et. al. 1993). Sreenivasan (1977) expressed the view that there is a direct relationship amongst the hardness, alkalinity and growth of phytoplankton consequent upon the changes in the p^H in water. Anderson (1961), Jonassan (1969), King (1972), Patil (1982), Singhi (1986), Palharya et. al. (1993) viewed p^H value as an important factor in maintaining the carbonate and bicarbonate system in a body of water which plays a vital role in the formation of algal blooms. Valladolid et. al. (1954) observed that an alkaline p^H

provides an optimum condition for congenial growth of the plankton biota. The present study reveals that the pH value of the water of four study ponds were relatively higher throughout the year. The pattern of monthly and seasonal fluctuation of pH values in the study ponds are found to be regular, maintaining an uniform value. No season is marked as having highest pH values than the other. As a matter of fact, high pH values were found to be maintained in the four seasons which can be attributed to moderately high productivity in the standing water of the study ponds in having ability to utilize carbon from carbonate, sulphur from sulphate, nitrogen from nitrate, phosphorus from phosphate thereby converting them into hydroxyl ion with an increase in the pH, keeping the total alkalinity and hardness values of water nearly constant (Zafar, 1966). These reasons may be at play in the study ponds having higher pH values and moderately constant high hardness and total alkalinity values. The moderately lower pH value in the Summer season in the four study ponds may be probably due to high turbidity and higher temperature in the Summer, might have enhanced the microbial activity causing higher production of CO₂ and reducing the p^H. Khalaf and Macdonald (1975) also established that at high turbidity and high water temperature, photosynthetic process is impared considerably leading to accumulation of CO₂ that causes lowering of the p^H level by forming carbonic acid. The above factors might be at play in the present study ponds in the Summer period as evidenced by the CO2 and temperature values during the study period. However, as the average value of pH is concerned, the four study ponds were always alkaline in nature and pH value never came down to less than 7.7. The seasonal changes of the pH values in pond -1 ranged from 7.7 - 8.7, 8.3 - 8.5, 8.5 - 8.7 and 8.6 in the Summer, Monsoon, Post-monsoon and Winter respectively; while in pond -2, the values ranged from 7.8 - 8.9, 8.2 - 8.4,

8.4 - 8.7 and 8.7 - 8.9 in the four seasons respectively. In study pond -3, the pH value never came down below 8.2 during the study period, the seasonal range being 8.2 - 8.9, 8.3 - 8.4, 8.6 and 8.6 - 8.8 in Summer, Monsoon, Post-monsoon and Winter respectively; on the other hand pond -4, showed similarly higher alkaline nature in having a seasonal range of 8.4 - 8.9, 8.1 - 8.5, 8.5 - 8.6 and 8.6 - 8.9 in the four seasons respectively. Zafar (1964), Buck and Rankin (1972) and Gautom (1990) expressed the view that during the Summer and Monsoon months the pH values remain lower, and higher during the Winter. The higher values of pH in Winter is thought to be due to increased photosynthesis activity by the phytoplankton and other hydrophytes. Lakshminarayana et.al. (1963), Nautiyal (1984), Doyle (1964) and Badalex &S Singh (1981) supported this fact by their respective findings. They held that lower values of pH during the Summer and Monsoon months was due to decreased photophythetic activity caused by high turbidity affecting penetration of light and the phytoplankton populations. The above causes are prevented in the present study ponds but on the contrary the pH level remained constantly high in all the seasons of the study year. This fact may be explained by the phenomenon of over charging of the study ponds with lime at frequent intervals resulting in higher alkalinity values. The factors stated above are to taken into consideration in explaining the striking behaviour of p^H values in the study ponds.

Free CO₂

Carbon-dioxide in a water system is generally derived from various sources e.g. the atmosphere, respiration of plants and animals, bacterial decomposition of organic matter in suspension and in bottom, inflow of ground water which seep into the ponds, lakes and streams, and from within the water

itself in combination with other substances mainly calcium and magnesium, Hutchinson (1957), Golterman (1975) and Jhingran (1975). Free CO₂ combines with water chemically and forms carbonic acid which affects the pH content of water. Carbonic acid dissociates partly and produce (H⁺) and bicarbonate (HCO3) ions. The bicarbonate ions dissociate further forming more hydrogen and carbonate ions. Mostly the carbon-dioxide is always present in the form of HCO₃ and CO₃. When the p^H is low, the combined carbon-dioxide is converted into the free form and when there is an increase in the bicarbonate and carbonate, water becomes alkaline and resists hydrogen ions. The CO₂, p^H and alkalinity are directly related to each other, since the pH depends upon the free carbon-dioxide and the bicarbonate-carbonate level (Michael, 1964; Golterman 1975; Kern, 1960; Emerson et. al., 1973 and Hutchinson, 1957). The higher growth of phytoplankton and other hydrophytes cause the enhanced uptake of CO₂, HCO₃ and CO₃, thereby increasing of p^H of water (Hutchinson, 1957, Stumn and Morgan, 1970) Free CO₂ occurs in water as dissolved gas, carbonic acid, carbonates and bicarbonates of calcium and magnesium. All these substances are in chemical equilibrium with one another (Hutchinson, 1957). The absence of free CO₂ may be either due to its complete utilization in photosynthesis or may be due to inhibition by the presence of appreciable amount of carbonates in water (Srinivasan, 1965; Sahai and Sinha, 1969).

Comparatively maximum value of free CO₂ was noted in the study pond -1, (2.8 - 34.7 mg/l) with highest value in the Summer (21.5 - 34.7 mg/l) followed by Winter (15.7 - 33.2 mg/l), Post-monsoon (23 - 26.7 mg/l) and Monsoon (2.8 - 10.1 mg/l). Highest value in Summer and moderate high values in the Post-monsoon and Winter may be due to high rate of decomposition of

organic matter by microbes, decreased of photosynthetic activity for low abundance of phytoplankton, and high rate of respiration by the biota therein. Consequently, the zooplankton was higher than that of the phytoplankton in the Summer, Monsoon, Post-monsoon and Winter resulting in higher rate of respiration raising the free CO2 content in the study pond. The lower value of free CO2 in the Summer may be probably due to reduced rate of decomposition of organic matters. Similar views were expressed by Adoni (1985), Sinha (1988), Pandey et:al. (1988) and Hegde and Bharti (1985). In study pond -2, the monthly and seasonal peaks are noteworthy although the CO2 content registered a moderately low value compared to pond -1. The seasonal range was found to be from 4.3 - 14.7, 8.2 - 8.9, 8.3 - 19 and 4.9 - 32.5 mg/l in the Summer, Monsoon, Post-monsoon and the Winter. In pond-3, the free CO₂ content showed noticeable fluctuations in the consecutive study months. The seasonal fluctuation ranged from 1 - 14.6, 0 - 4.3, 2.9 - 6.3 and 0 - 6.4 mg/l in the Summer, Monsoon, Post-monsoon and the Winter respectively, while in pond -4, the CO₂ values ranged from 0 - 2.3 mg/l in Summer, zero in Monsoon, 0 - 4.3 mg/l in Post-monsoon and 0 - 5.7 mg/l in the Winter.

Carbonate alkalinity

The values of CO₃ alkalinity as depicted in Tables 1, 2, 3 and 4 indicates that the four ponds under study, were very low. In pond -1, the CO₃ alkalinity values were found to vary from month to months and season to season. The seasonal fluctuation ranged from 0 - 17.2 mg/l in the Monsoon, and 0 - 2.8 mg/l in the Winter, with zero values recorded in the Summer and Post-monsoon. Khan and Siddiqui (1970), and Trivedy (1982) expressed the view that the fluctuations in CO₃ alkalinity values were mainly due to photosynthetic activity

of aquatic plants and phytoplankton. Bhatnagar and Sharma (1973) observe that the presence of carbonate alkalinity is an indication of higher rate of carbon assimilation. Zero value of CO3 alkalinity in the Summer and Post-monsoon in the study pond -1, may be probably due to the reduced photosynthesis rates and higher number of microorganisms, resulting into higher concentration of free CO2. Consequently the contents of CO3 in this study pond in Summer and Postmonsoon were 21.5 - 34.5 mg/l and 23 - 26.7 mg/l. Similar results were obtained by Zafar (1966) and Braj Nandan (1985). Similarly, the seasonal CO₃ alkalinity values ranged in p -2, from 0 - 15.2, 0 - 4.8, 0.9 - 2.8 and 0 - 10.8 mg/l in Summer, Monsoon, Post-monsoon and Winter seasons. In study pond -3, a very low CO3 alkalinity value was noted with no zero value in any month or season. The seasonal fluctuations ranged from 1.8 - 4.8, 1.2 - 1.8, 1.7 - 3.3 and 1.4 - 8 mg/l in the four seasons respectively; while in study pond -4, these values ranged from 4.9 - 6.6, 2.7 - 5.8, 1.8 - 4.8 and 4.3 - 7.7 mg/l in the four seasons respectively. The lower value of CO₃ alkalinity or phenolphthalein alkalinity in the study ponds are indication of lower rates carbon assimilation (Tripathy and Pandey, 1993).

Bicarbonate alkalinity

During the present study, the bicarbonate alkalinity showed two distinct high peaks in the Summer and Winter seasons. The Monsoon and Post-monsoon periods experienced moderately lower values compared to Summer and Winter. All the four study ponds exhibited the similar phenomenon. Similar results were obtained by Alikhuni (1957) and Basu et. al. (1970) in their works. The accumulation of large quantity of bicarbonate during the Summer and Winter months associated with excess quantity of free carbon-dioxide produced in the

process of decomposition may be the reasons for higher values of bicarbonate alkalinity in the four study ponds in Summer and Winter. Munawar (1970), Ruttner (1953), Saha and Pandit (1986), Sunder (1988), Hegde and Bharati (1985) expressed similar views in connection with their works. They also opined that high values of bicarbonate of calcium and magnesium in summer could be the reason for high bicarbonate alkalinity values when the water temperature was also high, while the Winter peak was due to increased activity of the phytoplankton. Bicarbonate content in the present study ponds could be also governed by the above factors also. Mention may be made of the liming of the ponds at the rate of the 750 kg/ha/year in first culturable year followed by 2300 kg/ha./year in second culturable year (personal communication from the hatchery authority). The liming may be regarded as one of the causes of higher alkalinity in the study ponds. Lower carbonate values in the Monsoon and Postmonsoon months were may be due to dilution effect and conditions less congenial for the photosynthetic conversion to carbonate. Rao (1970), Trivedy (1982), Braj Nandan (1985), Hegde and Bharti (1985) and Chakrabarty et.al. (1966) also reported similar results in their works and expressed the view that dilution is an important factor in lessening the bicarbonate alkalinity of water. In the present study pond -1, the seasonal bicarbonate values were found to range from 30.5 - 403, 21 - 49, 114 - 163.3 and 184 - 301.2 mg/l in Summer, Monsoon, Post-monsoon and Winter seasons. In study pond -2 the values varied from 21.2 - 356.7, 18.7 - 41.3, 109.7 160.5 and 171.4 - 290.1 mg/l in the four seasons respectively. Similarly, in pond -3 the seasonal range of the bicarbonate values were 25.6 - 445.2, 15.1 - 81.3, 113.8 - 134.8 and 156.4 - 283.7 mg/l in the four seasonal respectively. In pond -4, comparatively lower seasonal bicarbonate values were obtained which ranged from 21.3 - 366.6, 15.2 - 78.

102 - 130.4 and 133.5 - 253.3 mg/l in the Summer, Monsoon, Post-monsoon and the Winter periods. The higher values of the bicarbonate alkalinity in the present study showed that all the study ponds possessed hard water throughout the period of investigation indicating their highly eutrophic nature. Many workers including Blume (1957), Barret (1953), Vass et.al. (1977), reported similar findings in their respective investigations.

Total hardness

Total hardness denotes the quality of water. In general, the hardness of water is mainly due to the presence of calcium and magnesium salts as bicarbonates, carbonates, sulphates and chloride and also cations of iron, manganese and strontium contribute to it. Hardness may be temporary due to bicarbonate and carbonate salts of the cations or permanent mainly due to sulphates and chlorides of metals i.e. calcium and magnesium (APHA - 1987). Since calcium and magnesium are principal cations causing hardness to water, the hardness may be expressed as their CaCO₃ equivalent. According to APHA (1987) and Bridge and Judy (1911) - soft water contains small amount of Ca ++ and mg++ in solution having bound CO₂ not exceeding 5 mg/l; a medium class water which contains bound CO₂ ranging from 5 - 22 mg/l and hard water which contains more than 22 mg/l of bound CO2 and even exceed 50 mg/l. Ruttner (1953) and Wetzel (1975) said that the total hardness is the total amount of alkaline earths present without reference to the particular anions to which they are bound. Calcium ions are one of the most important components of the plant tissues and participate in various cellular metabolism, including the translocation of carbohydrates and also facilitate the availability of other ions, Wetzel (1975) and Livingtone (1963).

Magnesium is required by chlorophyllous plants as the magnesium-porphyrin component of the chlorophyll molecules and as co-factor for various enzymatic transformation within the cell, especially in the transphosphorylation in algal, fungal and bacterial cells (Wetzel, 1975; Palharya et.al., 1993; APHA, 1987; Hutchinson, 1957 and Ruttner, 1953).

During the present investigation, the total hardness values showed monthly and seasonal fluctuation. The seasonal fluctuation were noteworthy. In all the study ponds, the total hardness values were very high in the Summer and Winter months while the Monsoon and Post-monsoon values were moderately high also. The Summer and Winter peaks may be due to high water temperature and low water levels respectively. Dilution caused by rain water might be responsible in lowering the hardness values in the Monsoon and Post-monsoon seasons. Similar views were held by Munwar (1970), Moss (1973), Sreenivasan (1974), Patil (1982) and Azmal and Din (1988). Higher values of total hardness in the study ponds may probably be due to rapid oxidation of organic matter in the pond water. Consequently, the study ponds are regularly charged with cow dung, bran, oil cake and blood for the stocking brood fish at the rate of 2-3% of the total body weight of total fish stock per day. As a matter of fact, brood fish stock density in the study ponds is more than 10 metric tone/ha/year. To support this over stacking higher amount of organic matter is used in the ponds as food supliments. Palmer (1967), Olaniya and Saxena (1977), Singh and Bhowmick (1985), Rai (1974) and Dutta (1988) also expressed the views that rapid oxidation of organic matter may raise the vales of hardness especially calcium hardness.

During the present investigation the seasonal variation of total hardness in pond - 1, was found to be 209 - 538; 200 - 295; 209 - 285 and 386 - 467 mg/l; in pond -2, the values ranged from 285.5 - 609.2, 226.2 - 307, 214.2 - 317.7 and 414.6 - 487 mg/l; in pond -3, the values ranged from 240.2 - 608.8, 228.7 - 273, 238.7 - 271.7 and 424 - 466.2 mg/l while in pond -4, they ranged from 263 - 469.3, 197.8 - 278, 216 - 232 and 317.3 - 372.2 mg/l in the Summer, Monsoon, Post-monsoon and Winter seasons respectively. According to Klein (1956), Sawyer (1960) and Sinha (1988) water with 50 - 150 mg/l hardness content is moderately hard; 150 - 300 mg/l hardness content is hard; while a hardness value above 300 mg/l is considered to be very hard. Based with the above rating, the present study ponds were having moderately hard to very hard water.

Chloride

Chlorides in the form of Chloride ions are one of the major inorganic anions present in natural waters. Waters containing 250 mg/l of Chloride of sodium may have a detectable salty taste. The typical salty taste may be absent in a water containing as much as 1000 mg/l Chloride when the cations are calcium and magnesium (APHA, 1987). Ganapati (1943), Moyle (1956) and Zafar (1964) described Chloride as an indicator of pollution. During the present investigations all the four study ponds showed moderately high and uniform values of Chloride content. Monthly and seasonal fluctuations were of short range throughout the study year. Regular charging of the study ponds with organic feed for brood fishes may be one of the reasons for such higher values of Chloride in the ponds. Zafar (1964), Rao (1971) Prasad and Qayyum (1976), Sunder (1988), Azmal (1988) and Singhi (1986) reported higher content of

Chloride of organic origin in their works. Munawar (1970) Bhatnagar and Sharma (1973), Trivedy (1982) and Adoni (1985) suggested that higher concentration of Chloride in water is an index of pollution of animal or organic origin and there is a direct correlation between Chloride concentration and pollution levels. Thresh et.al. (1944) also held that high Chloride content are indications of a large amount of organic matter in the water. Krul Liefrinck (1966) held that about 50 mg/l of Chloride content in natural or culture water indicates medium level of pollution. The present study ponds can, therefore, be termed as moderately polluted as far as the Chloride values in the study months and seasons are concerned. The seasonal range of Chloride values in pond -1 were from 112 - 131, 103 - 143, 99 - 109 and 87 - 107 mg/l in Summer, Monsoon, Post-monsoon and the Winter respectively; while the values ranged from 103.6 - 138.7, 121.6 - 191.7, 129.8 - 146.7 and 104.6 - 121.7 mg/l in pond -2 in the four seasons respectively. Pond -3 registered, comparatively high Chloride values than the other three ponds and the seasonal values were found to range from 132.4 - 156.7, 103.4 - 142, 107.5 - 132.7 and 178 - 201.3 mg/l in Winter period Summer, Post-monsoon and respectively. Monsoon, Comparatively lower monthly and seasonal values were obtained in study pond -4 which ranged from 98.6 - 121.6, 109.2 - 137, 113.6 - 142 and 93.4 -121 mg/l in the four seasons. As regards the obtained values of Chloride in the study ponds, no particular season could be termed as having uphill or downhill peaks during the period of investigation. However, there have no definite pattern of fluctuation of the Chloride values in the study ponds. As for the moderate value of Chloride is concerned in the study ponds, the level could be termed as of biological importance, a view expressed by many workers including

Jayangouder (1964), Lakshminarayana (1965), Gonzalves and Joshi (1946) and Venkateswarlu (1969).

Dissolved Oxygen

Dissolved Oxygen (DO) is one of the most important chemical factor in water which reflects the physical and biological processes prevailing in a body of water. It is also very important in the assessment of water quality standards. The presence of DO is essential to maintain all forms of life in water. The main source of DO in water is the diffusion from the atmosphere and the photosynthetic activity of the hydrophytes including the phytoplankton. A running water contains relatively higher concentration of DO under natural condition than a standing water system (Welch, 1952). DO level in water comes down in a water body due to respiration of the biota, decomposition of organic matter, addition or inflow of oxygen deficient waste water or effluent, inorganic reductants such as hydrogen sulphide, ammonia, nitrates, ferrous ion and other oxidizable substances (Welch, 1952, Wetzel, 1975, Hutchinson, 1957 and 1984). Generally low DO concentration is associated contamination of water by organic matter and sometimes oxygen totally disappears from water giving rise to a condition of anoxia. Generally the changes in the DO concentration in different seasons is directly or indirectly governed by the changes or fluctuations of water temperature and BOD. Solubility of oxygen increases with the decreases of water temperature and solubility of oxygen is low at higher temperature. Quadri and Shah (1984) recorded DO pulses in the Winter and Summer. During the present investigation seasonal and spatial or sudden changes in DO concentration in pond -2, p -3 and pond -4 have been noticed. Winter pulses of DO concentration may be attributed

to the low water temperature facilitating increased solubility of oxygen. Thomas (1966) reported similar findings. The Summer, Monsoon and Post-monsoon pulses and spatial rises in DO concentration may be due to the photosynthetic release of oxygen by phytoplankton. Consequently, the abundance of phytoplankton during these seasonal in the study ponds was quite conspicuously high. Similar views were expressed by Ray et.al.(1966), Vass et.al.(1977), Rajesh (1981), Chakrabarty et.al. (1976), Pahwa and Mehrotra (1966), Agrawal et.al.(1976), Rai (1978), Rao (1979), Nicholas (1970), Patil (1982) and Hannan (1979). The present study ponds are occasionally agitated by drawing nets and other physical methods which may regarded as an important reason for moderately high to higher values of DO content. In some monthly samples, comparatively lower values of DO concentration have been recorded which may be due to high rate of consumption of DO for organic decomposition and also for high respiratory activities by the biota specially the phyto- and zooplankton. Consequently, the abundance of zooplankton was found to supersede the monthly and seasonal abundance of the phytoplankton in the study ponds. Similar views were also held by Hutchinson (1967), Moss (1972), Brezonic and Fox (1972), Morrissetle and Manivic (1978) and McColl (1952). In study pond -1 the DO values had a annual range from 4 - 7.6 mg/l with a percentage saturation of oxygen (P.C. Sat) value of 45 to 92, which indicates that the pond water always remained undersaturated, although the undersaturation did not come below 45%. This range of DO content and saturation values are indication of highly eutrophic nature of the study pond. This view is supported by many scientists including Lakshminarayana (1965), Jayangauder (1964), Munawar (1970), Zafar (1966), Pahwa and Mehrotra (1966) and Venkateswarlu (1969). The seasonal range of DO content in pond -1 were from 5.4 - 6.1, 5.2 - 6.8,

5.7 - 6.9 and 4 - 7.6 mg/l with saturation values of 70 - 82, 68 - 92, 69 - 92 and 45 - 86% in the Summer, Monsoon, Post-monsoon and Winter seasons respectively.

Monthly DO concentration in pond -2 showed a uphill and downhill values with peaks in August, September, October and April. Lower values of DO content was recorded in December. The annual range of DO values were found to be from 5 to 8.4 with a saturation value ranging from 53 - 114.4%, which indicate conditions of undersaturation and supersaturation in this study pond. According to Lakshminarayana (1965) a saturation value above 40% is indicative of nearly stressed condition but not harmful to the biota, while values above it reaching upto 100% saturation are indicative of eutrophic condition of water yielding high productivity. Supersaturation values were reached in April (Summer), August (Monsoon, 105%) and September and October (Postmonsoon, 113% and 104%). The seasonal values of DO content ranged from 6 -8.4 mg/l (83 - 114.4% saturation) in Summer, 6.1 - 7.6 mg/l (80 - 105%) saturation) in Monsoon, 6.8 - 8.4 mg/l (81 - 113% saturation) in Post-monsoon, and 5 - 7.9 mg/l (53.2 - 86.5% saturation) in the Winter. Lower values of DO concentration in the Winter might be due to high rate of decomposition of bottom organic matter associated with higher rate of utilization of oxygen for respiration of the biota. Consequently, the phytoplankton and zooplankton abundance were high in the Winter in comparison to the other seasonal (116341 unit/l and 442213 unit/l respectively). Ellis (1973) and Elmore (1961) opined that high concentration of suspended matter which imports turbidity in water is an important reason for low value of DO concentration. Consequently the lower value of DO was associated with high turbidity and high TDS value of water in Winter season in this study pond.

A regular pattern of seasonal fluctuation of the DO concentration was noticed in study pond -3. Monthly and seasonal values were conspecuously high with sudden fall in November when the DO content came down to 5.9 mg/l. The annual range of DO content was found from 5.7 to 9.6 mg/l and saturation value ranged from 69.2 - 126%, which clearly indicates that the DO values of water of this study pond -3, did not come below 5.7 mg/l with a percentage of saturation never fell below 69.2%. Supersaturation was found to reach in April (Summer, 114.2% saturation), May (Summer, 105.4% saturation), June (monsoon, 101.9% saturation), September (Post-monsoon, 126% saturation) and December (Winter, 101.2% saturation).

The seasonal range of DO values were 7.1 - 8.3 mg/l (93.4 - 114.2 % sat.) in Summer, 7 - 7.7 mg/l (95.5 - 101.9 % sat.) in Monsoon and 5.7 - 9.6 mg/l (70 - 126 % sat.) Post-monsoon, 5.9 -9.6 mg/l (69.2 - 101.2% sat.) in the Winter. This study pend is comparatively large and constant wind action producing surface waves might be one of the reasons for high value of oxygen in water. Moreover, the abundance of phytoplankton was also found to be comparatively high throughout the study year.

The DO concentration in pond -4 followed a uphill and downhill pattern throughout the year. Peaks of DO concentration were observed in September (10.6 mg/l and 140% sat.), October (7.9 mg/l and 101.2% sat.), February (8.2 mg/l and 100% sat.), March (7.2 mg/l and 101% sat.) and in April (8.2 mg/l and 113% sat.). The annual range of DO concentration in this study pond varied from 6.3 to 10.6 mg/l and the percentage saturation varied from 77.5 and 140%. The DO concentration level was always found to be below 6.5 mg/l did not rise up above 10.6 mg/l during the study months which indicate the eutrophic to

hypertrophic nature of the study pond. Moreover, similar reasons like pond-3 may have also prevailed here for a higher value of DO concentration. DO concentration reached supersaturation twice in Post-monsoon (140% in September and 101.2% in October), once in Winter (101% in March) and once in Summer (113% in April). Moreover, 100% sat. was observed in February. The Seasonal variation of DO concentration in this study pond were 7.1 - 8.2 mg/l with 97.4 - 113 % sat. values in the Summer; 6.3 - 6.7 mg/l with 83.7 - 91% sat. values in the Monsoon, 6.5 - 10.6 mg/l with 77.5 - 140% sat. values in the Post-monsoon and 7.9 - 8.2 mg/l with 88 - 100% sat. in the Winter. The Post-monsoon, Winter and Summer DO maxima with supersaturation were conspecuous, the maginitude of undersaturation was low during the study period.

Biochemical Oxygen Demand (BOD)

Biochemical oxygen demand (BOD) is the amount of oxygen required for the oxidation of organic matter by bacteria in the water body. It, therefore, indicates the amount of organic matter present in the water systems. BOD is of immense importance in the assessment of water quality. Monthly and seasonal values of BOD appears to be a function of changes in the magnitude of dilution, quantity of the organic matter and the activities of micro organisms which carry out decomposition of carbonaceous and nitrogenous matter. precisely the BOD values of a water body are indicative of the amount of biodegradable organic matter at a particular time as it varies with further addition of organic matter into water through various agencies. During the present investigation, the BOD₅ level in the study ponds were found to be of moderate values. In pond -1, the BOD values were found to have a short range of variation in the consecutive sampling

months. Seasonal values also varied with a short range. Compared to others study ponds the BOD5 level in this pond was found to be low and within the limit of acceptance in many occasions. The seasonal changes were found to range from 2.7 - 6.9 mg/l in the Summer, 5.4 -12 mg/l in the Monsoon, 3.1- 4.9 mg/l in the Post-monsoon and 3.2 - 7 mg/l in the Winter. The findings indicate that the pond water contained relatively small amount of biodegradable organic matter. The study pond -2 also was found to contain relatively small amount of biodegradable organic matter, and the seasonal values varied from 2.5 - 3.7 mg/l in Summer, 2.3 - 9.6 mg/l in Monsoon, 3.9 - 8.7 mg/l in Post-monsoon and 2.9 - 3.7 mg/l in the Winter. However, the highest BOD₅ value was noted in the Post-monsoon and Winter probably due to the accumulation of more organic matter that led to increased metabolic activities.. Monsoon and Winter rises of BOD value was also reported by Robert (1969), Richard (1966), Voznaya (1981) and Golterman (1983). Comparatively high BOD values were obtained from pond -3 with highest value in the Monsoon and Post-monsoon. The Summer values were relatively high but the Winter values were low. However, the BOD values show a moderately high level of organic matter present in this study pond. The seasonal variations of BOD values in this study pond were found to be from 9 - 11.6 mg/l in Summer, 9.4 - 13.6 mg/l in Monsoon, 6.7 -12.8 mg/l in Post-monsoon and 3.2 - 7.6 mg/l in the Winter. Study pond -4 showed comparatively higher amount of biodegradable organic matter throughout the study year. Higher values were noticed in the Monsoon and Postmonsoon indicating relatively large amount of biodegradable organic matter. However, the Summer and Winter values were also comparatively high. The seasonal values of BOD in the study pond were found to range from 8.6 - 13 mg/l in Summer, 13.2 - 16.8 mg/l in Monsoon, 12.4 -16.8 mg/l Post- monsoon

and 9.3 - 13.6 mg/l in the Winter. COD - BOD or P values were also calculated which were 4.4, 8.2, 4.6 and 7.4 in pond -1 in Summer, Monsoon, Postmonsoon and Winter respectively, while in pond - 2 the 'P' values were 4.3, 7.6, 4.4 and 5.4 in the four seasons respectively. In pond -3 the 'P' values were 11.2 in Summer, 14 in Monsoon, 14.4 in Post-monsoon and 8.4 in the Winter. In study pond - 4, The 'P' values were found to be 10.8 in Summer, 19.2 in Monsoon, 15.3 Post-monsoon and 11.2 in the Winter. This study pond appeared to have comparatively higher 'P' values than the other three study ponds. High BOD, COD, P^H and P values indicate higher amount of organic matter, (Goutam 1990). Study pond-4 is, therefore rated as having higher amount of organic matter amongst the four study ponds.

Chemical Oxygen Demand (COD):

The COD is an indication of total organic matter (including biodegradable and non-biodegradable) present in the water of ponds. It may be defined as the measure of oxygen which is required in oxidizing the organic compounds present in water by means of chemical reactions involving oxidizing substances. (Swarup et. al., 1992). An increase in the COD values indicate an increase in the biodegradable and non-biodegradable organic matters in water.

During the period of investigation, the study ponds indicated a comparative by lower load of non-biodegradable and biodegradable organic matters. In all monthly samplings in the study ponds, the COD value was within the permissible limit of tolerance (45 - 75 mg/l) set by ISI (Trivedy, 1990). The COD values of the study ponds were comparatively low and the 'P' values (COD-BOD) were also comparatively low.

The COD values in pond -1 were found to have a seasonal range from 7.5 - 15 mg/l in Summer, 11.9 - 22 mg/l in Monsoon, 7.2-10.6 mg / 1 in Postmonsoon and 9.8-15.2 mg/l in the winter with a 'P' value of 6.4, 8.2, 4.6 and 7.4 respectively. In pond -2, the COD values varied from 6.4 - 8.4 mg/l in Summer, 6.7 - 20.7 mg/l in Monsoon, 8.2 - 12.8 mg/l in Post-monsoon and 7.3 - 10.1 mg/l in the Winter and the 'P' values were 4.3, 7.6, 4.4 and 5.4 respectively. Study pond-3 registered a comparatively higher value of COD in comparison to study pond -1 and 2. The COD value in this pond were found to range from 19.6 - 26.3 mg/l in Summer, 22.3 - 28.7 mg/l in Monsoon, 14.5 - 27.7 mg/l in Post-monsoon and 8.7 - 18.9 in the Winter with the 'P' values of 11.2, 14, 11.4 and 8.4 in study seasons respectively. Study pond -4 had always highest COD value than the other three study ponds. The season range of COD values were 17.3 - 25.9 mg/l in Summer, 31.2 - 37.2 mg/l in Monsoon, 26.2 - 33.6 mg/l in Post-monsoon and 20.7 - 24.8 mg/l in the Winter with corresponding 'P' values were 10.8, 19.2, 15.3 and 11.2 in this study pond. Moderately high PH value with lower BOD, COD and 'P' values are indications of organic load within the limit of tolerance in the ponds. Similar views were also expressed by Gautom (1990), Trivedy (1990), Ghose and Sharma (1989), and Khanna (1993). Lakshminarayana (1965) and Jayanguder (1964) are of opinion that water bodies with these values of PH, BOD and COD with higher DO values are eutrophic in nature. The present study ponds, therefore, can be termed as eutrophic or highly productive in nature. Consequently the stock density of brood fishes of exotic origin was as high as 10 metric tones/ha./year (personal communication from the hatchery authorities). This productivity was rated as very high in the study ponds in terms of exotic varieties of brood and stock fishes. (Hatchery authorities).

Ammonia (NH₃ - N):

Ammonia is liberated in water as an end product of decomposition of organic matter and also as an excretory product of aquatic animals. Waste matter of domestic and industrial origin is rich in nitrogen organic matter which increases the ammonia load in water and raises the toxic level in some accasions (Swarup et. al. 1992). During the present investigation, the NH₃ - N level was found to be in a very low level and ranged from a zero value to 0.2 mg/l. The level of tolerance is known to be 0.05 mg/l for the existing biota (Ghose & Sharma, 1989). This indicates the NH₃ - N values exceeded the tolerance limit in many occasions in the study ponds. Any clear or definite pattern of fluctuation of NH₃ - N values have not been observed in the study ponds. However, the regular charging of the ponds with organic matter as fish feed may be the probable reason such rises in the NH₃ - N values. The Winter and Monsoon values were comparatively high in the four ponds. Higher Winter, Summer and Monsoon values were also reported by Trivedy (1990) and Ghose and Sharma (1985) Ghose and Sharma (1989) reported an average value of 0.11 mg/l in less polluted waters. The present ponds may also, therefore be class as less polluted, although in some cases the limit of tolerance have been surpassed in the ponds. The seasonal values of NH3 - N in pond -1 were found to range from 00 - 0,10 mg/l in Summer, 0 - 0.10 mg/l in Monsoon, 0.10 - 0.11 mg/l in Postmonsoon and 0.10 - 0.13 mg/l in the Winter, while in pond -2 the values varied from 0 - 0.10 mg/l, 0 - 0.10 mg/l, 0.1- 0.2 mg/l and 0.03 - 0.10 mg/l in four seasons respectively. In study pond -3, the NH₃ - N values ranged from 0 - 0.10 mg/l, 0 - 0.10 mg/l, 0.10 - 0.11 mg/l and 0.10 - 0.13 mg/l in Summer, Monsoon, Post-monsoon and Winter respectively. Similarly the values varied from 0 - 0.10 mg/l, 0 - 0.10 mg/l, 0.02 - 0.2 mg/l and 0.10 - 0.2 mg/l in study pond -4 respectively.

Phosphate (PO₄) content

The phosphate in one of the major nutrients essentially required by plant. It is an essential component of life. Phosphate is used by phytoplankton and stimulates plant growth when in excess of critical concentration. In waters PO₄ occurs as complex or inorganic water-soluble form and also in organic bound form. During the present investigation the inorganic PO4 content was found to vary from pond to pond in different seasons. The PO₄ values ranged from 0.01 to 1 mg/l in the study ponds. The average value of phosphate in pond -1 reveal that the level of phosphate was high in the Winter (0.7 mg/l) and Summer (0.7 mg/l). Similar values were reported by Gautom (1990) in his investigations. The seasonal values ranged from 0.1 - 0.7 mg/l in Summer, 0.1 - 0.2 mg/l in Monsoon and Post-monsoon and 0.1 - 0.7 mg/l in the Winter. In pond -2, the PO₄ values ranged from 0 - 0.1, 0 - 0.1, 0.1 - 0.2 and 0.03 - 0.1 mg/l in Summer, Monsoon, Post-monsoon and Winter respectively, while in study pond -3, the seasonal PO₄ values were found to range from 0.03 - 0.1, 0.03 - 0.3, 0.1 - 0.2 and 0.01 - 0.3 mg/l respectively. In study pond -4, Summer and Monsoon values were comparatively high then the other season. The values ranged from 0.02 - 0.4 mg/l in Summer, 0.3 - 1.0 mg/l in Monsoon, 0.02 - 0.2 mg/l in Postmonsoon and 0.1 - 0.2 mg/l in the Winter. Phosphate level of extent present in the study ponds indicate their high trophic nature. Lakshminarayana (1965) and Jayangader (1964) eprssed similar views and opined that PO₄ is of immense need forming the nutrients of Biological importance.

Oxidation-reduction potential and oxidation reduction index (Eh and rH_{2})

In a body of water or aquatic ecosystem where biological metabolism is going on, there is a continual change in the ration between the materials in reduced form and the materials in oxidized form. If the watery habitat that possesses organic material the concentration of reduced form is higher which results in lower values of Eh and rH₂. But after a short time when the material degrades, the water body starts to attain it original form and the Eh and rH₂ values increases. In case of continuos addition of the organic material, a continuous decrease in the Eh and rH₂ values occur (Voznaya, 1981; Gautom, 1990 and Warner, 1966). Lower values of Eh and rH₂ indicate the presence of higher amount of organic matter. The decomposition of the organic matter complex redox reactions resulting in the heterotrophic activities i.e. respiration succeeded by NO₃ reduction, fermentation, SO₄ reduction, CO₂ reduction etc. accompanied by microbial succession occur, lower as the Eh & rH₂ level. The neutral point of rH₂ is 28 (Voznaya, 1981).

In the present study ponds, the Eh values were found to be lower throughout the study period with lower ranges of seasonal variations. In pond -1, the seasonal Eh values were found to be almost similar and ranged from 0.25 to 0.35 mv. Similarly in pond -2, the values varies from 0.22 to 0.35 mv. In pond -3, the Eh values were found to range from 0.23 to 0.31 mv. The Eh values were lowest in the Summer and Winter indicating the organic load in study pond. Similar Summer and Winter low values were obtained by Voznaya (1981). In study pond -4 the Eh values varied annually from 0.23 to 0.32 mv., while the Summer and Winter values were comparatively lower than the Monsoon and Post-monsoon.

The rH₂ values in pond -1, were found to have an annual variation from 25.8 to 27.8 indicating lower load of organic matter. The seasonal values were found to be fairly uniform. The seasonal values were 26.2 - 27.8 in Summer, 26 - 26.6 in Monsoon, 25.8 - 26.3 in Post-monsoon and 26.1 in the Winter. Similarly in pond -2, the rH₂ values were also found to be uniform each month and season as well. The variation of rH2 values ranged from 25.3 - 27.6, 26.4 -27.09, 26 - 26.4 an 25.7 -26.02 in the Summer, Monsoon, Post-monsoon and Winter respectively. The values were slightly below the neutral indicating comparatively small amount of organic matter (Voznaya, 1981). Uniformly higher rH₂ values were also recorded from study pond -3 and the range of variation were from 25.7 - 27, 25.7 - 26.4, 26.1 - 26.2 and 25 - 26.1 in the Summer, Monsoon, Post-monsoon and Winter seasons respectively, indicating relatively lower load of organic matter in the pond water. In study pond -4, lower values of rH₂ in the Summer indicated higher load on organic matter. Winter values were comparatively low. The seasonal variation of rH₂ values were found to range from 15.8 - 26.4 in the Summer, 26.3 - 27.2 in Monsoon, 26.02 -26.3 in Post-monsoon and 25.7 - 26.1 in the Winter period.

Low Eh, rH₂ and low BOD, COD, P^H and P values indicate higher amount of organic matter in an aquatic body (Gautom, 1990). In the present study, moderately low values of Eh and rH₂ and moderately high value of P^H, P, BOD, COD and DO are indication of presence of organic matter in the four study ponds. Coupled with the NH₃ - N and PO₄ values the whole physico-chemical variables together formed a factor complex of nutrient status of biological importance imparting a eutrophic to hypertrophic nature to the study ponds.

ZOOPLANKTION

A Total of 38 genera of zooplankton were recorded from the four study ponds during the study period, of which 10 genera belonged to Copepoda, 9 to Cladocera and 19 to Rotifera. Pond-1, 2 and 3 were characterized by the presence of the Copepods, Cladoceran and the Rotifers. But the rotifers were found to occur in pond-4 only. The other two groups were not encountered in this pond during the study period. All the Zooplankton members occurred sporadically in the study ponds and a definite pattern of succession could not be noticed. The feeding habit of the exotic brood fishes may be the reason for such appearance of the zooplankton member. The occurrence of different genera of zooplankton have been depicted in Table - 1, 2, 3 & 4.

A total of 27 genera of zooplankton were recorded from pond-1, of which 8 genera belonged to Copepoda, 7 to Cladophera and 12 to Rotifera. Pond -2 was represented by 31 genera of which 10 belonged to copepoda, 8 to cladocera and 13 to rotifera. In pond -3 only 18 genera of zooplankton were noticed of which 4 belonged to the copepoda, 6 to cladocera and 8 to rotifera. Study pond -4 was represented by 12 genera of rotifers only. The reason for the absence of copepods and cladocerans could not be ascertained. It may so happen that the feeding habit of the brood fish silver carp; similar views were expressed by Kumer, 1992 & Biswas, 1996 and the water chemistry played a role for such disappearance. Higher values of turbidity, TDS, conductivity, HCO₃ alkalinity, total hardness and super saturation of oxygen throughout the study period may probably have created a factor-complex uncogenial for the members of the copepods and cladocerans (Jayangauder, 1964).

The seasonal abundance values in pond -1 indicate that the copepods were in their peaks in the Monsoon, while cladocerans showed their peaks in Summer and Monsoon. The rotifers were in their peak in the Winter and Maintained almost uniform density in the other seasons. In study pond -2, the copepoda had a peak in Summer, while the rotifers had two peaks - one in monsoon and the other in Winter. Similar Summer and Winter peaks were reported by Jayangauder (1964) and Lakshminarayana (1965). The cladocerans showed a uniform population density throughout the study period in this study pond -2. In study pond -3, the copepods had a high peak in the Post-monsoon, while the cladocerans had two peaks, one in Summer and one in Post-monsoon. The cladocerans density was almost uniform in the Monsoon and Winter. The rotifers were in high peaks in the Summer and Monsoon, with a comparatively high population density in the Winter & Post-monsoon. Pond -4 was represented by the rotifers only where Brachionus was found to be the dominant genus followed by Harringia, Trichocera Philodina and others. A very low population density was exhibited by *Platyias* and *Notholca*. It is noteworthy that all the 12 genera of rotifera were found to occur in all monthly samples in pond -4, but they were only of sporadic appearance in other study ponds. The rotifers, are therefore tolerant to varying degree of physico-chemical and biological condition. Arora (1966) observed that the rotifers occur in eutrophic waters in high abundance. Summer and Winter pulses of rotifers were reported by George (1966). Arora (1966), Thunmark (1945), Lilieroth (1950) and Berzins (1949) designated a large number of rotifer genera including Brachionus, Keratella and Filinia as indicators of eutrophic or polluted waters. Considering the occurrence of large number of genera of rotifera including Brachionus, Keratella and Filinia, in the four study ponds in varying density throughout the study period,

the study ponds can be indicated as eutrophic in nature. The study ponds may be marked as having lower level of pollution based on the values of the redox characteristics including the P^H, DO, BOD, COD, Eh and rH₂ and also the higher conductivity and chloride values (Gautam, 1990).

The seasonal abundance and percentage composition of phyto- and zooplankton in the four study ponds were depicted in Table -7. The data show that the zooplankton population density was always higher than that of the phytoplankton in all study seasons in the four ponds. In pond -1, the zooplankton constituted of 58.2 %, 58.3 %, 82.6 % and 90.5 % of the total plankton population in Summer, Monsoon, Post-monsoon and Winter respectively. In pond -2, the zooplankton density was 77.7 %, 73.2 %, 64.5 % and 79.2 % of the total plankton population in the respective seasons. In study pond -3, the zooplankton were represented by 67.2 %, 74.1 %, 56.4 % and 62.1 % of the total plankton population in Summer, Monsoon, Post-monsoon and Winter seasons respectively. In study pond -4, where only the rotifers occurred, the zooplankton density was 83.3 %, 61.2 %, 73.2 % and 44.6 % of the total plankton population in the respective seasons. It appears that the phytplankton density superceded the zooplankton density in the Winter in pond -4 only. The phytplankton and zooplankton seasonal ratio in pond -1 were 1:1.39, 1:1.39, 1:4.7 and 1:10 in Summer, Monsoon, Post-monsoon and Winter respectively while in pond -2, the ratio in the respective seasons were 1:3.5, 1:2.7, 1:1.8 and 1:3.8. In pond -3 the ratio in respective seasons were 1:3.7, 1:1.68, 1: 2.2 and I: 1.28, while in study pond -4 they were I: 2.7, I: 2.6, I: 1.6 and I: 0.8.

Primary Productivity.

The dark and light bottle method of estimation of GPP and NPP (Gross primary productivity and Net primary productivity) showed that all the study ponds were moderately to highly productive varying in different seasons. In pond -1, the GPP and NPP values were high in Summer, whereas in P-2 the GPP and NPP values were high (mg C/l/h) in all the four seasons and the negative NPP values, indicated a high rate of respiration by the fishes, zooplankton and other biota. In pond -3, the GPP values were moderately high in the four study seasons and negative NPP values were noted in the Monsoon and Winter only. Pond -4 presented higher GPP values in Monsoon and Post-monsoon with moderately high values in Summer and Winter. Negative NPP values were noted once in the Summer and another in the Monsoon.

PHYTOPLANKTON

Biological parameters have enormous importance on ecological point of view. All forms of natural and artificial water bodies contain variety of organism, both plants and animals as their natural flora and fauna. Study of these biota and their interrelationship with the existing physico-chemical variables is necessary to properly evaluate the trophic status and productivity levels for suitable management. The level of pollution and response of the biota to such condition can also be assessed for necessary abetement of the polluted condition. The present investigation was undertaken to see the effect of present cultural practices on the phytoplankton and zooplankton organisms in the culture ponds. Consequently, the monthly and seasonal occurrence, abundance and percentage composition of the available phyto and zooplankton members have been studied. (Table P -1, 2, 3, 4, ZP -1a, ZP -2a, 3a, 3b, 5, 5b, P -1a, 2a, 3a, 4a, P -1b, 2b, 3b & 4b).

Noteworthy feature of the study ponds is the occurrence of the major group of the algae throughout the year. Some planktonic algal genera were found to occur all the year round, while there were some occurring only sporadically in some sampling months. No algal group had been found to appear as a dense visual bloom during the period of investigation, though some of the phytoplankton members were found to be in high abundance in different months and seasons as well. However, the study pond water remained slightly chemical undersaturated with DO during the study year, with exceptions in a few occasions when the water was found to be over saturated with oxygen. Anoxiating condition have were been found to occur in any study pond. The nutritive substances of biological importance i.e. Moderately high nitrogen indicated by the organic load of the ponds, phosphate content were notably high. The Monsoon and Post-monsoon periods with comparatively high water temperature, high limit of visibility, high PH, DO, total alkalinity, total nitrogen, PO₄ might have provided favourable conditions for the growth of some phytoplankton member occurring in high abundance. It is noteworthy that the members of Chlorophyceae, Cyanophyceae, Bacillariophyceae and the Euglenophytes were in their peaks in different seasons and simultaneous peaks were observed in fewer occasions. The occurrence of a fewer number genera of indicates that the study ponds suffers from not having a wide variety of planktonic algae. As far as the monthly and seasonal abundance of phytoplankton is concerned, the study ponds herbour a quantitatively lower population abundance compared to a tropical or temperated water body. Given all the physico-chemical condition to a congenial level, the phytoplankton with respect to wider variety of members and monthly and seasonal abundance and percentage composition. But the present study ponds showed a poor variety of

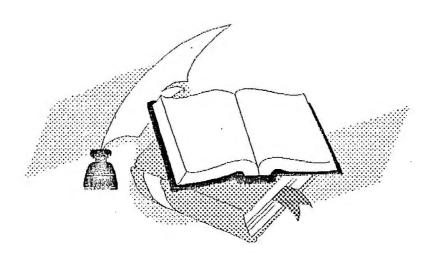
algal plankton and their abundance. It may be mentioned that the study ponds were used as stocking ponds for the exotic brood fishes with very high stock density. As a matter of fact, the exotic brood fishes including silvers and other carps are voracious phytoplankton feeders, and normally they consume much higher amount of plankton food than the indigenous variety of fishes. This type of feeding habit may have probably played in the study ponds for lowering the population density of the phytoplankton members. Moreover, the abundance of zooplankton population was always found to be high in all the ponds in corresponding seasons. Most of zooplanktons belonging to the copepoda, Cladecera and Rotifera are phytoplankton feeders. Their feeding pressure may also be regarded as a probable reason for such lower phytoplankton abundance in the study ponds. George (1966) reported similar situation in his study in India. Nauman (1927), Rao (1955), Yoshimura (1932), Zafar (1964), George (1966), Moyle (1946), Lakshminarayana (1965) and many other workers stressed that a body of water is eutrophic or highly productive characterized by it rich nitrogen, PO₄, total alkalinity, total hardness and DO contents. The present study ponds, thus, may be classed as eutrophic with respect to their content of the above parameters. Therefore, the lower population abundance and ununiform density of the phytoplankton may be due to the feeding pressure of the exotic brood fishes and other animal organism present in the study ponds. As has been stated in the observation chapter, the total number of genera of phytoplankton in pond -1 were 33, of which 12 belonged to the Chlorophyceae, 7 to Cyanophyceae, 10 to Bacillariophyceae, 3 to Euglenophyceac and I to Dinophyceae. As for the seasonal abundance of different group of phytoplankton algae in pond -1, the Summer, Monsoon, Post-monsoon and Winter abundance were 66044 unit/l (27.2 % Chlorophyceae, 24.8 %

Cyanophyceae, 14.8 % Bacillariophyceae, 34.5 % Euglenophyceae and 1.6 % Dinophyceac), 82225 unit/l (25.2 %, 19.8 %, 10.6 %, 43.2 % and 1.1 % Chlorophyceae, Cyanophyceae, Bacillariophyceae, Euglenophyceae Dinophyceae respectively), 66123 unit/l (22.8 %, 18.3 %, 21.8 %, 36.7 % and 0.12 % Chlorophyceae, Cyanophyceae, Bacillariophyceae, Euglenophyceae and Dinophyceae respectively) and 57161 unit/l (22.9 %, 31.5 %, 16.6 %, 28.7 % and 0.22 % Chlorophyceae, Cyanophyceae, Bacillariophyceae, Euglenophyceae and Dinophyceae respectively) in this study pond respectively. It appears that the Euglenoids were numerically dominant followed by Chlorophyceae, Cyanophyceae, Bacillariophyceae, Euglenophyceae and the Dinophyceae. Similarly in pond -2, 26 genera of phytplankton were recorded in the study period of which 9 genera belonged to the Chlorophyceae, 7 genera to Cyanophyceae, 6 to Bacillariophyceae, 3 to Euglenophyceae and 1 Dinophyceae. As regards to the seasonal abundance and percentage composition, the Chlorophyceae constituted of 30.2 %, 28.5 % 28.9 % of the total phytoplankton population in Summer, Monsoon, Post-monsoon and Winter respectively, while the Cyanophytes constituted of 22.2 %, 26.4 %, 11.2 % and 23 8 % of the total phytoplankton in the four consecutive seasons respectively. the Bacillariophytes constituted of 16.2 %, 19.7 %, 34.2 % and 24.4 % of the total phytoplankton in the Summer, Monsoon, Post-monsoon and Winter respectively, while the Euglenoids comprised of 31.4 %, 25 %, 23.7 % and 6.2 % of the total plankton in the consecutive seasons respectively. The Dinophytes were found to occur in the Post-monsoon (1.8 %) and Winter (9.5 %). In study pond -3, a total of 30 genera of phytoplankton occurred during the period of investigation of which 11 genera belonged to Chlorophyceae, 7 genera to Cyanophyceae, 10 to Bacillariophyceae, 2 to Euglenophyceae and no member of

Dinophyceae have been observed in this pond during the sampling period. The observation in the seasonal abundance shows that the Chlorophyceae constituted of 17.2 %, 24.6 %, 7.8 % and 6.8 % of the total phytoplankton population in the Summer, Monsoon, Post-monsoon and Winter seasons respectively. Similarly, the Cyanophytes constituted of 40.5 %, 46.1 %, 38.9 % and 42.9 % of the total phytoplankton in the respective season as well. The Cyanophytes were found to dominant group on annual and seasonal basis followed by the ad Euglenophytes. The Bacillariophytes Bacillariophytes, Chlorophytes constituted of 21.4 %, 28.1 %, 25.7 % and 44.5 % of the total population of in the respective seasons. In the study pond -4, 11 members of Chlorophyceae, 8 members of Cyanophyceae, 9 members of Bacillariophyceae and 2 members of Euglenophyceae were recorded during the study period. As regards the seasonal and percentage comparison is concerned, the Chlorophytes constituted of 11.5 %, 9.7 %, 23.6 % and 19.8 % of the total phytoplankton population in the Summer, Monsoon, Post-monsoon and Winter seasons respectively, while in the respective seasons, the Cyanophytes comprised of 45.2 %, 40.9 %, 37.4 % and 26.2 % of the total population of phytoplankton. The Bacillariophytes were found to constitute 15.5 %, 7.3 %, 20 % and 41.5 % in the total respective seasons, while the Euglenophytes constituted of 27.5 %, 37.2 %, 15 % and 12.5 % of the total phytoplankton population in the respective four seasons. The seasonally dominant Cyanophytes were found to be followed Euglenophyceae, Bacillariophyceae, and Chlorophyceae. The monthly and seasonal abundance of different group and genera of phytoplankton have been stated in the observation chapter with a number of tables and charts, which indicate that a definite pattern of seasonal succession is not clearly observed in the study ponds.

CHAPTER -5

REFERENCES



- Abbott. W. 1957 unusal phosphorus sources for plankton Algac. Ecol. 38 (1) 52
- Adebisi. A. A. 1980. The physico-chemical Hydrology of a tropical seasonal upper Oyun river, Hydrob. 79: 159-165.
- Adoni, A.D. 1985. Work Book on Limnology. Dept. Bot. Dr. Harisingh Gour Vishwavi. Sagar, India.
- Agrawal, D. K.; Gaur, S. D.; Tiwar, I. C and S. Narayan 1976, Physico-Chemical Characteristic of Ganges of water of Varanasi, Indi. J. Envir Hith 18: 201-206.
- Ahlastrom, E. 1940. A revision of the rotarian genera Brachionus and Platyias with description of one new species and two new varieties. Bull, Amer. Nat. Hist. 77: 143-184.
- Ajmal, M. and Raziuddin. 1988. Studies on the pollution if Hindon river and Kali nadi (India) Ashish Pub. House. New Delhi India.
- Alam, A. K. M. N, Islam, M. A, Mollah, M. F. A, and M. S. Haq. 1987. Status of phytoplankton in newly constructed ponds and their relation to some metrological and limnological factors. Bang. J. Fish. 10(1): 75-81.
- Ali, S and T. Chakrabarty 1992. A book on fresh water Invertebrates of Bangladesh, Bang Acad P: 1-207.
- Ali, S. Rahman, A. K. A, Patwary, A. K. and R. K. Islam 1982. studies on the diurnal Variations in Physico-chemical factors and Zooplankton in a fresh water pond. Beng. J. Fish. 2-5(1-2): 15-23.
- Ali, S; Chowdhury, A. and A. R. Roy 1998. Ecology and seasonal abundance of zooplankton in a pond in Tongi, Dhaka. Bang. J. Zool. 8(1): 41-49.

- Allanson, B. R. and J. M. T. M. Gieskes 1961. An Introduction to limnology of Hartbeespoort dan, with speial reference to the effect of industrial and domestic pollution Hydrob, 18:77-94.
- Ameen, M. Begum, Z. N. T. Ali. S. Rahman, M. M. and T. K. Roy. 1986. A Comparative limnological study of two fish ponds Raipur. D. Uni, Stud. Part E 1(1): 25-34.
- Ameen, M., Begum, Z. N. T., Mustafa, A. and .S. Ali 1988. Seasonal and dial profile of temperature, light penetration, dissolved oxygen and free carbondioxide of a fish pond from south of Bangladesh. J. Zool, 3:1-8.
- Ameen. M, Begum. Z. N. T, Rahman, P. M. M. and S. A. Haidar. 1987 Effect of fertilizer on plankton, the natural food for fish in ponds. J. Zool. 2: 19-33.
- Andrewartha, H. G. and L. C. Birch. 1954. The distribution and abundance of animals. Chicago. Uni. press Chicago. P:1-782
- APHA 1976 standard methods for the examination of water and waste water.

 American public Health Association Washington.
- APHA 1989 standard methods for the examination of water and waste water (Latest). American public Health Association, Washington
- APHA.1946 Standard method for the examination of water and waste water.

 American public Health Association Washington, 14th. edi. p:
- Arora, C. H. 1961. Rotifers as Indicators of pollution, CPHRI Bull. 3, 4:24. 1966. Responses of Rotifera to variation in some ecological factors, por Ind. Acad. Sci. 63:57-66.

- Arora, CH. 1966. Rotifera as indicators of trophic Nature of Environments Hydrob. XXXII, Phac. 1-2: 146-159.
- Atkins, W. R. G. 1926. A quantitative consideration of some factors. Concerned in plant growth in water .11. some Chemical factors, J. du, Counc-1.3: 197-226.
- Badala, S. P and H.R. Singh. 1981. Fish and fisheries of River Alaknanda. Proc. Nat. Acad. Sci., 15(B): 133-142.
- Banu, A. N. H., Ali M, M. and A. Islam 1987 plankton study of some selected ponds in different locations of Bangladesh. Bang. J. Aqua. 9 (2):55-59.
- Begum A. Mostafa G. Ali, S. and K. Ahmed 1989. studies on limnology in a minipond and growth of Tilapia nilotica. Beng J. Zool. 17(1): 35-45.
- Begum, A. 1958 . A short note on plankton of fresh water pond of Dacca. Agri. Pak. 9: 370-392.
- Begum. Z. N. T. and M. J. Alam 1987 plankton abundance in relation to physico-chemical in two ponds in Maijdee court, Mohakhali, J. Asi. soc. Bang. (sc)3:35-45.
- Benergee, J. C. 1936 studies on the Myxophyceac of lower Bengal, J. Ind. Bot. soc. XV(5): 285-309.
- Benoit, R. J. 1955. Relation of phosphorous content to algal bloom sewage and Inland waters. 27: 267-269.
- Berg, K. 1927 A faunistic and biological Study of the Danish Cladocera. Vidensk. Medd. Dansk. Daturh Foren 88:31-111.
- Berjins, B. 1949. Zur. lomnologe der seen Sudostlentdlend. schweig. 2. Hydrob. 11:583-607.

- Berner, L.M. 1951. Limnology of the lower Missouri River, Ecol. 32:1-12.
- Bhantager, G. P. and G. P. Sharma 1973. physico-chemical features of sewage polluted lower lake Bhopal "Environ-agents and their Biological effect." proc.intern.symp. Int. 5-4: 212-223
- Bhouyain and Asmat 1992. Fresh water Zooplankton from Bangladesh. Ghazipur. pub.
- Bhouyain, A. M and D. K. Das 1985. Primary production of a fish pond. uni, J. Zool. Raj. uni. 4:55-58.
- Bhouyain, A. M. 1983 Fresh water and Brachish water pollution in Bangladesh F. I. B. 3:32 PP
- Bhouyain, A. M., Hafizuddin, A.K.M. and M.K. Pasha 1981. Diurnal movement of plankton in a pond. Chitta Uni. Stud.(P-2), 5:41-49.
- Biswas, A. 1993. Limnology of three Fish ponds in Rajshahi hatchery, M. Sc. thesis(unpub). P:
- Biswas, K. P. 1966. A Text Book of Fish, Fisheries & Technology, Narendra pub. House, Delhi, India, 2nd. edi, P: 1-578.
- Blum, J. L. 1957 An ecological study of the alga of saline river near Michigan, Hydrob. 9:301-408.
- Braj Nandan, Prasad, Y.C. Jatly and Singh. 1985. Periodicity and interrelationship of physico-chemical factors in ponds. Proc. Nat. Symp. pure & Appl. Limnol. Bull. Bot. Soc. Sagar. 32;1-11.
- Brezonic, P.L. and J.L. Fox. 1974. The Limnology of slected Guatemalan Lakes. Hydrob. 45:467-487.
- Brige, E.A.and C. Judy: 1971. The Inland lakes of Wisconsion. The Dissolved gass of water and their biological significance. Wis. Geal. & Nat. Hist. Surv.

- Butcher R. W. 1946 studies on the ecology of the river. 1. The algal growth in highly Calcareous streams. J. Ecol. 33:268-283.
- Carpenter J. H. 1966 New measurements of Oxygen Solubility in pure and natural water. Limnol. oceangr, 11:264-277.
- Chakrabarty, A. H. 1995. Studies on the Physico-chemical and biological condition of a canal receving indutrial efflents and their effect on the river Padma in Rajshahi: M.Sc. Thesis, Pub: Bang. Bot. soc. 8th. Bien. Bot. Cof. Dec. 12-13, 1994.
- Chakraborty, R. D.; Roy, P. and S.B. Sinha. 1966. A Quantitative study of the plankton and the physico-chemical conditions of the river Jamuna at Allahabad in 1954-55. Ind. J. Fish. 6(1): 186-203.
- Chandler, D. C. 1940. Limnological Studies of Wester Lake Eric. I Plankton and Certain Physico-chemical data of the Bass Island region, from Sept.-1938 to Nov.- 1939. Ohio. J. Sci. 40: 291-336.
- Chandler, D. C. 1942. Limnological Studies of Western Lake, Eric III Phytoplankton and Physico-chemical data from Nov.-1939 to 1940 Ohio J. Sci., 42: 24-44.
- Chandler, D. C. and O. B. Weeks. 1945. Limnological Studies of Western Lake, Eris. V. Relation of Limnological and Metrological conditions to the production of Phytoplankton in 1992. Ecol. Monogr. 15: 435-456.
- Chowdhury, C. S. and A. Khair. 1982. Phytoplankton members of Kaptai lake. Chitta. Hill. Tr. I. Desmidiaceae. Chitta. Uni. Stud. 6(II):129-136.

- Chowdury, A. N., Begum, S. and N. Sultana. 1989. Occurrence of Seasonal Variation of Zooplankton in fish Pond in relation to some Physicochemical factors. Bang. J. Zool. 17(2): 101-106.
- Chowdury, S. H. and A. Mazumder. 1981. Limnology of Kaptai, I. Physicochemical features. Bang. J. Zool. 9(1): 59-72.
- Das S. M. and Akhtar, S. 1970. A report on fresh Water Plankton from the Dal Lake, Kashmir, Kashmir Sc. 7(1-2): 133-137.
- Das, N. G. and A. L. Bhuiyan. 1974. Limnoplankton of some inland water of Dhaka City. Bang. J. Zool. 2(1): 27-42.
- Das, S. M. and V. K. Srivastava. 1956. Quantitative studies on fresh water Plankton, of fish tank in Lucknow, India Proc. Nat. Acad. Sci. India. B 26: 85-92.
- Das, S.M. and J. Pandey. 1981. Some Physico-chemical and biological indicators of pollution in Lake Nainital, Kumaon(UP). Ind. J. Ecol. 5:7-16.
- David, A. 1956. Studies on the Pollution of the Bhadra river at Bhadrawati fisheries effluents. Pro. Nat. Inst. Set. India, 93(3): 132-160.
- Desikachary, T. V. 1959. Cyanophyta. I. C. A. R. New Delhi, P:
- Doyle, C.B. 1966. Effectiveness of high p^H for the destruction if Pathogens in raw sludge filler cake. J. of WPCF. 39:1043.
- Edmondson, W. T. 1955. Factors affecting productivity in fertilized salt water. Deep. Sea. Res. 3:451-464.
- Edmondson, W. T. 1972. Nutrients and Phytoplankton in lake Washington, Limno. Oceangr. Spe. Symp. P(1): 172-193.

- Ellis, M. M. 1937. Detection and measurement of stream pollution, U. S. Fish Bull. No. 48.
- Elmore, H.L. and W.P. West. 1961. Effect of water temperature on stream reaeration. IASCF 87, SA. 6: 59 New Delhi. India.
- Emerson, S.; Broecker, W. S. and D. W. Schindler. 1973. Gas exchanges rates in a small lake as determined by the radex method. J. Fish. Res.
- Ernst, D. and D. Reinhardt. 1980. Primary productivity measurement and carbon metabolism in steinhuder macr lake Dummor, Develop. Hydrobiol.Jewson, D. and Junk. W. Pub: the Haque. P(3):
- Fakruzzaman, M. 1997. Studies on the helioplankton of lower Barial region in relation to physico-chemical conditions. Unpb. M. Sc. Thesis Dept. of Bot. Raj. Uni. Bangladesh, P: 1-229.
- Fao. 1976. Ref. 1984. Physical and Chemical methods of soil and water analysis. No. 10.
- Fritch, F. E. 1931 some aspects oc the ecology of fresh water algae (with special reference to static water). J. Ecol., 19: 233-272.
- Fritsch, F. E. 1970. A general consideration of subaerial and fresh water algae flora of ceylon. A contribution to the study of tropical algal ecology. Part I. Subaerial algae and algae of the inland fresh waters. proc. Roy. SOC. (B. Series) 79: 23-272.
- Gaarder, T. and H. H. Gran. 1927. Investigation of the production of plankton in Oslo Fjord Rapp. Proc. verb. Cons. Perm. Int. Explor. Mer. P: 421-448.

- Ganapati, S. V. 1940. The ecology of temple tank. Containing a permanent bloom of microcystis aeruginosa (kutz) Hefner. J. Bombay, Nat. Hist. SOC. 42(1): 65-77.
- Ganapati, S. V. 1957. Limonological studies of the upland water in Madras state. Arch. f. Hydrb. B. 53: 30-61.
- Ganapati, S. V. 1960. Ecology of the tropical waters. Proc. Symp. Algology. ICAR. New Delhi.
- Ganapati, S.V. 1943. An Ecological study of garden pond containing abundant zooplankton Proc. Ind. Acad. Sci. 17:41-54.
- Gauthier, L.L. 1963. Oedogoniaceae, Africaines, Nova Hed. VI & VII, P: 1-588.
- George, M.G. 1966. Comparative plankton Ecology of five fish Tanks in Delhi, India. Hydrob. 1-2:81-108.
- George, M.G. 1966. Diurnal variation in physico-chemical factors and zooplankton in the surface layers of three fresh water ponds. Ind. J. Fish. 13:48-82.
- Gerog, M. G. 1966. Comparative plankton ecology of five tanks in Delhi, India, Hydrob. 27: 81-108.
- Ghose, N.C. and C.B. Sharma. 1989. pollution of Ganga river. Ashish pub.House. New Delhi. India.pp.240.
- Goldman, C. B. and R. G. Wetzel. 1963. A Study of primary productivity of Clear lake, California, Ecol. 44: 283-294.
- Golterman, H. L. 1975. Chemistry in "River Ecology" Califo. Uni. Press. P: 53 54.
- Golterman, H. L. 1975. Physiological Limnology, Els. Sci. Pub. Co. N. Y. P: 1-489.
- Golterman, H.L. R.G. Sly. and R.L.Thomas. 1983. Study of the relationship between water quality and sediment. UNESCO.Pub.

- Gonzalves, E. A. and D. B. Joshi. 1946. Fresh water algae near Bombay, J. Bombay, Nat. Hist. Soc. 46(1): 154-174
- Gonzalves, E.A. and D.B. Joshi. 1946. Fresh water algae near Bombay. 1. Seasonal succession of the algae in the tank at Bandra. J. Bom. Nat. Hist. Phi. Soc. ,46:154-176.
- Goutom, A. 1990. Ecology and pollution of mountain water (A case study of Bhagirathi River). Ashis Pub. House, 8/81, Punjabi Bagh, New Delhi, Indial, P: 1-209.
- Goutom, A. 1992. Aquatic Environment, Ashis, Pub. House, 8/81, Punjabi Bagh, New Delhi, India, P: 1-209.
- Green, J. 1956. Growth, size and reproduction in Daphnia. Proc. Zool. Soc. Land. 135: 491-523.
- Gunther, R. E. R. 1936. A report on the Oceanographical investigations in the Peru Coastal Current. Disco. report. 13: 107-276.
- Habib, M. A. B. and M. Badruddin. 1983. Estimation of some available nutrient in six pond bottom and adjacent land soil of the village singrol in Barind tract. Bang. J. Fish. 6(1-2): 49-52.
- Habib, M. A. B.; Ahmed, M.; Islam, A. and A. K. M. A. Haque. 1983. A comparative study on some chemical characteristics of pond bottom and their adjacent land soils of fourteen soil types of Bangladesh. Bang. J. Fish. 6(1-2): 19-26.
- Habib, M. A. B.; Islam, M. A.; Mohsinuzzaman, M. and M.S. Rahman. 1984. Effect of some physico-chemical factors of water on the abundance of fluctuation of Zooplankton of two selected Ponds. Uni. J. Zool. Raj. 3:27-34.

- Habib, M. A. B.; M. S. Rahman.; Islam, M. A. and M. Badruddin. 1984. A comparative study on some chemical characteristics of pond bottom and their adjacent land soils of fourteen soil types of Bangladesh. Bang. J. Fish. 7(1-2): 61-70.
- Hannan, H. 1979. Chemical modification in reservoir regulated streams. In the ecology of regulated streams. J. W. Ward and J. A. Stanford edi. Plenum Corp. Pub. P: 75-94.
- Harmer, U. T. 1964. The succession of blooms species of blue-green algae and some causal factors verb Int. Verein Limnol. 15: 829-936.
- Hegde, C.R. and S.G. Bharati. 1985. Comparative phytoplankton ecology of fresh water ponds and lakes of Dharwad, Karnataka, India. Proc. Nat. Symp. Pure & Appl. Limnol. 32:24-29.
- Henson, E. B., Bradshow, A. S. and D. C. Chandler. 1961. The Physical Limnology of Cayuga Lake, N. Y. Cern. Uni. Agr. Exp. Sta Memoir. 378: 3-63.
- Hepher, B. 1962. Primary production in fish ponds and its application to fertilization experiments, Limnol. Oceangro. 7:131-136.
- Hickman, M. 1979a. Phytoplankton of shallow lakes., Seasonal succession standing crop and the chief determinants of primary productivity I. Cooking lake. Alberta Canada. Hola. Ecol. 1: 337-350.
- Hickman, M. 1979b. Seasonal succession, standing crop and determinants of the Phytoplankton of Ministik lake, Alberta, Canada, Hydrob. 64: 105-121.
- Hodgettes, W. J. 1921. A study of the factor controlling the periodicity of fresh water algae in nature. New Phytol., 20: 150-227.

- Hossain, S. M. J. 1987. Studies on some physico-chemical parameters of tidefed shrimp Ponds. Bang. J. Fish. 10(2): 47-56.
- Hussain, M. G., Islam, M. A. and M. E. Chowdhury. 1978. A study of relationship betwee primary productivity and some Limnological parameters in a local Pond in Mymensingh. Bang. J. Fish. 1(2): 111-119.
- Hustedt, F. 1930. Bacillariophyta in paseher,s die siisswasser flora Von. Mitteleuropas. Heft 10. Jena-468.
- Hutchinson, G. E. 1957. A. treatise on Limnology. John-Wiley & Sons Inc. Lond. P(1): 1-
- Hutchinson, G. E. 1967. Introduction to lake Biology and Limnoplankton. John-Wiley & Sons Inc. Lond. P(1): 1-Hydrobad, India 11. Hydrob. 31:101-108.
- Ichimura, S. and Y. Aruga. 1958. Some characteristics of Photosynthesis of fresh water phytoplankton. Bot. Mag. Tokyo. 71(841): 262-269.
- Islam, A. K. M. N and A. Aziz. 1975. Study of Marine Phytoplankton from the North. Eastern Bay of Bengal, Bangladesh, Bang. J. Bot. 4(1-2): 1-32.
- Islam, A. K. M. N and A. Aziz. 1979. Algal flora of Moheshkhali Island, Bangladesh, D. Uni. Stud. B. 27(2): 105-122.
- Islam, A. K. M. N and A. K. Y. Haroon 1975. Limnological studies of the river Burigonga. II. Biological aspects. D. Uni. Stud. B. 23(1): 25-44.
- Islam, A. K. M. N and A. K. Y. Haroon 1978. New reports of some members of chaetophoraceae from Bangladesh. Nova Hed, 29(3-4): 537-545.
- Islam, A. K. M. N and A. K. Y. Haroon 1980. Desmids of Bangladesh. Intern. Rev. Gest. Hydrob. 65(4): 551-601.

- Islam, A. K. M. N and A. R. Chowdhury. 1979. Hydrobiological Studies of Dhanmundi lake, Dacca. II. Phytoplankton. J. Asia. Soc. Bang. (Sc.). 5(2): 47-57.
- Islam, A. K. M. N and F. Mendes 1976. Limnological Studies of a Jheel in Sher-E-Bang. Nagar Dacca. D. Uni. Stud. B. 24(2): 63-71.
- Islam, A. K. M. N and J. K. Shaha. 1975. Limnological Studies of the Ramna lake at Dacca. D. Uni. Stud. B. 23(2): 31-37.
- Islam, A. K. M. N and K. M. Zaman 1974. Fresh water algae of Bangladesh VIII Utrichales. D. Uni. Stud. B. 22(2): 83-98.
- Islam, A. K. M. N and K. M. Zaman 1981. Euglenophyta of Bangladesh I. Genus. *Trachelomonas*. Int. Rev. Ges. Hydrb. 66(1): 109-125.
- Islam, A. K. M. N and M. A Uddin 1978 (b). Blue- green algae from Dacca, Bangladesh. 111. Nostocaceae. D.Uni.stud. B.26 (1): 85-93.
- Islam, A. K. M. N and M. A. Uddin 1978 (a) Blue- green algae from Dacca, Bangladesh. 11. Oscillatoriaceae: D.Uni. Stud. B. 26 (1): 73-84.
- Islam, A. K. M. N and N. Paul. 1978. Hydrobiological Study of the Haor Hakaluki. in Sylhet J. Asia.Soc. Bang. (Sc.). 4(1): 83-91.
- Islam, A. K. M. N and P. sarma 1964(b) Preliminary survey of the epizoic Species of *Oedogonium* on the shells of fresh water Molluscs in East Pak. Rev. Alg. 7 (2): 178-186.
- Islam, A. K. M. N and R. J. M. S. Alam 1980. Members Cladopharaceae of Dacca district. D. Uni. Stud. B. 28(1): 61-70.

- Islam, A. K. M. N and S. K. T. Hossain 1978. Algal floor of the ablution tanks of mosques in Dacca City. J. Asi. Soc.Bang. (Sc.). 4(1): 103-113.
- Islam, A. K. M. N and P. sarma.1965. New and rare Species and varieties of Oedogoniales from Dacca district, East Pakistan, pak. J. Biol. L Agri. Sci. B (1): 169-188.
- Islam, A. K. M. N. 1960. Some Subaerial green algae from East Pakistan Trns. Amer. Micro. Soc. 79: 471-479.
- Islam, A. K. M. N. 1963. A revision of the genus *stigeoclanium*, Nova. Hed. 10: 1-165.
- Islam, A. K. M. N. 1964. The genus *Cladophella* newly found in East Pakistan. Rev. Alg. 7(4): 275-289.
- Islam, A. K. M. N. 1969. A preliminary report on the phytoplankton and other algae of Chittagong Hill Tracts J. Asi. Soc. Pak. 14(3): 343-363.
- Islam, A. K. M. N. 1970. Contribution to the knowledge of Desmids of East Pakistan, Part-1 Nova Hedwigia, 20: 903-983.
- Islam, A. K. M. N. 1972. New and rare species of some green algae from Bangladesh, Nova Hed. 22: 655-663.
- Islam, A. K. M. N. 1973a. Fresh water algae of Bangladesh, 1. Chlorophyceae, Xanthophyceae and Chrysophyceae D. Uni. Stud. B.21(1): 69-84.
- Islam, A. K. M. N. 1973b. Fresh water algae of Bangladesh, III. Chrysophyceae D. Uni. Stud. B.21(2): 133-138.

- Islam, A. K. M. N. 1974a. Fresh water algae of Bangladesh, 1V. Aphanaechaete, Colcochaete & Chaetosphaeridium, Bang. J. Bot. 3(1): 35-43.
- Islam, A. K. M. N. 1980. Study on *Triplastrum* found in Bang with a rate on its species. Bangladesh. J. Bot. 9 (1): 1-12.
- Islam, A. K. M. N. 1982. Marsh algae from southern Iraq. Int. Rev. Ges. Hydrbiol. 67(2): 245-260.
- Islam, A. K. M. N. and A. N. M. Ahia. 1964. Contribution to the knowledge of chaetophoraceae of Dacca district. Pak. J. Biol & Agri. Sci. 7(1) 103-110.
- Islam, A. K. M. N. and D. Sarma. 1968. The characeae of East.Pakistan.1.*Chara* and *Lychnothamnus*. J. Asi.Soc. Pak. 13(3): 357-376.
- Islam, A. K. M. N. and D. Sarma. 1976. the Characeae of Bangladesh. 11. the genus *Nitella*. J. Asia.Soc. Bang. 2(1): 43-61
- Islam, A. K. M. N. and L. Nahar 1967 Ibid, Ibid, 4(2-3):141-149.
- Islam, A. K. M. N. and M. A. Uddin 1973. Fresh water algae of Bangladesh.11. Cyanophyceae D. Uni. Stud. B 21(2):139-148.
- Islam, A. K. M. N. and M. Khatun 1966. Preliminary studies of the phytoplankton of polluted waters. Sci. Res .3(2): 94-109.
- Islam, A. K. M. N. and P. Sarma. 1964(a).contribution to the knowledge of oedogoniales of Dacca district, East Pakistan. Pak.J.Biol and Agri. Sci. 7 (1): 132-135.

- Islam, A. K. M. N. and P. Sarma. 1976. Fresh water algae of Bangladesh. v. genus Oedogonium. Nova Hed, XXX: 1-253.
- Islam, A. K. M. N. and Z. T. Begum 1970. Studies on the plyto plankton of Dacca district. J. Asia. Soc. pak.15(3): 227-271.
- Islam, A. K. M. N.; Haque, M. A.; Islam, M. A.; Karim, R. and M. F. A. Mallah. 1978. A Preliminary of Plankton population of a derelict pond before and after reclamation. Bang. J. Fish. 1(2): 145-154.
- Islam, A. K. M. N.; Anatunnesa and A. K. Y. Haroon 1980. Hydrbiological studies in and around Naogaon, Rajshahi. D. Uni. Stud. B. 28(2): 31-47.
- Islam, A. K. M. N.; Haque, M. A.; Islam, M. A.; Karim, R. and M. F. A. Mallah. 1978. Pattern of recovery of physico-chemical qualities of water and bottom soil of a derelict pond under natural condition. Bang. J. Fish. 1(2): 121-131.
- Islam, A. K. M. N.; Rahman, M. and A. R. Chowdury 1979. Hydrbiological studies of Dhanmundi Lake, Dacca. I. microphytwews and benthic flora, J, Asi, soc, Bang. (Sc.). 5(1): 59-75.
- Islam, A. K. M. N; A. K. Y. Haroon and K. M. Zaman 1974. Limnological studies of the river, Buriganga. I. Physico-chemical aspects. D. Uni. Stud. B 22(2): 99-111.
- Islam, M. A. Chowdhury, M. Y. and R. Karim 1978. A Comparative study of some physico-chemical factors and the growth of Major carps in Ponds. J. Aqua. 1(1): 61-78.
- Iyenger, M. O. P. and M. V. Bai. 1941. Desmids From Kodaikanal South India, Ind. J. Bot. Soc.XX(1-2): 73-103.

- Jackson, D. F. 1961. Comparative studies on Phytoplankton photosynthesis in relation to total alkalinity. Verh. Int. ver. Limnol. 14: 125-133.
- James, H. R. 1941. Beer's law and the prospective of organic matter in lake waters. Trans. wise Acad. Sci. Arti. & Lit., P: 73-82.
- Jana, B. B. 1973. Seasonal periodicity of plankton in a fresh water pond in West Bengal, India, Int. Rev. Ges, Hydrob. 58: 127-143.
- Jayangaudar, I. 1964. The Bio-ecological study of Nuggikari Lake in Dharwas, Mysore state, South India. Hydrob. 23(3-4): 515-532.
- Jhingran, V.G. 1975. Fish and fisheries in India Hindu. pub. corpo New Delhi India.pp.954.
- Kafoid, C. A. 1908. The plankton of the Illinois river, 1894-1899, with introductory notes upon the Illinois river and its basin. III. Constituent organisms and their seasonal distribution, Bull. III. State Lab. Nat. Hist. 8: 1-354.
- Kant, S. and Anand, V. K. 1978. Interrelationship of phytoplankton and physical factors in Mansar lake, Jammu (J. and K.), India. J. Ecol. 52: 134-140.
- Kaul, V. 1985. Primary productivity of inland aquatic ecosystems under varying climate. A Review. Trop. Ecol. 26: 164-178.
- Kern, D. M. 1960. The hydration of carbon-dioxide. J. Chem. Edu. 37: 14-23.
- Khalaf, A.K. and L.J.MacDonald. 1975. Physico-chemical coditions in temporary Ponds in the New Forest. Hydrob. 47:301-318.
- Khan, J.A. and A. Qayyum Siddiqui. 1970. water nitrogen and phosphorus in fresh water plankton. Hydrob. 37(3-4):531-536.

- Khan, S. M., M. H., Aziz, K. M. S., Morshed, M. G. and M. Shafi. 1990. Seasonal variations in physico-chemical conditions of Dhanmondi lake water, Bang. J. Zool. 16(1): 61-66.
- Khanna, D. R. 1933. Ecology and Pollution of Gonga river. Ashish Pub. House, 8/81 New Delhi India .pp 241.
- Khondker, M. 1994. The status of Limnological research in Bangladesh. M. H. Inten Verein Limnol. 24: 147-154.
- Khondker, M. 1995. Limnology, D. Uni. Press. P. 1-464.
- Khondker, M. and L. Perveen. 1992. study on the physical and chemical Limnology of a shallow, hypertrophical artificial lake, Bang. J. Sci. Res. 10(1): 9-16.
- Khondker, M. and S. Rahim. 1991. Investigation on the water quality of Dhanmondi lake: physico-chemical features. Bang. J. Bot. 20(2): 183-191.
- Khondker, M., Islam, A. K. M. N. and R. Islam. 1988. Studies on the primary productivity of Dhanmondi lake. D. Uni. Stud. (P-E), 3: 15-21.
- Khondker, M., Islam, A. K. M. N.; Begum, Z. N. T. and S. Haque, 1990. Limnological Studies of four polluted ponds in and around Dhaka city with reference to indicator species. Bang. J. Bot. 19(1): 51-63.
- Klein, L. 1957. Aspects of river pollution. Butter worth Sci. Pub. London.
- Kolbe, R. W. 1932. Grundlinien einer algameinen Okologieder, Diatomeen, Ergeb. Biol. 8: 221-348.
- Kolthoff, I. M. and V. A. Stenger. 1947. Volumetric analysis. Inte. Sci. Pub. New York. P(2): 242-258.

- Komarek, J. 1975. Blaualgen aus dem naturscutzuwbiet Rayicoe Nova. Hed. XXVI(2-3): 1-644.
- Komarovsky, B. 193. A comparative study of the phytoplankton of several fish ponds in relation to some of the essential chemical constituents of the water. Bull. res. coun. Israel. II(4): 379-410.
- Kumer, D. 1992. Fish culture in undrainable ponds: Mannual for extension, FAO. 325: 1-219.
- Kundu, B. C. 1934. Charophytes of Bengal. II. Chrophyte notes from the district of Rajshahi. Beng. proc. Ind. Sci. Cong: 293-294.
- Kundu, B. C. 1935. Charophytes notes from the district of Dinajpur. Beng. proc. Ind. Sci. Congs: 247
- Kundu, B. C. 1938. A. New Species of *Nitella* from Rajshahi. Beng. J. Ind. Bot. soc. 16: 223-226.
- Lakshminarayana, J. S. S. 1965. Studies on the phytoplankton of the river Ganges. Vanarasi. India. II. The seasonal growth and succession of plankton algae in the river Ganga. Hydrb. 25: 138-165.
- Latif, M. A.; Ali, M. D. M. and M. N. Islam. 1986. A comparative physicochemical study of a well managed fish pond and a direlict pond. Bang. J. Aqua. 6-7(1): 71-78.
- Lauff, G. H. 1953. A contribution to the water chemistry and phytoplankton relationship of Rogers lake. Flathead country. Mont. Proc. Mont. Acad. Sci. 13: 5-19.

- Lind, F. M. 1938. Studies on the periodicity of the algae in the Beauchief ponds, Sheffield. J. Ecol., 26: 257-274.
- Livingstrone, D.A. 1963. Chemical composition of rivers and lakes. U.S. Geol. Surv. Washingtone DC.
- Macan, T. T. 1963. Fresh water Ecology. Lowe and Brydone Ltd. London. Great Britain, P: 1-
- Mahmood, N. 1986. Hydrobiology of Kaptai Reservoir. Final report: FAO/UNDP contact no. DP/BGD/79/615-4/ Fl; 190 pp.
- Mahmood, N.; Y. S. A. Khan and M. K. Ahmed. 1976. Studies on Hydrobilogy of Karnafully estuary. J. Asi. Soc. Bang (Sc.). 1(2): 117-122.
- Mairs, D. F. 1966. A total alkalinity atlas for marine lake waters, Limnol. occanogr, 11: 68-72.
- Marray. C. N. and J. P. Riley. 1969. The solubility of gas in distilled water and sea water. 2. Oxygen. Deep Sea res. 16: 311-320.
- Matsudaria, Y. and T. Kato. 1943. The quantities and the qualities of the impurities contained in the rain water falling in the cities Osaka and Kibe, Japan. Umitosora. 23: 71-86.
- McColl, R.H.S. 1972. Chemistry and Trophic status of New Zealand Lakes. N.Z.J. Mar. Fresh water Res. 6(4):447-461.
- McCombie, A. M. 1953. Factors influencing the growth of phytoplankton, J. Fish. Res. Bd.10(5):
- Mia, M. I.; Bhuiyan, Z. H.; Islam, A. and S. Dewan. 1981. A comparative study on the rate of growth of major carps in relation to physico-chemical and biological factors Proc. 2nd. Nat. Zool. Cong. March. 15-17. 1981. Dhaka.

- Michael, R. G. 1964. Diurnal variation of the plankton correlated with physicochemical factors in three different ponds. Ph. D. Thesis (Unpub), Cal. Uni. P: 1-115.
- Michael, R. G. 1968. Seasonal changes in physico-chemical factors and plankton of fresh water fish ponds and their role in fish culture. Hydrob. P: 144-160.
- Mirza, J. A., Haque, M. R., Haque, A. K. M. A. and M. Y. Chowdhury, 1985. Studies on the phytoplankton of the river old Brahmaputra. B. J. Aqua. 6-7(1): 25-29.
- Mishra, G. P. and A. K. Yadav. 1978. A comparative study of physico-chemical characteristics of river and lake water in central India. Hydrob. 159(3): 275-278.
- Mishra, P.C. and R. K. Trivedy. 1993. Ecology of Pallution of Indian lakes and Reservoirs, Ashish Pub.house. New Delhi.
- Mishra, S.R and D.N. Saksena. 1992. Aqualic Ecology, Asish pub. house 8/81, punjabi Bagh, New Delhi, P: 1-331.
- Moitra, S.K. and B.K. Bhattacharya. 1965. Some hydrological factors affecting plankton production in a fish pond in Kalyani, W. Beng. India Ichthyolo. 4(1-2): 8-12.
- Mollah, M. F. A. and A. K. M. A. Haque 1978(b). Studies on monthly variation of plankton in relation to the physico-chemical conditions of water and bottom soil of two ponds. II. Zooplankton. Bang. J. Fish. 192: 99-103.
- Mollah, M.F.A. and A.K.M.A. Haqu 1978.(a) studies on monthly variation of plankton in relation to the physico-chemical condition of water of and bottom soil of two ponds.1 phytoplankton Bang. J. Fish. 191: 29-39.

- Montogomary, H. A. C.; Thom. N. S. and A. Cockburn. 1964. Determination of dissolved oxygen by the winkler method and solubility of oxygen in pure water and Sea water. J. Appl. Chem. 14: 280-296.
- Morrissette, D.G. and D.S. Mavinic. 1978. BOD test variavles. J. Envir. Engin. Dev. E.P.6:1213-1222.
- Mortimer, C. H. 1956. The Oxygen content of air-saturated fresh waters and aids in calculating percentage saturation. Mitt. Int. Verein. Theor. Angew. Limnol, No. 6.
- Moss, B.1973. The Influence of environmental factors on the distribution of fresh water algae. J.Ecol. 61:157-177.
- Moyle, J. B. 1946. Some indices of lake productivity. Trans. Amer. Fish. SOC., 76: 322-334.
- Mumtazuddin, M.; Rahman, M. S. and G. Mostafa. 1982. Limnological studies of four selected rearing ponds at the aquaculture experiment station, Mymenshingh. Bang. J. Fish. 2-5(1-2): 83-90.
- Munawar, M. 1970a. Limnological studies on fresh water ponds, Hydrabad, India. I. The Biotype. Hydrob. 35: 127-162.
- Munawar, M. 1970b. Limnological studies on fresh water ponds of Hydrabad, India. II. The Biocenose distribution of unicellular and colonial phytoplankton in palluted and unpolluted environments, Hydrob. 36: 105-128.
- Munawar, M.1970. Limnological studies on fresh water ponds of
- Mushfaqua, N. 1995. Studies on the physico-chemical conditions and occurrence of the plankton biota in Ramsagar at Dinajpur, Unpub. M. Sc. Thesis, Dept. Bot. Raj. Uni. Bangladesh. P: 1-134.

- Naj. S. 1995. Limnology of two ponds: Implication for fishery development at Rajshahi. Pond fisheries in Bangladesh. Ed. Wazed A Shai. Envior and Res. Ana. Centre, Dhaka. pp. 101.
- Nasar, S. A. K. and M. Sharma. 1980. Primary productivity in relation to abiotic factors in a temporary fresh water pond, Acta Hydro-chem. Hydrob. 8(5): 435-442.
- Naumann, E. 1927. Dependence of phytoplankton type on water type. Arch. Bot. 3: 1-24.
- Naumann, E. 1941. The Nitrogen content of a moderately eutrophic lake, Z. Fish. U. Hilfswiss. 89(3): 387-405.
- Nautiyal, P. 1984. Studies on riverine Ecology of Torrential waters in the Indian uplands of the Garhwal region. 11. Seasonal fluctuation in Diatom density. Proc. Ind. Acad. Sci. 93(1): 671-674.
- Nayar, C. K. G. 1965. Cyclomorphosis
- Naz, S. 1992. Studies on physico-chemical conditions and plankton of fish ponds in Rajshahi, M. Sc. Thesis (Unpub.). Dept. of Bot. Rajshahi University.
- Neeri. 1979. A course manual of water and waste water analysis.
- Olaniya, M.S., K.L.Saxena and H.C. Sharma. 1977. pollution studies of chambal river and its distributions at Kota. India.J.Enver. Hith. 18(3): 219-226.
- Ooshima, K. 1977. *Pediastrum simplex* and *Pediastrum tetras*. Surface structure of the cell wall of *P. simplex* and *P. tetras*. Chlo. Bull. Nippon. Dental. Uni. Gen. Edu. Japan. P: 133-138.
- Oppenhemer, J.; Ahmed, M. G.; Haque, A.; Haque, K. A.; Alam, A. K. M. A.; Aziz, K. M. S.; Ali, S. and A. S. M. M. Hauqe. 1978. Limnological studies of three ponds in Dacca, Bangladesh. Bang. J. Fish. 1(1): 1-28.

- Pahwa, D. V. and S. N. Mehrotra. 1966. Observation on fluctuations in the abundance of plankton in relation to certain Hydrobiological conditions of river Ganga. proc. Nat. Acad. Sci. Ind. B. 36(2): 157-189.
- Palharya, J. P.; Siriah, V. K. and S. Malviya. 1993. Environmental impact of Sewage and effluent disposal on the river system. Ash. pub. House. 8/81, Punjabi Bagh, New Delhi. P: 1-
- Palmer, C.M. 1967. Environmental needs of nuisance organism. proc. 4th Ann. water quality Res. Symp. N.Y. state Dept. Health. N.Y, U.S.A.
- Pandey, D. C. 1965. A study of the algae of paddy field soil of Ballia and Ghazipur Districts of U. P. India. Part II. (A): Taxonomic consideration Cyanophyceae. Nova. Hed. X (1): 177-209.
- Pandey, R.K., D.S. Rawat and A. Pant. 1988. Seasonal rhythm in the physicochemical properties of Nana Kosi River, in Ecology of pollution of Indian Rivers. Ashish pub. House. New Delhi, India. pp. 294.
- Pankjam, N. 1956. Limnology of Bhavani-sagar Reservoir, part B. plankton, Fish Sta. Report & year book (1954-55). P: 333-350.
- Paramasiviam, M. and A. Sreenivasan. 1981. "Changes in algal flora due to pollution in cauvery rivers" Ind. J. Env. Hith. 23(3): 222-238.
- Patil, P. 1982. An ecological study of the algal flora of the lakes of Bhopal, Ph. D. Thesis (unpub.). Bhopal Uni. India.
- Patra, R. R. W. and M. A. Azadi. 1987. Ecological studies on the planktonic organisms of the Halda river. Bang. J. Zool. 15(2): 109-123.
- Patrick, R. 1945. A. Taxonomic and ecological study of some diams from the pocono plateau and adjacent region. Farlowia, 2: 143-214.

- Patrick, r. 1966. Bacillariophy ceae in Fresh water Biology. Edi. by Edmondson. (2nd. edi.). John Wiley & sons Inc. Lond. P:
- Pearsall, W. H. 1923. The phytoplankton of Rostherne Mere. Mem. Manchest. Lit. Phil. Soc., 67: 45-55.
- Pearsall, W. H. 1924 Phytoplankton and environment in the lake district Rev. Algal., 1: 53-65.
- Pearsall, W. H. 1930. phytoplankton in English lake. I. The proportion in the water of some dissolved substances of Biological importance. J. Ecol. 18: 306-320.
- Pearsall, W. H. 1932. Ibid. II. Ibid. 20: 241-262.
- Pearsall, W. H. and E. M. Linda. 1942. The distribution of phytoplankton in some North-Irish Laughs, proc. RCY. Irish Acad. 488: 1-24.
- Pearsall, W. H.; Gardines, A. C. and S. F. Green. 1946. Fresh water biology and water supply in Britain Fresh water Biol. Ass. British Emp. Soc. pub. II. 1-90.
- Pennak, R. W. 1944. Diurnal movements of Zooplankton organizations in some colorado mountain lakes. Ecol. 25: 387-403.
- Pennak, R. W. 1949. Annual Limnological Cycles in some Colorado reservoir lakes, Ecol. Monogr., 19: 233-267.
- Pennak, R. W. 1955. Comparative Limnology of eight Colorado mountain lakes. Uni. Colo. Stud. Biol., 2: 1-75.
- Philipose, M. T. 1957. Chlorococcales. ICAR, New Delhi, India.

- Philipose, M. T. 1959. Freshwater phytoplankton inland fisheries.proc. symp on algo of the symposium on algology, Dan. 1959.
- Philipose, M. T. 1960. Fresh water phytoplankton on inland fisheries proc. sym. algology. ICAR, New Delhi, P: 272-291.
- Pijler. B. 1957. Taxonoical and ecological studies on plantonic Rotatoria from central Sweden kungl. svenska vetena kasakad. Handl. Fjarda serien. Bd. 6. No. 7.
- Prescott and Vinyard. 1965. Ecology of Alaskan fresh water algae. V. Limnology and flora of Malikpuk lake; Trans. Micros. Soc. 84(4): 427-478.
- Prescott, G. W. 1948. Objectionable algae with reference to the killing of fish and other animals. Hydrob. 1: 1-13.
- Prescott, G. W. 1951. Algae of the western great lakes area: desmids and diatoms: Cranbook Inst. Sc. No. 31.
- Prescott, G. W. and H. T. Croadale. 1955. Algae of Panama Canal and its tributaries. I. Flagellated organisms Ohio. J. Sc. 55: 99-121.
- Prescott. 1969. The Algae: A Review of Thomas Nelson and Sons Ltd. Great Britain.
- Printz, H. 1964. Die chaetophoralen der Binnengewasser. Hydrob. XXIV(1-2): 1-376.
- Qudri, M. Y. and G. M. Shah. 1984. Hydro-biological features of Hokarsar: A typical wetland of Kashmir I. Biotope. Ind. J. Ecol. II. 2: 203-206.

- Rahman, M. S.; Chowdhury, M. Y.; Haque, A. K. M. A. and M. S. Haq. 1982. Limnological studies of four ponds. Bangla. J. Fish. 2-5(1-2): 25-35.
- Rai, H. 1974. Limnological studies on river Yamuna at Delhi. India. Arch. Hydrob. 73(3):369-393.
- Rai, L. C. 1978. Ecological studies of algal communities of the Ganges river at Vanarasi, India. J. Ecol. 5: 1-6.
- Rai, L.C. 197847. Ecological studies of algal Communities of the Ganges river at Varanasi, India. J. Ecol. 5:1-6.
- Raina, V.; Shah, A. R.; Shakti and R. Ahmed. pollution studies on river Jhelum, I an assessment of water quality, Ind. J. Envir. Hith. 26(3): 287-301.
- Rajesh, K. 1981. Some aspects of Limnological studies in Rive Jhelum, Ph. D. Thesis, Kamis Uni. P: 1-420.
- Ramanathan, K. R. 1964. Ulotrichales ICAR. New Delhi, India.
- Randhawa, N. S. 1959. Zygnemaceae. ICAR. New Delhi India.
- Rao, C.B. 1955. Algal periodicity. J.Ecol. 43:291-308.
- Rao, S. V. R. 1979. Impact of pollutants on the primary production of the flora of river Khan, unpub. Ph.D. Thesis in Botany, Vikram Uni. Ujain.
- Rao, V. S. 1976. An ecological study of three fresh water ponds of Hydrabad. India. III. The phytoplankton (Diatoms), Euglenophyceae and Myxophyceae. Hydrob. 531: 13-32.
- Rao, V.S.1970. An Ecological study of three fresh water ponds of Hyderabad, India.1. The Environment. Hydrob. 38:213-223.
- Rao, V. S. 1977. An ecological study of three fresh water ponds of Hyderabad, India. IV. (The phytoplankton Diatoms, Euglenineae and Myxophyceae) Hydrobiol. 53(1): 13-32.

- Rawson, D. S. 1944. Nomogram for obtaining oxygen saturation values at different temperature and at different attitudes liminal. Soc. Amer. 15(1944) (c). Welch 1948.
- Rawson, D. S. 1958. Indies to lake productivity and their significance in predicting condition reservoirs and lakes with disturbed water levels. Invest in fish power problem Uni. British Columbia. P: 27-43.
- Ray P. and A. David. 1966. Effects of Industrial wastes and sewage upon the chemical and biological composition and fisheries of the river Ganga of Kanpur. Envr. Hith 8(4): 307-339.
- Ray, P.; Singh, S. B. and K. L. Sehgal. 1966. A study of some aspects of Ecology of the rivers Ganga and Jamuna at Allahabad (U.P.) in 1958-1959. Proc. Net. Acad. Sci. Ind. Sect. B. 3: 235-272.
- Raymond, M. R. 1937. A Limnological study of the plankton of a concentration forming Marl lake. Trans. Amer. Micro. Soc. 46(4): 405-430.
- Reid, G. K. 1961. Ecology of inland water and estuaries. Litton Edu. pub. Incp. P: 1-375.
- Rice, C. H. 1938. studies in the phytoplankton of the river Thames, I & II. Ann. Bot. 32: 539-581.
- Richard, L.W. 1966. Environmental hazard of water pollution. New Engl. J. Medi. 275:819-825.
- Riley, G. A. and D. F. Bumpus. 1946. phytoplankton-Zooplankton relationship on Georges bank J. Mar. Res. VI: 33-46.
- Robert, D.H. 1969. Water pollution . Biococenos. 19:476.

- Ronald, R. W. P. and M. A. Azadi. 1987. Ecological studies on the planktonic organisms of the Halda river, Bang. J. Zool. 15(2):109-123.
- Rosenberg, M. 1939. Algal physiology and organic production. Ann. Appl. Biol. 26: 172-174.
- Ruther, F. 1937. Limnology. Study an einigen seen der ostalpen Arch. Hydrob. 32:167-319.
- Ruttner, F. 1953. Fundamental of Limnology. Toronto Uni. Press. P: 1-295.
- Ruttner, F. 1953. Fundamentals of Limnology. Uni. Toronto, Toronto Press. pp. 295.
- Saha, I.C. and B. Pandit. 1986. Comparative limnology of Bhagalpur Ponds. Comp. Physiol. Ecol. 11(4):213-216.
- Sahai and Sihna, A. B. 1969. Investigation on Bioecology of inland water of Gorakh pur. Hydrob. 34: 433-447.
- Saito, E. and T. Yamagishi. 1975. Genus Oedogocladium in Japan. Bull Jap. Soc. Phycol. XXIII(2): 53-59.
- Sawyer, N.C. and L.P. Mc Carty. 1978. Chemistry of Environmental helth. Vol. 26.No. 3.187-201.
- Schaindler, D. W.; Armstrong, F. A. J.; Holingren, S. K. and G. J. Brunskill. 1971. Eutrophication of lake 227, Experimental lakes area, N. W. Ontario, by addition phosphate and Nitrate, J. Fish, Res. Bd. Canada, 28: 1763-1782.
- Schindler and Fee. 1973. Diurnal Variation of dissolved inorganic carbon and its use in lake, J. Fish. Res. Bd. Canada. 80: 1501-1510.
- Schindler, D. W.; Brunskill, G. J.; Emerson, S.; Broecher, W. S. and T. H. Peng. 1972. Atmospheric carbon-dioxide its. role in maintaining phytoplankton standing crop. Sci. 177: 192-194.

- Sengar, R. M. S.; Sharma, K. D. and P. D. Pathak. 1985. Studies on distribution of algal flora in polluted and non-polluted regions in Yamuna river at Agra (U. P.) J. Ind. Bot. Soc. 64: 365-376.
- Shafi, M.; Quddus, M. M. and N. Islam. 1978. Studies on the limnology of river Meghna. Bang. J. Fish. 1(2): 85-97.
- Shrivastava, G.L. 1986. Fisheries of Easter Uttar Pradesh. Vishwavi. Prakash, Varanasi. pp 165.
- Singh, A. K. and S. H. and S. H. Ahmed, 1990. A. Comparative study of the phytoplankton of the river Ganga and pond of Patna (Bihar), India. J. Indian. Bot. Soc. Vol. 69: 153-158.
- Singh, A.K. and B.N. Bhaumick. 1985. Effect of sewage on physico-chemical characteristics and Bacterial population on river Ganga at Patna. Ind. J. Ecol.12(1).
- Singh, M. 1968. Phytoplankton periodicity in a small lake near Delhi. I. Seasonal fluctuations of the main group of phytoplankton. Phycos. 7(192): 126-135.
- Singh, R.N. 1953. Limnological relation of Indian inland waters with special reference to water blooms. Verh.Int.Ver. Ther. A new. Limnol. 12: 831-836.
- Singh, Y. 1960. Phytoplankton ecology of the inland waters of uttar pradesh proc. sym. algal. ICAR. New DElhi. P: 243-271.
- Sinha, M.P. 1988. Effect of waste disposal on water quality of river Damodar in Bihar. Ashish pub. New Delhi India. pp.326.
- Smirnov, N. N. 1964. *Pleuroxus* (Chydoridae), field observation and growth. Hydrob. 3-4: 305-320
- Smyle, W. J. P. 1957. Distribution and seasonal abundance of Entrmostraca in moorland pond near windermere. Hydrbiol. 11: 59-72.

- Sohai, P. and A.B. Sinha. 1979. Studies on some limnological features of Jhalwani pond of Gorakhpur. Proc. Nat. Acad. Sci. India 49(B):W.
- Sreenivasan, A. 1963. Primary production in three upland lakes of Madras state, India, Curr. Sci. 32:130-131.
- Sreenivasan, A. 1964. A hydrological study of a tropical impoundment, Bhavani sagar reservoir, Madras, India, for three years 1956-61. Hydrob. XXV(3-4): 501-516.
- Sreenivasan, A. 1964. A hydrological study of a tropical impoundment, Bhavani sagar reservoir, Madras state, India, for the years 1956-1961. Hydrobiol XXV (3-4): 501-516.
- Sreenivasan, A. 1965. Limnological study of a tropical impoundment. III. Limnology and productivity of Amarabathi reservoir (Madras state), India, Hydrob. XXV(3-4): 501-516.
- Sreenivasan, A. 1965. Limnology of tropical impoundments. I. Hydrobiological features and fish production in the Stanley reservoir, Mittur Dam. Int. Rev. Ges. Hydrob. 51:295-306.
- Sreenivasan, A. 1965. Limnology of tropical impoundment, I. Hydrbiological features and fish production in the stanley reservoir, Mettur Dam Int. Rev. Ges. Hydrob. 51: 295-306.
- Straskraba, M. 1965. Contribution to the productivity of the littoral region of pools and ponds. I. Quantitative study of the littoral zooplankton of the rich vegetation of the back water labicko Hydrbiol. XXV(3-4): 421-443.
- Strm, K. M. 1930. Limnological observations on Narwegian lakes, Arch. Hydrb. 21:97-124.

- Stumn, W. and J. J. Morgan. 1970. Aquatic Chemistry. An Introduction Emphasizing chemical Equilibrium in natural water, Wiley, N. Y. P: 1-583.
- Sulka, A. C. 1971. Systematic description of algae from Panki rice field. Rev. Alg. X (3): 257-270.
- Sundar, S. 1988. Mounting the water quality in a stretch of river Jhelum, Kashmir Ashish pub. House. New Delhi India pp. 161.
- Swarup, R., S. N. Mishara and V.P. Jauhari. 1992. Encyclopaedia of Ecology, Environment and Pollution Control. Envir. Air and water Ana. vol. 17.
- Talling, J. F. 1957. The Limnological succession of the water on characteristics in white Nile Hydrb. 9:73-89.
- Tarzwell, C. M. and C. M. Palmer. 1951. Ecology of significant organisms in surface water supplies J. Amer. water works. Ass. 43(7): 568-578.
- Thomas, J.F.J.1966. Churchill river and Mississippi river dainage basin in .
- Thresh, J.C. and E.V. Suckingg and J.F. Beale. 1944. The Examination of water supplies.
- Tiffany, L.H. 1928. The algal genus *Bulbochaets*. Trans. Amer. Micro. Soc. 4847(2): 121-177.
- Tiffany, L.H. and M. E. Britaon. 1952. Algae of Illinois. Uni. Chicago. Press.
- Transeau, E. N. 1916. Periodicity of fresh water Algae. Amer. J. Bot. 3: 121-133.
- Transeau, E. N. 1951. Zygnemaceae. Ohio state Uni. Press. Colum. U. S. A.
- Tripathy, A.K. and S.N. Pandey. 1933. water Pollution. Ashish Pub. House. 8/81, New Delhi India. pp. 326.
- Trivedi, P. R. 1990. River pollution bagh, New Delhi. P: 1-294.

- Trivedi, P. R. and G. Raj. 1992. Encyclopedia of Environmental Sciences-25, water pollution, Akashdeep pub. house. New Delhi.
- Trivedy, R.K. 1990. River pollution in India. Ashish pub. House. New Delhi, India. pp.294.
- Tuckur, A-1957. The relation of phytoplankton periodicity to the nature of physico-chemical environment with special reference to phosphorus, Amer. Midl. Nature. Vol. 57.
- Van, O. P. 1934. Quelques données sur l'ecologe des desmidices. Bull. SOC. Roy. Bot. Belg. 69 : 65-75.
- Vass, K. K.; Raina, H. S.; Zutshi, D. P. and M. A. Khan. 1977 Hydrological studies of river Jhelum, Geobios, 4: 238-242.
- Venkateswarlu, V. 1969. The algal Periodicity. Hydrob. 34:533-560.
- Venkateswarlu, V. 1986. "Ecological studies on the river Andhra Pradesh with special reference to water quality and pollution" proc. Indi. Sci. Acad. 96(6): 495-508.
- Venkteswarlu, T. 1969. An ecological study of the algae of the river Moosi, Hydrabad, (India), with special reference to the water pollution. I. physico-chemical complexes. Hydrob. 2291): 117-143.
- Voznaya, N.F.1981. Chemitry of water and microbiology. Mir. Pub. Moscow.
- Vyas, L. N. 1968. Studies in Phytoplankton ecology of Picchola lake, Udaipur, Proc. Symp. Recent advances in Trop. Ecol. pp 347.
- Vyas, L. N. 1968. Studies on in phytoplankton ecology of Picchola lake, Udaipur. proc. sym. Recent Adv. Trop. Ecol. P: 334-347.
- Watanabe, M. and A. K. M. N. Islam. 1980. Fresh water algae lake Akan (4). Japan J. Phyco. 28: 37-45.

- Watanabe, M.; Prescott, G. W. and T. Yamagishi. 1979. Frsh water algae of Papua New Guinea. (2). Desmids from Waitape, central district, Stud. Croipto, Papua N. Guinea 49-46. Acad. Sci. Book Inc. Tokyo.
- Weive, A. H. 1930. Investigation in Plankton production in fish ponds. Bull. U.S. Bor. Fish. 46: 137-176.
- Welch S. P. 1948. Limnological methods. McGraw-Hill Book Co. New York.
- Welch S. P. 1952. Limnology. McGraw-Hill Book Co. New York.
- Welch, S. P. 1935. Limnology McGraw-Hill Book Co. New York and London. P: 1-538.
- West, G. S. and F. E. Fritsch. 1927. A Treatise on the British fresh water algae. Cambridge Uni. Press.
- Wetgel, R.g 36. 1975. Limnology. W.B.Saunders Co.Philadelphia. U.S.A.pp.743.
- Wetzel, R. G. 1966. Limnology, Verh. Inten at. Verein Limnol. P: 16-321.
- Whipple, G. C. and H. N. Parker. 1902. On the amount of Oxygen and carbonic acid dissolved in natural waters and the occurrance of microscopic organisms. Trans. Amer. Micro. SOC. 23: 103-144.
- White, E.; Downes, M.; Gibbs, M.; Kemp, L.; Mackenzie, L. and G. Payne. 1980. Aspects of Physics, Chemistry and phytoplankton biology of lake Taupo. NZ. J. Marine & Fresh water Res. 14(2): 139-148.
- Whitton, B. A. 1975. In Biological indicators of water quality, John Wiley and Sons. New York.
- Who. 1982. Examination of water for population contro. a reference book, Vol. No. 1,2.

- Wohlschlang, D. E. and A. D. Hasler. 1951. Some qualitative aspects of algal growth in lake Mend. Ecol. 32(4): 581.
- Yakubson, S. 1980. phytoplankton of some fresh water bodies from Zulia state (Venuzuela). Nova. Hed. 33: 279-339.
- Yakunson, S. 1980. phytoplankton of some fresh water bodies from zulia state (venzuela). Nova. Hed. 33: 279-339.
- Yamagishi, T. 1970. A Check list of the Euglenophyceae and Chrysophyceae in the Alaskan Arctic. Gen. Edu. Rev. College. Agri & Vet. Med. Nihor Uni. 6:31-35.
- Yamagishi, T. 1975. Plankton algae from Papua New Guinea Nat. Sci. Mus. Tokyo. 2: 43-74.
- Yamagishi, T. and E. Saito. 1971. Studies Oedogoniaceae from Combodia. Gen. Edu. Rev. College Agri & Vet. Med; Nihon. Uni. 8: 35-39.
- Yamagishi, T. and H. Kobayasi. 1971. Algae from Sphagrum bogs of mount Omine. Gen. Edu. Rev. College. Agri. & Vet. Med., Nihon, Uni. 7: 25-51.
- Yamagishi, T. and M. Hirano. 1973. Some fresh water algae from Combodia. Contib Biol. Lab. Kyoto. Uni. 24(2): 61-85.
- Yaron, Z. 1964. Notes on the ecology and Entromostracan fauna of temporary rainpools in Israil. Hydrob. 3-4: 489-513.
- Yoshimura, S. 1932. Seasonal variation of Contents of Nitrogenous compounds and phosphates in water of Takasuka pond. Saitama. Japan. Arch. Hydrob. 24 (1): 155-176.

- Zafar, A. R. 1964. On the Ecology of algae in certain fish ponds in Hydarabad, India, I. Physico-Chemical complexes, Hydrob. 23: 179-195.
- Zafar, A. R. 1967. On the Ecology of algae in fish ponds of Hydarabad, India, III. The periodicity. Hydrob. 30 (1): 96-112.
- Zafar, A.R. 1966. Limnology of the Hussain Sagar Lake, Hyderabad, India. Phykos., 5:115-126.
- Zaman, L.; Khondker, M. and M. R. Nabi. 1993. A comparative limnology of three ponds in Jahangirnagar University Campus: Physical and Chemical aspects, Bangla J. Bot. 22(1): 81-87.
- Zaman, M. 1991. Studies on the algal flora of Chalan Beel in relation to its physico-chemical conditions, unpub. Ph. D. Thesis. Phycology and Limnology lab. Dept. Bot. Raj. Uni. Rajshahi. P: 1-441.

CHAPTER -6

APPENDIX



Factors	May '94	June	July	Aug	Sep	Oct	Nov	Dec	Jan	Feb	March	April '95
Air temp. C ⁰	34.8 ±	29.1 ±	32.1 ±	33.6 ±	30.3 ±	27.6 ±	23.1 ± 2.2	20.2 ± 2,9	17.5 ±	23.8 ± 2.5	32.4 ± 2.4	38.3 ±
	1.6 31.8±	2.9 28.9 ±	1.6 31.1 ±	2.9 31.5 ±	0.4 30.4 ±	27.7±	24.1 ±	19.3 ±	3.3 19.6 ±	23.9 ±	29.0 ±	30.8 ±
Water temp, C ⁰	0.9	0.8	0.7	1.4	0.6	1.7	2.4	2	1.3	2.2	1.7	1.3
Av. depth .m.	1.1 ± 0.1	1.1 ± 0.1	1.1 ± 0.1	1.2 ±	0.9 ±	0.9 ± 0	1.2 ± 0	1.0 ± 0.1	1.0 ± 0	1.0 ± 0.1	1.0 ± 0.1	1.1 ± 0.1
S. D. depth. m.	0.2 ± 0	0.2 ± 0	0.2 ± 0	0.2 <u>+</u> 0	0.1 ± 0	0.2 ± 0.1	0.2 ± 0.1	0.2 ± 0	0.3 ± 0	0.2 ± 0	0.2 ± 0	0.2± 0
T. D. S. mg/l	489.5	494.3	484.3	454.7	353.0	456.0	488.8	450.5	440.4	484.3	444.0	420.2
E. cond. μ moh/cm	753 ±	751.± 37	745 ± 2	700 ±	543 ± 46	702 ± 66	752 ± 14	693 ± 46	678 ± 39	745 ± 14	683 ± 62	647 ± 38
pН	7.7 ± 0.1	8.3 ± 0.1	8.3 ± 0.1	8.3 ± 0.1	8.5 ± 0.1	8.7 ± 0.1	8.6 ± 0.1	8.6 ± 0.1	8.6± 0.1	3.6± 0.2	8.7± 0.1	8.7± 0.1
Free CO ₂ mg/l	21.5 ± 6.0	9.7 ± 3.8	10.1 ± 6.8	2.8 ± 6.2	26.7 ± 14.4	23.5 ± 6.7	13.0 ± 3.3	27.3 ± 3.6	15.7 ± 15	33.2 ± 8.6	34.7± 6.4	28.0± 4.7
CO3 Alk. mg/l	0± 0	0 ± 0	0 ± 0	17.2 ± 8.4	0±0	0 ± 0	0 ± 0	0 ± 0	2.8 ± 4.0	0 ± 0	0 ± 0	0 ± 0
HCO3 alk. mg/l	30.5 ± 8.0	23.5 ± 3.2	21.0 ± 1.7	49.0 ± 24.0	114.1± 14.4	165.3 ± 4.3	135.0 ± 26.0	184.0 ± 22.0	231.0 ± 16.1	301.2 ± 39.0	346.3 ± 30.0	403.0 ± 56.0
Total hardness mg/l	209.0 ± 13.3	295.0 ± 48.1	287.0 ± 60.0	200.0 ± 18.0	209.0 ± 26 .0	267.0 ± 36.2	285.0 ± 31.0	386.0± 113.0	467.3 ± 31.0	440.0 ± 95.0	429.0 ± 50.0	538.0 ± 147.0
Chloride mg/l	112.0	121.0	103.0	143.0	102.0	109.0	99.0	87.0	107.0	107.0	122.0	131.0
DO mg/l	6.1 ± 1.1	5.2 ± 0.8	6.3 ± 0.7	6.8± 0.8	6.9± 1.0	6.3 ± 0.1	5.7 ± 0.9	5.4 ± 0.9	7.6± 1.3	4.0 ± 1.1	5.4 ± 1.4	5.4 ± 1.3
% of sat O ₂	82.0	68.0	86.0	92.0	92.0	81.0	69.0	60.0	86.0	45.0	70.0	73.0
BOD ₅ mg/l	6.9	7.6	5.4	12.0	3.4	3.1	4.9	5.6	7.0	3.2	3.1	2.7
COD mg/l	15.3	18.8	11.9	22.5	8.4	7.2	10.6	12.2	15.2	9.8	8.5	7.4
NH ₃ -N mg/l	0.10 ± 0	0.10 ± 0	0.10 ± 0	0.10 ± 0	0. 10 ± 0	0.10 ± 0	0.11±0	0.10 ± 0	0.13 ± 0	0.10 ± 0	0.10 ± 0	0±0
PO ₄ mg/l	0.1	0.1	0.1	0.2	0. 1	0.2	0.1	0.1	0.7	0.2	0.7	0.2
Eh mv	0.36	0.29	0.29	0.29	0. 27	0.25	0.25	0.26	0.26	0.26	0.25	0.25
rH ₂	27.81	26.60	26.60	26.60	26.31	26.02	25.82	26.17	26.17	26.17	26.02	26.02
GPP mg C/l/h	0.2	0.2	0.3	0.3	0.2	0.3	0.2	0.2	0.4	0.3	0.4	0.3
NPP mgC/l/h	0.10	-0.10	0.04	0.02	0.04	0.03	-0.11	-0.14	0.10	0.12	0.14	0.08

Factors	May '94	June'94	July'94	Aug'94	Sep'94	Oct'94	Nov'94	Dec'94	Jan'95	Feb'95	March'95	April'95
Air temp. C ⁰	34.8 <u>+</u> 1.6	28.6± 2.4	32.0± 1.6	33.6± 2.9	30.3± 0.4	27.6± 1.2	23.1± 2.3	20.2± 2.9	17.5± 3.3	23.8± 2.5	32.6± 2.4	38.7 ± 1.9
Water temp. C ⁰	32.8 <u>+</u> 0.9	29.7± 0.8	31.0± 0.6	32.1± 1.2	31.0± 0.7	27.4± 1.6	23,1± 1.6	18.4± 1.6	18.5± 1.8	24.0± 2.4	29.2± 2.8	31.8 ± 1.7
Av. depth. M.	1.7± 0.1	1.6± 0.1	1.5± 0.1	1.8± 0	1.5± 0	1.4± 0	1.4± 0	1.4± 0	1.6± 0	1.4± 0	1.4± 0	1.8 ± 0.1
S. D. depth .M.	0.2 <u>+</u> 0.1	0.2 <u>+</u> 0.1	0.2± 0.1	0.2 <u>+</u> 0.1	0.2 <u>+</u> 0.1	0.2 <u>+</u> 0.1	0.2± 0.1	0.4± 0.1	0.2± 0	0.1± 0	0.3± 0.1	0.2 ± 0
T. D. S. mg/l	404.3	418.0	455.0	383.0	344.0	499.0	435.0	482.0	440.4	406.9	436.2	408.9
E. cond.μ moh/Cm	422.0 <u>+</u> 16.0	443.0± 27.0	700.0± 6.0	589.0± 0	529.0± 32.0	767.0± 37.0	669.0± 69.0	741.0± 44.0	677.5± 38.5	709.0± 19.0	671.0 ± 60.0	629.0 ± 115.0
pН	7.8± 0.1	8.2± 0.2	8.4± 0.3	8.2± 0.1	8.4± 0.1	8.7± 0.1	8.7± 0.1	8.7 ± 0.2	8.8 ± 0.1	8.9 ± 0.1	8.9 ± 0.1	8.8 ± 0.2
Free CO ₂ mg/l	14.7± 4.9	8.2± 2.0	8.9± 8.6	8.6± 7.5	19.0± 11.7	8.3± 10.7	11.3± 10.4	32.5± 6.7	4.9± 5.8	13.1± 18.8	12.2 ± 10.3	4.3 ± 9.5
CO3 alk. mg/l	0 ± 0	0 ± 0	0 ± 0	4.8± 6.8	0.9± 2.1	2.7± 3.4	2.8± 4.1	0 ± 0	7.3± 8.0	10.8± 9.1	5.3 ± 7.6	15.2 ± 8.7
HCO3 alk. mg/l	21.2± 2.3	18.8± 1.7	18.7± 2.5	41.3± 22.4	109.7± 15.1	160.5± 13.1	141.3± 21.2	171.4± 16.6	195.3± 21.5	290.1± 48.1	329.7 ± 4.2	356.7 ± 122.6
Total hardness mg/l	285.5± 98.8	307.0± 32.1	227.5± 41.3	226.2± 21.0	214.2± 43.1	317.7± 36.8	311.0± 20.4	414.6± 112.1	487.0± 63.0	464.8± 24.7	465.2 ± 63.0	609.2 ± 56.3
Chloride mg/l	138.7	121.6	191.7	133.7	136.9	129.8	146.7	104.6	121.7	120.6	103.6	136.6
DO mg/l	6.0± 1.1	6.1± 1.0	6.3± 1.0	7.6± 1.2	8.4± 1.1	8.1± 0.4	6.8± 1.4	5.0± 1.5	7.9 ± 0.8	7.1 ± 1.8	6.8 ± 1.6	8.4 ± 1.1
% of sat O₂	83.0	80.0	85.3	105.0	113.0	104.0	81.0	53.2	86.5	85.5	90.0	114.4
BOD ₅ mg/l	3.2	2.3	8.5	9.6	3.9	8.7	3.6	3.7	3.4	2.9	2.5	3.7
CODmg/l	7.3	6.7	14.5	20,7	8.7	12.8	8.2	10.1	8.7	7.3	7.1	8.4
NH3-N mg/l	0.10±0	0.10±0	0.10±0	0.10±0	0.10±0	0.10±0	0.11±0	0.12 ± 0	0.10±0	0.10±0	0.10 ± 0	0
PO₄ mg/l	0.1	0.1	0.1	0.1	0.1	0.02	0.1	0.1	0.1	0.03	0.1	0.1
Eh. mv	0.35	0.31	0.28	0.31	0.28	0.25	0.25	0.25	0.24	0.23	0.22	0.24
rH ₂	27.67	27.09	26.46	27.09	26.46	26.02	26.02	26.02	25.88	25.73	25.39	25.88
GPP mgC/l/h	0.3	0.3	0.4	0.4	0.3	0.3	0.3	0.1	0.4	0.3	0.2	0.3
NPP mgC/l/h	-0.9.3	0.10	-0.10	0.10	0.01	0.01	0.21	-0.16	0.05	0.15	-0.07	0.02

Factors	May '94	June	July	Aug	Sep	Oct	Nov	Dec	Jan	Feb	March	April '95
Air temp. C ⁰	33.8 ± 1.8	30.4 ± 2.5	30.2 ± 1.2	32.6 ± 2.0	28.5 ± 0.9	27.4 ± 1.9	22.8± 16	19.3± 21	19.0 ± 3.6	24.3 ± 2.6	32 .6± 1.7	35.1 ± 0.8
Water temp. C ⁰	32.4 ± 1.3	29.8 ± 1.0	31.1 ± 0.4	31.9 ± 1.0	29.2 ± 0.6	29.0 ± 1.3	22.9 ± 12	19.0 ±	20.1 ± 1.2	24.3 ± 1.2	29.3 ± 0	32.5 ± 1.0
Av. depth .m.	2.2 ± 0.1	2.1 ± 0	2.1 ± 0.1	2.3 ± 0.1	1.7± 0.4	2.0 ± 0	1.7± 0.1	1.7± 0.2	1.9± 0	1.8 ± 0.1	1.8 ± 0.2	2.0 ± 0.2
S. D. depth. m.	0.2 ± 0	0.2 ± 0	0.3 ± 0	0.2 <u>+</u> 0	0.2 ± 0	0.2 ± 0	0.2 ± 0	0.2 ± 0	0.1± 0	0.3 ± 0.1	0.2 ± 0	0.2 ± 0
T. D. S. mg/l	397.2	412.1	419.3	381.6	387.7	389.4	448.5	447.9	450.1	441.7	445.9	423.2
E. cond. μ moh/cm	611.0 ± 5.0	634.0 ± 18.0	645.0 ± 7.0	587.0 ± 2.0	597.0 ± 36.0	599.0 ± 0	690.0 48.0	689.0± 73.0	693.0± 47.0	680.0 ± 37.0	686.0 ± 45.0	651.0± 137.0
pН	8.2 ± 0.1	8.5 ± 0.1	8.4 ± 0.1	8.4 ± 0.1	8.6 ± 0.2	8.6 ± 0.1	8.6 ± 0.1	8.6 ± 0.1	8.7± 0.1	3.8 ± 0.1	8.7 ± 0.1	8.9± 0.1
Free CO ₂ mg/l	1.3 ± 3.1	0± 0	2.8± 4.1	4.3 ± 6.1	6.3 ± 4.0	2.9 ± 4.5	3.8 ± 7.5	0±0	0 ± 0	6.4 ± 7.0	14.6± 15.2	1.0 ± 2.3
CO₃ Alk. mg/l	4.8 ± 2.3	1.8 ± 1.1	1.3 ± 1.6	1.2 ± 1.6	3.2 ± 2.4	1.7 ± 1.2	3.3 ±2.4	4.3±1.7	8.0± 4.7	1.4 ± 1.7	1.8 ± 1.9	2.7 ± 1.7
HCO3 alk. mg/l	25.6± 6.5	15.1 ± 4.5	16.7 ± 2.3	81.3 ± 30.4	113.8 ± 10.9	134.8 ± 38.6	114 2 ± 34.2	156.4± 31.1	218.8 ± 12.9	283.7 ± 51.4	298.2 ± 60.5	445.2± 60.5
Total hardness mg/l	240.2 ± 102.3	273.0 ± 54.0	228.7± 60.0	246.8 ± 22.7	238.7 ± 32.0	247.5 ± 27.0	271.7± 18.0	434.3 ± 31.0	424.0 ± 25.0	466.2 ± 50.5	500.0 ± 24.4	608.8± 44.6
Chloride mg/l	132.4	142.0	103.4	113.6	107.5	132.1	121.6	188.7	201.3	178.0	140.7	156.7
DO mg/l	7.7 ± 1.0	7.7 ± 1.2	7.3 ± 1.3	7.0± 0.9	9.6 ± 1. 4	6.8 ± 2.2	5.9± 1.3	9.1 ± 1.1	8.2± 0.9	5.7± 2.8	7.1 ± 1.8	8.3 ± 2.8
% of sat O ₂	105.4	101.9	98.9	95.5	126.0	89.1	70.0	101.2	93.0	69.2	93.4	114.2
BOD ₅ mg/l	9.0	13.6	13.0	9.4	6.7	12.8	7.4	7.6	3.4	3.2	11.6	10.6
COD mg/l	19.6	28.7	25.8	22.3	14.5	27.7	16.5	18.9	9.3	8.7	26.3	20.8
NH3-N mg/l	0.1 ± 0	0.10 ± 0	0.10 ± 0	0.10 ± 0	0. 10 ± 0	0.10 ± 0	0.11± 0	0.13 ± 0	0.1 ± 0	0.1 ± 0	0.1 ± 0	0±0
PO ₄ mg/l	0.1	0.3	0.1	0.03	0. 1	0.1	0.2	0.3	0.1	0.01	0.03	0.1
Eh mv	0.31	0.27	0.28	0.28	0. 26	0.26	0.26	0.26	0.25	0.23	0.25	0.23
rH ₂	27.09	25.71	26.46	26.46	26.17	26.17	26.17	26.17	25.02	25.52	26.02	25.73
GPP mg C/l/h	0.4	0.3	0.4	0.3	0.3	0.3	0.2	0.3	0.3	0.3	0.3	0.2
NPP mg C/Mh	0.03	0.03	0.10	-0.01	0.02	0.11	0.10	0.10	0,10	-0.20	0.10	0.03

Physico-chemical condition of P₄ (Study pond-4, Tab-4.

Factors	May '94	June'94	July'94	Aug'94	Sep'94	Oct'94	No. 294	Dec'94	Jan'95	Feb'95	March'95	April'95
Air temp. C ⁶	33.8± 1.8	30.4±	30.2±	32.6± 2.0	28.5±	27.4± 1.9	22.8± 1.6	19.3± 2.1	, 19.0 ±	24.4 ± 2.6	32.6± 1.7	35.1 ± 0.8
Water temp. C ⁰	32.5 ± 0.9	2.5 29.7± 0.8	1.2 30.9±	31.6± 1.0	29.6±	28.7± 1.7	23.0±	19.1± 1.2	20.3 ±	24.4 ± 1.8	29.4 ±	32.6± 0.9
Av. depth. m.	1.6± 0.1	1.6± 0	1.8± 0.	2.0± 0	1.4± 0.2	1.4± 0	1.5± 0	1.7± 0.3	1.4± 0	1.4 ± 0	1.6± 0.1	1.9 ± 0.2
S. D. depth.m.	0.2± 0	0.2±	0.2=	0.2±	0.3± 0.1	0.2± 0	0.2±	0.1±0	0.2 ± D	· 0.2 ±	0.2± 0	0.2 ±
T. D. S. mg/!	362.7	382.5	403.3	354.9	380.3	370.8	404.6	479.6	397.5	1 451.4	378,6	351.3
E. cond.µ moh/Cm	553.0 <u>+</u> 0	589.0± 28.0	621.0± 5.0	346.0± 1.0	585.0± 31.0	571.0± 10.0	623.0± 102.0	724.0± 61.0	: 612.0± 25.0	695.0 ± 9.0	583.0± 10.0	541.0 ± 45.0
pH	8.4± 0.1	8.5± 0.1	8.1± 0.1	8.4± 0.1	3. 6 4 . 0.1	8.6± 0.1	3.5± 0.1	8.6 ± 0.1	€.7± 0.1	8.9 ± 0.2	8.8± 0.1	8.9 ± 0.1
Free CO ₂ mg/l	2,3± 5.2	0± 0	0=	0 <u>±</u> 0	+ 0± : 0	4.0± ♣.4	4.3± ↓.5	0± 0	5.4 ± 0 .7	0± 0	0± 0	0±
CO ₃ alk. mg/l	4.9 ± 3.3	2.7 ± 1.2	2.3± 1.4	5.8± 1.8	4.8± - 2.5	1.8± 1.9	2.5± 2.0	5.3± 2.3	4.3±3.4	7.7 ± 1.3	6.6± 0.9	6.6 ± . 2.2
HCO, alk mg'l	21.3=	16.3±	15.2± 3.2	73.0± 22.2	113.8± 12.0	130,4± 6,5	102.0± 17.4	133.5± 26.0	174.0±	251.3 ± 25.3	251.0 ± 34.5	336.6 ± 35.2
Total hardness myle	268.0± 1.0	272.0± 28.4	197.8=	221.0± 16.3	216.0 <u>4</u> 16.4	232.0± 6.0	228.3 . 15.0	340.3± 22.0	317.3 ± 46.4	377.2 ± 24.0	409.0 ± 37.6	469.3 ± 110.6
Chloride mg/l	98.6	132.0	: 137.0	109.2	113.6	142.0	137.0	121.0	98.6	93.4	110.6	122.6
DO mg/l	7.1± 1.0	6.3± 0.5	6.4± €.5	! 6.7± : 0.8	± 10.6± 3.5	7.9±	: 6.5≘ 1.8	1 7.9± 1.3	7.9± 6.7	8.2 ±	1.7±	6.2 ±
9n of sat (h	97.4	83.7	86.5	91.6	140.0	101.2	77.5	\$6.0	. 90.1	100.0	101.0	113.6
BODs mg/l	12.6	16.8	13.3	14.7	12.4	12.5	16.8	12.7	13.6	9.3	8.6	13.0
CODmg/l	25.8	37.2	26.4-	; 31.2	26.≟	27.3	33.6	22.8	24.5	20 ₹	17.3	25.9
NH ₃ -N mg/l	0.10±0	0.10±0	0.10±0	0.10±0	0.19±0	0.10±0	6.1±0	0.1 ± 0	; 0.11± 0	0.12± 0	0.1 = 0	0 ± 0
PO ₄ mg/l	0.4	0.3	0	1.0	0.1	0.02	0.2	0	. 0.1	0.2	0.02	0.2
Eh. mv	0.28	0.27	0.32	0.28	0.26	0.25	0.27	0.2จั	0.25	0.23	024	0.23
rH ₂	26.46	26.31	27.23	26.46	26.17	26.02	26.31	26.17	26.02	25.73	25.88	25.75
GPP mgC/Vh	0.3	0.3	0.4	0.4	0.4	0,3	0.3	03	0.3	0.2	0.2	ڌ.0
NPP mgC/l/h	0.12	-0.04	-0.17	0.03	0.09	0.05	0.09	0.13	-0.02	-0.16	-0.17	-0.05

hints

Table-1: Seasonal Variation of Physico-Chemical and Biological conditions in the study Pond-1, during the Study Period.

Parameters	Summer	Monsoon	Post-Monsoon	Winter
	March-May	June-Aug	Sept-Nov	Dec-Feb
Air Temp. C	32.4-38.3	29.1-33.6	23.1-30.3	17.5-23.8
Water Temp. C	29.0-31.8	28.9	24.1-30.4	19.3-23.9
Av. depth M	1.0-1.1	1.0-1.2	0.9-1.2	1.0-1.1
S D depth M	0.2-0.2	0.2-0.2	0.0-0.2	0.2-0.3
TDS mg/l	420.2-489.5	454.7-49.4	353.0-488.3	440.4-484.3
E. Cond. μ moh/cm	647-753	700-751	543-752	678-745
P^{H}	7.7-8.7	8.3-8.83	8.5-8.7	8.6-8.6
Free CO ₂ mg/1	21.5-34.7	2.8-10.1	23-26.7	15.7-33.2
CO ₃ Alk. mg/1	00	0-17.2	00	0-2.8
HCO ₃ Alk. mg/l	30.5-403.	21-49	114.1-163.3	184301.2
T. hardness mg/1	209-538	200-295	209-285	386-467
Chloride mg/1	112-131	103-143	99-109	87-107
DO mg/1	5.4-6.1	5.2-6.8	5.7-6.9	4-7.6
% Sat of 02	70-82	68-92	69-92	45-86
BOD ₅ mg/1	2.7-6.9	5.4-12	3.1-4.9	3.2-7
COD mg/l	7.4-15	11.9-22	7.2-10.6	9.8-15.2
NH ₃ -N mg/1	0-0.10	0-0.10	0.10-0.11	0.10-0.13
PO ₄ mg/1	0.1-0.7	0.1-0.2	0.1-0.2	0.1-0.7
Eh mv.	0.25-0.35	0.29	0.25-0.27	0.26
rH^2	26.2-27.8	26-26.6	25.8-26.3	26.1
GPP mg C/1/h	0.2-0.3	0.2-0.3	0.2-0.2	0.2-0.4
NPP mg C/1/h	0.08-0.10	-0.1-0.04	-0.11-0.04	-0.14-0.12
Phytoplankton uni/1	66044	82225	66123	57161
Zooplankton unit/1	91923	114981	313183	541917
Phyto. :Zoo.	1: 1.39	1: 1.39	1: 4.74	1: 10
'P' value	6.4	8.2	4.6	7.4

Table.2: Seasonal variation of Physico-chimical and Biological conditions in the study pond-2, during the study period.

Parameters	Summer	Monsoon	Post-Monsoon	Winter
1 6.0	March-May	June-Aug	Sept-Nov	Dec-Feb
Air Temp. C	32.6-38.7	28.6-33.6	23.3-30.3	17.5-23.8
Water Temp. C	31.8-32.8	29.7-32.1	23.1-31.0	18.4-24.0
Av. depth M	1.4-1.8	1.5-1.8	1.4-1.5	1.4-1.6
S D depth M	0.2-0.3	0.2	0.2	0.1-0.4
TDS mg/1	404.3-436.2	383.0-455	435-499	406.3-482.0
E. Cond.μ moh/cm	422-671	443-700	529-767	677.5-741
P ^H	7.8-8.9	8.2-8.4	8.4-8.7	8.7-8.9
Free CO2 mg/l	4.3-14.7	8.2-8.9	8.3-19	4.9-32.5
CO ₃ Alk. mg/1	0-15.2	0-4.8	0.4-2.8	0-10.8
HCO ₃ Alk. mg/l	21.2-356.7	18.7-41.3	109.7-160.5	171.4-290.1
T. hardness mg/l	285.5-609.2	226.2-307	214.2-317.7	414.6-487
Chloride mg/l	103.6-138.7	121.6-191.7	129.8-146.7	104.6-121.7
DO mg/1	6.0-8.4	6.1-7.6	6.8-8.4	5-7.9
% Sat of 02	83.0-114.4	80.0-105.0	81.0-113.0	53.2-86.5
BOD ₅ mg/l	2.5-3.7	2.3-9.6	3.9	8.7
COD mg/l	6.4-8.4	6.7-20.7	8.2-12.8	7.3-10.1
NH ₃ -N mg/1	0-0.1	0.1	0.111	0.10-0.12
PO ₄ mg/1	0-0.1	0.1	0.1-0.2	0.03-0.1
Eh mv.	0.22-0.35	0.28-0.31	0.25-0.28	0.23-0.25
rH ²	25.39-27.67	26.46-27.09	26-26.46	25.73-26.02
GPP mgC/1/h	0.2-0.3	0.3-0.4	0.3	0.1-0.4
NPP mgC/1/h	-0.07-0.02	-0.01-0.1	0.1-0.21	-0.16-0.15
Phytoplankton uni/l	107883	92858	145523	116341
Zooplankton unit/1	375451	255918	264824	442213
Phyto.: Zoo.	1: 3.48	1: 2.75	1: 1.82	1: 3.8
'P' value	4.3	7.6	4.4	5.4

Table-3 Seasonal Variation of Physico-Chemical and Biological conditions in the study Pond-3, during the study period.

Parameters	Summer	Monsoon	Post-Monsoon	Winter
	March-May	June-Aug	Sept-Nov	Dec-Feb
Air Temp. C	32.6-35.1	30.2-32.6	22.8-28.5	19-24.3
Water Temp. C	29.3-32.5	29.8-31.9	22.9-29.2	19-24.3
Av. depth M	1.8-2.2	2.1-2.3	1.7-2.0	1.7-1.9
S D depth M	0.2	0.2-0.3	0.2	0.1-0.3
TDS mg/1	397.2-445.9	381.6-419.3	3387.7-448.5	441.7-450.1
E. Cond. μ moh/cm	611686	587-645	597-690	680-693
P ^H	8.2-8.9	8.3-8.4	8.6	8.6-8.8
Free CO ₂ mg/1	1-14.6	0-4.3	2.9-6.3	0-6.4
CO ₃ Alk. mg/1	1.8-4.8	1.2-1.8	1.7-3.3	1.4-8
HCO ₃ Alk. mg/1	25.6-445.2	15.1-81.3	113.8-134.8	156.4-283.7
T. hardness mg/l	240.2-608.8	228.7-273	238.7-271.7	424-466.2
Chloride mg/l	132.4-156.7	103.4-142	107.5-132.1	178201.3
DO mg/1	7.1-8.3	7-7.7	5.9-9.6	5.7-9.1
% Sat of 02	93.4-114.2	95.5-101.9	70-126	69.2-101.2
BOD ₅ mg/1	9-11.6	. 9.4-13.6	6.7-12.8	3.2-7.6
COD mg/1	19.6-26.3	22.3-28.7	14.5-27.7	8.7-18.9
NH ₃ -N mg/1	0-0.10	0.1	0.10-0.11	0.1-0.13
PO ₄ mg/1	0.03-0.1	0.03-0.3	0.1-0.2	0.01-0.3
Eh mv.	.2331	.2728	0.26	.2326
rH ²	25.73-27	25.7-26.4	26.17	2526.1
GPP mgC/1/h	0.2-0.4	0.3-0.4	0.2-0.3	0.3
NPP mgC/1/h	0.03-0.10	-0.01-0.10	0.0211	-0.02-0.1
Phytoplankton uni/1	114396	72746	105668	83704
Zooplankton unit/l	424278	122378	233468	107953
Phyto.: Zoo.	1: 3.71	1: 1.68	1: 2.21	1: 1.28
'P' value	11.2	14	11.4	8.4

Table. 4 Seasonal Variation of Physico-Chemical and Biological conditions in the study Pond-4, during the study period.

Parameters	Summer	Monsoon	Post-Monsoon	Winter
	March-May	June-Aug	Sept-Nov	Dec-Feb
Air Temp. C	32.6-35.1	30.2-32.6	22.8-28.5	19-24.4
Water Temp. C	29.4-32.5	29.7-31.6	23.0-29.6	19.1-24.0
Av. depth M	1.6-1.9	1.6-2	1.4-1.5	1.4-1.7
S D depth M	0.2	0.2	0.2-0.3	0.1-0.2
TDS mg/l	351.3-378.6	354.9-403.3	370.8-404.6	397.5-470.6
E. Cond.μ moh/cm	541-583	546-621	571-623	612-724
P^H	8.4-8.9	8.1-8.5	8.5-8.6	8.6-8.9
Free CO ₂ mg/1	0-2.3	0	0-4.3	0-5.7
CO ₃ Alk. mg/l	4.9-6.6	2.7-5.8	1.8-4.8	4.3-7.7
HCO3 Alk. mg/l	21.3-366.6	15.2-78	102-130.4	133.5-253.3
T. hardness mg/l	263-469.3	197.8-278	216-232	317.3-372.2
Chloride mg/l	98.6-121.6	109.2-137	113.6-142	93.4-121
DO mg/1	7.1-8.2	6.3-6.7	6.5-10.6	7.9-8.2
% Sat of 02	97.4-113	83.7-91.0	77.5-140	88.0-100
BOD ₅ mg/1	8.6-13	13.2-16.8	12.4-16.8	9.3-13.6
COD mg/1	17.3-25.9	31.3-37.2	26.2-33.6	20.7-24.8
NH ₃ -N mg/1	0.0-0.10	0.10	0.1	0.10-0.12
PO ₄ mg/1	0.02-0.40	0.3-1.0	0.02-0.2	0.1-0.2
Eh mv.	0.23-0.28	0.27-0.32	0.25-0.27	0.23-0.26
rH^2	15.88-26.46	26.31-27.23	26.02-26.31	25.73-26.17
GPP mgC/1/h	0.2-0.3	0.3-0.4	0.3-0.4	0.2-0.3
NPP mgC/1/h	-0.05-0.12	0.04-0.17	0.05-0.09	0.02-0.16
Phytoplankton uni/1	85473	77776	85478	134057
Zooplankton unit/1	234049	208150	136820	136957
Phyto.: Zoo.	1:2.73	1: 2.67	1: 1.60	1: 1.02
'P' value	10.8	19.2	15.3	11.2

Tab.1. Abundance of Zooplankton: unit/l in P-1

Name of the Gen.	May-94	Jun	Jul	Aug	Sep	Oct	Nov	Dec
Copepoda					, , , , , , , , , , , , , , , , , , ,			
Cleptocamptus					2417		54563	
Cyclops	10698	15202		39023	27627	25555	114219	34792
Merocyclops						691	48001	
Mesocyclops	3626	2064		8806		1209	66390	4230
Diaptomus	7882	1877		14072	20288	9583	62904	7252
Paradiaptomas		-				604	19511	1554
Phyllodiaptomus				8104			13813	
Neodiaptomus						863	14936	
Total = 8 genera	22206	19143		70005	50332	38505	394337	47828
Cladocera								
Alona				6734		7770	6130	
Bosmina						1986	3885	
Daphnia	23836	29279	16921	62680	44116	28922	90305	17698
Diaphanosoma	1689	7695	7856	17439	11655	10187	47553	
Macrothrix						989	1727	
Moina	49549	1689	4748	16403	2762	7252	23310	2800
Sida	40165			1813			3367	2417
Total = 7 genera	115239	38663	29525	105069	58533	57106	176277	22915
Rotifera								
Asplanchinopus								
Brachionus	176615	62500	13295	3626	28835	12259	119787	15972
Chromogaster	48236							
Filinia								
Harringia						863		
Keratella	94220		950	1986	32116	7425	10915	7030
Manfredium						1813		
Notholca								
Philodina							2590	777
Pompholyx					2331			
Rotaria							1727	
Trichocerca								6825
Total = 12 gen.	319071	62500	14245	5612	63282	22360	129019	30604
Grand Total = 27								
gen.						L		

Tab.1. Abundance of Zooplankton: unit/l in P-1

Name of the Gen.	Jan-95	Feb	Mar	Apr	Total	Y.mean
Copepoda					, , , , , , , , , , , , , , , , , , , ,	
Cleptocamptus	6561	3540	950		68031	5669
Cyclops	106017	28317	6302	1295	409047	34087
Merocyclops	12432				61124	5094
Mesocyclops	12432				98752	8230
Diaptomus	17008	45066	6820		192752	16063
Paradiaptomas	7252	1468			30389	2533
Phyllodiaptomus	2849	1640			26406	2201
Neodiaptomus	3539	1813			21151	1763
Total = 8 genera	168090	81844	14072	1295	907657	75638
Cladocera						
Alona	3108	21497			45239	3770
Bosmina	2763				8634	720
Daphnia	76664	61901	20375	1640	474337	39528
Diaphanosoma	19252	25037	10705	863	159931	13328
Macrothrix	863				3579	298
Moina	2762	39109	2676		153060	12755
Sida	2417			1036	51215	7268
Total = 7 genera	107829	147544	33756	3539	895995	77217
Rotifera						
Asplanchinopus	4921				4921	410
Brachionus	133644		3281	8288	572102	47675
Chromogaster					48236	4020
Filinia	3194		950		4144	346
Harringia					863	72
Keratella	869635	863		5180	1030320	85860
Manfredium					1813	151
Notholca		863			863	72
Philodina					3367	281
Pompholyx	7252				9583	799
Rotaria					1727	144
Trichocerca	3971				10796	900
Total = 12 gen.	1022617	1726	4231	13468	1688735	140730
Grand Total						

Tab.2. Abundance of zooplankton Unit/l in P-2

Name of the Gen.	May-94	Jun-94	Jul-94	Aug-94	Sep-94	Oct-94	Nov-94	Dec-94
Copepoda								
Cleptocamptus	4504	1501					12518	
Cyclops	31532	29843	17267	17735	27281	17094	110334	27195
Merocyclops						8202	77787	
Mesocyclops			2417			6388	63541	690
Diaptomus	50025	18393	13362	12864	6389	9928	21497	1554
Heliodiaptomus								
Paradiaptomas			2504				8719	
Phyllodiaptomus						2072	6043	
Naodiaptomus						5094		
Neodiaptomus	5443					4230	2417	
Total = 10 genera	91504	49737	35550	30599	33670	53008	302856	29439
Cladocera								
Alona	22897						9669	1813
Bosmina						8374	12432	
Ceriodaphnia		1126	2245					
Daphnia	72447	79768	39454	32202	25468	18734	25382	135716
Diaphanosoma	42417	17642	11050	32116	5439	14331	4230	6561
Moina	9572	6757	5612	4058	5266	7079	14331	17353
Polyphemus							10964	
Sida	35660		1381	691		31598	6907	4921
Total = 8 genera	182993	105293	59742	69067	36173	80116	83915	166364
Rotifera								
Asplanchna	6569		3194					
Brachionus	130068	2815	120608	114145	29958	29699	61124	
Chromogaster	51239							
Cupelopagis				691				
Dorystoma			604	2590				
Filinia			13295	2244				
Harringia			604					
Keratella	61561	39720	57671	58189	45670	4662	4748	
Philodina					2245		2417	
Pompholyx				691	691	12432		
Scaridium					863			
Trichocera								
Tripleuchanis				691				
Total = 13 genera								
Grand Total = 31 gen								
Ji gen	<u> </u>							

Tab.2. Abundance of zooplankton Unit/l in P-2

Name of Species	Jan.95	Feb. '95	March, '95	April, '95	Total	Y.mean
Copepoda					10101	1
Cleptocamptus	6216	2331	4921	2331	34322	2860
Cyclops	32634	27972	27368	355176	721431	6012
Merocyclops	21065				107054	58921
Mesocyclops	21238	13123			107397	8950
Diaptomus	23655	22101	23990	17871	221629	18469
Heliodiaptomus	13036	7943			20979	2248
Paradiaptomus	8461	13986	604	1640	35914	2993
Phyllodiaptomus					8115	676
Naodiaptomus	5439	12259			22792	1899
Neodiaptomus	6302				18392	1533
Total = 10 gen	138046	99715	56883	377018	1298025	104560
Cladocora						
Alona	690				35069	2922
Bosmina	3885				24691	2058
Ceriodaphnia					3371	281
Daphnia	37382	6648	46361	5957	525519	43793
Diaphanosoma	16835	34188	20806		205615	13135
Monia	7597		6302	3194	87121	7260
Polyphemus	2590				13554	1129
Sida	1640			2417	85215	7101
Total	70619	40836	73469	11568	980155	77679
Rotifera						
Asplanchna	4317				14080	1173
Brachionius	115255		48433	20202	672307	56026
Chromogaster					57887	4824
Cupelopagis					691	58
Dorystoma					2590	266
Filinia					22791	1899
Harringia		-			604	50
Keratella	579383			14849	866453	72204
Philodina					4662	389
Pompholyx	26072				39886	3324
Scaridium					863	72
Trichocora	3281	•			3281	274
Tripleuchanis	0201				691	58
Total					140577	
Grand Total 31 gen						

Tab -3: Abundance of Zooplankton unit/l in Pond -3.

Name of Species	May,94	Jun, '94	July '04	A.10	Continu	0-1-10-1	.	D 104
Name of Obcolog	Way,04	ouri, 94	July, 94	Aug. '94	Sept'94	Oct. 94	Nov. '94	Dec. '94
Copepoda								
Cleptocamptus			8633	1036				
Cyclops	40728	18956	49296	23483	54570	43339	33152	27368
Mesocyclops	5067		3798					
Diaptomus	3566		41354			23224	10015	
Total = 4 gen	49361	18956	103081	24519	54570	66563	43167	27368
Cladocera								
Alona			96175					
Bosmina								
Daphnia	88401	10886	41526	24087	42735	30388	56462	24346
Diaphanosoma	108296		32548			63109	16136	
Moina	11824		8288	949				
Sida	7132	·	863					
Total = 6 gen	215653	10886	179400	25036	42735	93497	72598	24346
Rotifera								
Brachionus	99654	8671	90823	18130	32720	92808	17267	34102
Chromogaster	45608				1209			
Filinia			2676	1122	20806	863		
Keratella	82770	95909		29094	46447	175084	17008	42994
Notholca			10964	1036				
Polyarthra			4144		2245			
Rotaria								1036
Trichocerca						66477		1899
Total = 8 gen	228032	104580	108607	49382	103427	335232	34275	80031
Grand total = 18 gen								

Tab -3: Abundance of Zooplankton unit/l in Pond -3.

S Spacias	Jan. 95	Ech IOF	N4	4 11 15 -	T	
Name of Species	Jan. 93	Feb. '95	March, '95	April, '95	Total	Y. mean
Copepoda						
Cleptocamptus					9669	806
Cyclops	29612	37317	38504	14935	411260	34272
Mesocyclops					8865	739
Diaptomus	16231		7079		101469	8456
Total = 4 gen	45843	37317	45583	14935	531263	40673
Cladocera						
Alona					96175	8015
Bosmina		9410	2331		11741	978
Daphnia	30994	38246	29267	17526	434864	36239
Diaphanosoma	25037		22447	3194	270767	22564
Moina	7511			1468	30040	2503
Sida	3281		690		11966	997
Total = 6 gen	66823	47656	54735	22188	855553	71296
Rotifera						
Brachionus	39541		34533	23828	492077	41006
Chromogaster					46817	4902
Filinia					25467	2123
Keratella	31771		8633	4662	534372	44531
Notholca					12000	1000
Polyarthra					6389	533
Rotaria					1036	86
Trichocerca					68376	5698
Total = 8 gen	71312	0	43166	28490	1186534	99879
Grand total = 18 gen						

Tab - 4: Abundance of Zooplankton unit/1 Pond p-4.

Name of Species	May, '94	June'94	July,'94	Aug, '94	Sept, '94	Oct, '94	Nov, '94	Dec, '94
Rotifera				3,	2001, 24	OUI, 34	1000, 94	1000, 94
Ascomorphella	10885	6569	4403	6569	30389	17612	12518	2815
Brachionus	184683	5537	16231	19598	22792			+
Chromogaster	45230	2815	15022	4403	2158	160666 5784	12545	18475
Filinia	4880	3126	9505	1640	4040	8892	10225	3367
Harringia	40540	6180	22706	9722	42130	61297	2935	1554
Keratella	140203	2223	23542	13978	18216	15885	65404	16178
Notholca	1381	691	863	518	691	604	1554 863	5266 691
Philodina	25901	8921	4230	10221	28577	27540		
Platyias	691	863	604	604	518	691	12680 691	15108 518
Polyarthra	58745	1313	27798	24432	1899	25555	23396	42217
Pompholyx	39602	13513	45555	1381	28576	7856	4489	7079
Trichocerca	73565	15774	29009	7079	2935	23914	13900	
Total	626306	67525	199468	100145	182921	356296	161200	3367 116635

Tab - 4: Abundance of Zooplankton unit/1 Pond p-4.

Rotifera	Jan.95	Feb. '95	March, '95	April, '95	Total	Y.mean
Ascomorphella	9842	2815	19080	39713	163210	13600
Brachionus	2935	31501	84261	30217	589441	49120
Chromogaster	12691	604	23342	13036	138677	11556
Filinia	8892	4880	31000	2935	84279	7023
Harringia	16912	13727	65872	92031	452699	37725
Keratella	1381	5266	3453	1684	232651	19388
Notholca	660	863	1381	863	10013	835
Philodina	30314	12000	25555	65700	266747	22229
Platyias	. % 18	518	863	518	7942	660
Polyarthra	1381	10619	15799	24691	257845	21487
Pompholyx	6216	6648	14331	10015	185261	15438
Trichocerca	7079	20030	85736	3454	285842	23830
Total	99110	109471	370673	284857	2674607	222881

Tab. ZP, 1a. Seasonal abundance of different genara and groups of Zooplankton (unit/l) and seasonal percentage in parenthesis of Pond -1.

Zooplankton genera	Summer	Monsoon	Post-Monsoon	Winter
	March- May	Jun-Aug	Sept-Nov	Dec-Feb
Copepoda		<u> </u>		
Cleptocamptus	317		1899	3367
Cyclops	5198	18135	55800	56375
Merocyclops			14564	4144
Mesocyclops	1209	3623	22533	5554
Diaptomas	4901	5316	30925	23109
Paradiaptomas			6705	3425
Phyllodiaptomus		2701	4604	1496
Neodiaptomus			5266	1784
Total = 8				
Total seasonal	11625	29775	142296	974770
abundance				
	(12.64)	(25.89)	(45.43)	(17.98)
CLADOCERA	,			
Alona		2245	4633	8201
Bosmina			1957	921
Daphnia	15284	36293	54474	52088
Diaphanosona	4419	10997	23132	14763
Macro Thrix			905	322
Moina	17408	7613	11108	14890
Sida	13734	605	1123	1611
Total = 7 gen				
Total seasonal	50845	57753	97332	92796
abundance				
	(55.31)	(50.22)	(31.07)	(17.12)

Tab. ZP, 1a. Seasonal abundance of different genara and groups of Zooplankton (unit/l) and seasonal percentage in parenthesis of Pond-1.

Zooplankton genera	Summer	Monsoon	Post-Monsoon	Winter
	March- May	Jun-Aug	Sept-Nov	Dec-Feb
ROTIFERA			•	
Asplanchna				1641
Brachionus	9743	26474	53627	49872
Chromogaster	16079			
Filinia	317			1065
Harringia			288	
Keratella	3314	979	16819	292509
Manfredium			605	
Notholca				288
Philodina			863	259
Pompholyx			777	2418
Rotaria			576	
Trichocera				3599
Total = 12 gen.		_		
Total seasonal	29453	27453	73555	351651
abundance				
	(32.13)	(23.87)	(23.48)	(64.89)
Total Zooplankton	91923	114981	313183	541917

Tab. - ZP, 2a Seasonal abundance of different genera and groups of zooplankton (unit/l) and seasonal percentage in parenthesis of pond-2.

Class & genera of	Summer March-	Monsoon Jun-	Post-monsoon	Winter
Zooplankton	May	Aug	Sept-Nov	Dec-Feb
Copepoda				
Cleptocamptus	3919	501	7173	2849
Cyclops	138025	21615	51570	29267
Merocyclops			28663	7022
Mesocyclops		806	23400	26302
Diaptomas	30629	14873	12605	15770
Heliodiaptomus				6993
Paradiaptomus •	748	835	2906	7403
Phyllodiaptomu			2705	
Naodiaptomus			1698	5899
Neodiaptomus	1814		2216	2101
Total = 10 genera	175135	38630	132936	103606
Cladocera				
Alona	7632			
Bosmina			6935	1296
Ceriodaphnia		1124		
Daphnia	41588	50475	23195	59915
Diaphanosoma	21074	20269	8000	19195
Monia	6356	5476	8892	8317
Polyphemus			3655	863
Sida	12692	691	12835	2187
Total = 8 gn	89342	78035	63512	91773
	(23.7)	(30.5)	(23.9)	(20.7)
Rotifera			1 1	
Asplanchna	2190	1065		1439
Brachionus	66234	79189	40260	38418
Chromogaster	17080		2216	
Cupelopasis		231		
Dorystoma		1065		
Filinia		5180	1324	1094
Harringia		202		
Keratella	25470	51860	18360	193128
Philodina	25 (75		1554	
Pompholyx		230	4374	8691
Scaridium			288	
Trichorocera				1094
Tripleuchlamis		231		
Total =13 gn	110974	139253	68376	246834
	(29.5)	(54.4)	(25.8)	(55.8)
Total seasonal plankton	375451	255918	264824	442213

Tab. -ZP, 3a. Seasonal abundance of different genera and groups of zooplankton (unit/l) and seasonal percentage in parenthesis of pond-3.

Class & genera	Summer	Monsoon	Post-Monsoon	Winter
	March- May	June-Aug.	Sept-Nov.	Dec-F eb.
Copepoda			1	
Cleptocamptus		3223		
Cyclops	31389	30578	43687	31433
Mesocyclops	1689	1266		
Diaptomas	3548	13785	7741	8749
Total = 4 gen				
Total seasonal	36626	48852	51428	40182
abundance				
	(15.64)	(23.46)	(35.58)	(29.34)
Cladocera				
Alona		32058		
Bosmina	777			3137
Daphnia	45065	25500	43195	31195
Diaphanosoma	44646	10849	26415	8346
Moina	4431	3079		2504
Sida	2607	288		1094
Total = 6 gen				
Total seasonal	97526	71774	69610	46276
abundance				
	(41.68)	(34.48)	(50.87)	(33.78)
Rottfera				
Brachionus	52672	39208	47598	24548
Chromogaster	15203		403	
Filinia		1266	7223	
Keratella	32022	41668	79513	24922
Notholca		4000		
Polyarthra		1382	748	
Rotaria				346
Trichocerca			22159	633
Total = 8 gen				
Total seasonal	99897	87524	15764	50499
abundance				(0.6.05)
	(42.68)	(42.04)	(11.52)	(36.87)
Total Zooplankton	234049	208150	136820	136957

Tab.- ZP, 4a. Seasonal abundance of different genera and groups of zooplankton (unit/l) and seasonal percentage in parenthesis of pond-4.

Class	Summer March- May	Monsoon June-Aug.	Post-Monsoon Sept-Nov.	Winter Dec-Feb.
Rotifera			23917.011	
Ascomorphella	23226	5847	20173	5157
Brachionus	99720	13789	65334	17637
Chromogaster	27203	7413	6056	5554
Filinia	12938	4757	5289	5109
Harringia ,	63148	12869	56277	15606
Keratella	48447	13248	11885	3971
Notholca	1208	690	719	720
Philodina	39052	7790	22932	19141
Platyias	691	690	633	629
Polyartlura	33078	17848	16950	18072
Pompholyx	21316	20150	13640	6648
Trichocerca	54251	17287	13580	10159
Total Seasonal	424278	122378	233468	107953
Abundance				

Tab. ZP, 1b. Seasonal abundance of different groups of Zooplankton and total abundance (unit/l) in Pond-1.

	Summer	Monsoon	Post-Monsoon	Winter
Zooplankton Groups	March-may	June-Aug	Sept-Nov	Dec-Feb
Copepoda	11625	29775	142296	97470
оргроми	(12.65)	(25.9)	(45.44)	(17.99)
Cladocera	50845	57753	97332	92796
	(55.31)	(50.23)	(31.08)	(17.12)
Rotifera	29453	27453	73555	351651
	(32.04)	(23.88)	(23.49)	(64.9)
Total Zooplankton	91923	114981	313183	541917

Tab. ZP, 2b. Seasonal abundance of different groups of Zooplankton and total abundance (unit/l) in Pond-2.

	Summer	Monsoon	Post-monsoon	Winter
Zooplankton Groups	March-may	Jun-Aug	Sept-Nov	Dec-Feb
Copepoda	175135	38630	132936	103606
	(46.6)	(15.1)	(46.8)	(23.4)
Cladocera	89342	78035	63512	91773
,	(23.7)	(30.5)	(23.9)	(20.8)
Rotifera	110974	139253	68376	246834
	(29.5)	(54.4)	(25.8)	(55.8)
Total Zooplankton	375451	255918	264824	442213

Tab.ZP, 3b. Seasonal abundance of different groups of Zooplankton and total abundance (unit/l) in Pond-3.

Zooplankton Groups	Summer	Monsoon	Post-Monsoon	Winter
	March- May	Jun-Aug	Sept-Nov	Dec-Feb
Copepoda	36626	48852	51928	40182
	(15.65)	(23.47)	(37.59)	(29.34)
Cladocera	97526	71774	69610	46276
	(41.67)	(34.48)	(50.88)	(33.79)
Rotifar	99897	87524	15764	50499
	(42.68)	(42.05)	(11.52)	(36.87)
Total Zooplankton	234049	208150	136820	136957

Tab. ZP, 4b. Seasonal abundance of different groups of Zooplankton and total abundance (unit/l) in Pond-4.

Zooplankton group[Summer	Monsoon	Post-Monsoon	Winter
	March- May	Jun-Aug	Sept-Nov	Dec-Feb
Copepoda	_	_	-	
Cladocera	_	-	_	
Rotifera	424278	122378	233468	107953
Total Zooplankton	424278	122378	233468	107953

Tab. Zp. 5a. Total Number of genara of different groups of Zooplankton and their percentage in parenthesis in four study ponds.

Class/Groups	Pond -1	Pond -2	Pond -3	Pond -4
Copepoda	8	10	4	-
	(29.6)	(32.2)	(22.2)	
Cladocera	7	8	6	-
	(25.9)	(25.8)	(33.3)	
Rotifera	12	13	8	12
	(44.4)	(41.9)	(44.4)	(100)
Total No of	27	31	18	12
Genera				

Tab. Zp-2, 5b. Occurrence of different groups of Zooplankton in Four Study Ponds.

Group & Genera	Pond-1	Pond-2	Pond-3	Pond-4
COPEPODA				
Cleptocamptus	+	+	+	-
Cyclops	+	+	+	-
Merocyclops	+	+		
Mesocyclops	+	+	+	-
Diaptomus	+	-	+ +	<u> </u>
Heliodiaptomus	-	+		
Paradiaptomus	+	+	-	
Phyllodiaptomus	+	+		
Naodiaptomus		+		-
Neodiaptomus		+		
10 gen.		+		
Clodocera		- 	-	-
Alona	+			
Bosmina		+	+	
		+	+	-
Ceriodaphnia		+		
Daphnia	+	+	+	-
Diaphanosoma	+	+	+	
Moina	+	+	+	-
Polyphamus		+		-
Sida	+	+	+	-
Macrothrix	+		-	
Total = 9 gen.				
Rotifera				
Ascomorphella		-	-	+
Brachionus	+	+	+	+
Chromogaster	+	+	+	+
Filinia	+	+	+	+
Harringia	+	+	+	+
Keratella	+	+	+	+
Asplanchna	+	+	-	-
Notholca	+	_	+	+
Philodina	+	+	-	+
Platias		-	-	+
Polyarthra			+	+
Pompholyx	+	+	-	+
Trichocera	+	+	+	+
	<u> </u>	+	-	-
Cupelopagis Donustoma	<u> </u>	+		-
Dorystoma Socialium		+		-
Scaridium		+		-
Tripleuchlanis	+	- - 	+	-
Rotaria	+		-	
Manfredium		- 		-
Total = 19 gen.				
Grand Total = 38 gen.	+ Present,			

Tab.1: Abundance of Phytoplankton unit/l in P-1 (Pond-1).

•			,	,				Dag
Name of the Genera.	May-94	June	July	Aug.	Sept.	Oct.	Nov.	Dec.
Chlorophyceae								012
Chlorella	1689	1895	3095	4317	8288	6116	1310	812
celenastrum	3166	1800	1,123	1210	620	466	318	968
Ankistrodesmus	9816	12178	8004	6617	5333	6267	1322	4281
Crncigenia	1689	915	764	712	345	590	1210	1666
Scenedesmus	1332	1210	1333	936	855	910	1236	1810
Pediastrum	1877	2110	1318	966	688	690	775	495
Actinastrum	1701.	1210	995	2590	683	1124	647	1124
Closterium	1200	900	1123	672	792	813	775	490
Cosmarium	678	666	815	332	466	540	1120	1470
Staurastrum	1600	875	227	590	417	224		<u> </u>
Ulothrix	378	283			210	190		
Spirogyara	289	300	187					389
Total Gen = 12	25415	24342	18984	18942	18697	17930	8713	13505
Cyanophyceae								
Microcystis	15390	5686	2763	8029	1813	9410	7710	8892
Merismopidia	2081	1828	2334	1667	1210	1363	836	775
Spirulina			2590	863				
Oscillatoria	3881	9384	3566	2104	1875	1124	5122	1890
Anabaena	4330	3677	4500	2600	1360	1446	768	813
Anabaenopsis					816	790	236	285
Cylindrospermum							436	444
Total gen.=7	25682	20575	15753	15263	7074	14133	15108	13099
Bacillariophyceae								
Brebissonia				1890			2555	
Fragilaria	330			4400	1139	1519	985	817
Eunotia	2330	1420	1124	876	936	2311	2014	1800
Gyrosigma	1390	931	875	668	333	614	214	246
Navicula	4008	3667	2100	1790	4666	7892	2336	1818
Pinnularia	1210	890	718	310	2871	1800	1314	1714
Gomphonema			818	790	416			
Tabellaria	1686				910	1208	1366	
Cymbella	510	770	331	910				
Nitzschia	280	260	240	566	718	2118	2310	1800
Total gen = 10	11744	7938	6206	12200	11989	17462	13094	8195
Euglenophyceae								
Euglena	26504	51368	37641	15022	3367	25343	44286	9561
Phacus	812	623	210					
Trachelomonas	1227	1330	890	701				
Total gene = 3	28543	53321	38741	15723	3367	25343	44286	9561
Dinophyceae	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,							
Ceratium	770	886	1210	618	233			
	1	.,50				·		

Tab.1: Abundance of Phytoplankton unit/l in P-1 (Pond-1).

Name of the Genera.	Janu, '95	Feb.	March	April	Yearly mean
Chlorophyceae					
Chlorella	1210	978	461	795	2581
Scienastrum	612	318	789	446	986
Ankistrodesmus	3761	4754	1541	6129	5852
Crucigenia	985	120	310	818	844
Scenedesmus	1444	1232	1610	1012	1244
Pediastrum	666	365	485	933	947
Actinastrum	1072	1813	768	433	1180
Closterium	314	1440	795	718	836
Cosmarium .	1690	664	888	714	837
Staurastrum		444	614	890	490
Ulothrix		333	266	246	159
Spirogyara	835	795	444	416	305
Total gen. = 12	12589	13256	8971	13550	16242
Cyanophyceae					
Microcystis	13123	8374	7079	8479	8104
Merismopidia	886	1126	675	885	1214
Spirulina	9497				1079
Oscillatoria	3123	4826	1372	1494	3313
Anabaena	945	1224	886	1366	1993
Anabaenopsis			667	208	250
Cylindrospermum	367	120			114
Total gen.= 7	27941	15670	10679	12932	16159
Bacillariophyceae					
Brebissonia	3392	3436			939
Fragilaria	1210	444			903
Eunotia	766	995	1667	1390	1469
Gyrosigma	229	1780	1690	4890	1155
Navicula	1736	2536	1123	2114	2982
Pinnularia	1510	1430	910	818	1291
Gomphonema			266	686	248
Tabellaria		818	610	455	588
Cymbella			188	790	291
Nitzschia					691
Total gen = 10	8843	11440	6454	11313	10557
Euglenophyceae					
Euglena	6553	30716	10267	27134	23980
Phacus	921	679	395	780	368
Trachelomonas		816	512	771	520
Total gene=3	7474	32211	11174	28685	24868
Dinophyceae					
Ceratium	!	390	1120	1223	537

Tab.2: Abndance of Phytoplankton unit/l in P-2 (Pond-2).

Name of the Genera.	May-'94	June	July	Aug.	Sept	Oct.	Nov.	Dec.
Chlorophyceae								
Ankistrodesmus	11732	9339	8778	4495	1923	2014	2168	
Scenedesmus	3215	2890	7115	3229	4983	4185	3108	7173
Pediastrum	976	695	837	366	366	604	812	927
Crucigenia	1192	1638	2158	1485	3185	1890	2127	7201
Gloeotaenium	366						712	927
Actinastrum	1816	1632	11639	13732	25692	24310		18310
Closterium	1378	1112	908	896	712	1195	2329	6043
Cosmarium	6667	1833	2401	3192	886	366	612	1606
Micrasterias					817	1837		
Total gen. = 9	27342	19139	33836	27395	38564	36401	51581	42187
Cyanophyceae					•			
Microcystis	13138	14557	6561	3022			3022	6734
Anabaenopsis	1212	938	1038					2725
Anabaena	2181	3189	3146	3684	1176	895	1640	1492
Spirulina	1000	1580	966					1184
Oscillatoria	13382	11298	7479	7833	6804	6185	28749	17621
Lyngbeya	666	892	766	366	366			
Cylindrospermum	1220	1318	4932					
Total gen.=7	32799	33772	24888	14905	8346	7080	33411	29756
Bacillariophyceae								
Fragilaria	3352	5133	8270	1390	8876	11867	4144	3985
Brebissonia			1832	9345	6133	5977	7770	1392
Gomophonema	1501					3986	6826	4826
Navicula	8390	6381	1780	1923	9395	10890	9332	11483
Pinnularia	4500	3550	4813	4453	11332	12690	6795	5833
Cymbella	778	1223	1190	3816	4918	10082	18389	4372
Total gen. = 6	18521	16287	17885	15927	40654	55492	53256	31891
Euglenophyceae								
Euglena	68994	15415	24603	27368	59584	22015	19851	2899
Phacus	7161	1312	932				824	792
Trachelomonas							1392	1182
Total gen.= 3	76155	17227	25535	27368	59584	22015	22067	4873
Dinophyceae								
Ceratium						1297	6775	3998

Tab. 2: Abundance of PhytoPlankton Unit/l in P-2 (Pond-2).

Name of the Genera.	Janu.'95	Feb.	March	April.	Yearly mean
Chlorophyceae					
Ankistrodesmus			13818	17229	5958
Scenedesmus	6336	920	865	692	3726
Pediastrum	4230	1132	1307	1640	1158
Crucigenia	3083	9211	4371	2114	3305
Gloeotaenium	732	612			279
Actinastrum					11404
Closterium	18178	34188	9116	9876	6994
Cosmarium	2300	2481	3115	1930	2282
Micrasterias			865	3540	588
Total gen. = 9	34859	48544	33457	37021	35858
Cyanophyceae					
Microcystis	13278	9324	3885	5784	6609
Anabaenopsis	1390	890	1080	1718	916
Anabaena	1785	1295	2935	4948	2364
Spirulina	936				472
Oscillatoria	11309	19770	9678	7849	12330
Lyngbeya		1823	1327	1092	608
Cylindrospermum		712	366	895	787
Total gen.=7	28698	33814	19271	22286	24086
Bacillariophyceae					
Fragilaria	6046	10032	13097	1130	6444
Brebissonia		•		1480	2411
Gomophonema	7894			1392	2202
Navicula	17864	2538	2749	1813	7045
Pinnularia	4837	4136	5449	6795	6265
Cymbella					373
Total gen.= 6	36641	16706	21295	12610	28098
Euglenophycene					
Euglena	8229	5617	12820	11397	23274
Phacus	368	368			980
Trachelomonas	1294	889	914	368	503
Total gen. = 3	9891	6874	13734	11765	24757
Dinophyceae					
Ceratium	11390	17865			3444

Tab. 3: Abundance of Phytoplankton unit/l in P-3 (Pond-3).

L. MA AT LITE GENELAG	May'94	June	July	Aug.	Sept.	Oct.	Nov.	Dec.
Name of the Genera. Chlorophyceae					-			
Eudorina	10442					1		
Scenedesmus	1566	8310	7762	6406	2465	3145	6912	
Pediastrum	1850	3210	3406			 		
Chlorella	600	1838	1766	950				
Tetraedron		 		1	· · · · · · · · · · · · · · · · · · ·			666
Tetraedron Selenastrum		 	<u> </u>			 		1478
Ankistrodesmus	1124	7821	6183	 -		-	4485	
Ankistrodeshida	850	666	1310	950	1590	1006	986	
Crucigenia		- 000	1310	7.50	1370	1000	1	950
Staurastrum	1350	1145	1090		 	 		
Euastrum	1330	1143	1030	2124	2208	1376	850	
Cosmarium .	17782	22990	21517	10430	6263	5527	13232	3094
Total gen =11	1//62	22990	21317	10430	0203	3321	13232	3074
Cyanophyceae	6757	12120	5071	2762		5057	3194	8029
Microcystis	1010	13138	5871	2763	· · · · · · · · · · · · · · · · · · ·	5957	3194	1535
Spirulina	62006	1 1 1 7 0	1136	2222	10221	12550	20203	
Oscillatoria	53885	14170	31648	27273	48331	13770	20293	37883
Anabaena		<u></u>				9931	9928	1034
Nodularia		ļ <u>-</u>				2310	3972	
Arthrospira						850		
Cylindrospermum	1310	818	1375	2050	2050	2148	850	
Total gen.= 7	61952	28126	40030	32426	50381	34966	38237	48481
Bacillariophyceae			'					
Navicula	3810	2390	2270	2180	5886	4395	10964	11735
Fragilaria	10308	11395	14812	1460	1095	4890	7810	18315
Synedra			,		3810	2306	2018	1986
Gyrosigma			4836					
Eunotia	1550	1780			,	2310		
Achnanthes	980	1175	666					
Cynobella	2850					880	1776	1322
Nitzschia				8390				1392
Tabellaria	950					666	950	1234
Pinnularia	3245	2817	2146	4832	6790	11850	13210	9332
Total genera =10	23693	19557	24730	16862	17581	27297	36728	45316
Euglenophycae								
Euglena	12790				30130	 	53527	
Phacus	3300	2700			2-120	1800	1327	850
Total gen.= 2	16090	2700			30130	1800	54854	850
Grand Total	100,0	2,00	<u> </u>		30130	1000	24024	- 050

Tab. 3: Abundance of Phytoplankton unit/l in P-3 (Pond-3).

Name of the Genera.	Jan- '95	Feb.	March	April	Yearly mean
Chlorophyceae					
Eudorina	885	676	950	14763	2310
Scenedesmus			2185	950	3309
Pediastrum		846	950	1554	985
Chlorella	2718	4168	2590		1219
Tetraedron	950	890	333	850	308
Selenastrum	1710	1500	666		446
Ankistrodesmus				1950	1797
Crucigenia			1785	1986	911
Staurastrum	666			2440	338
Euastrum			1636	1246	539
Cosmarium				4508	922
Total gen. = 11	6929	8080	11095	30247	13098
Cyanophyceae					
Microcystis	3453			5698	4564
Spirulina			1150	1987	484
Oscillatoria	12742	3 9109	32030	19850	29249
Anabaena	2378	666	850	950	2145
Nodularia				950	603
Arthrospira	1000	1500	4985	7890	1352
Cylindrospermum				850	983
Total gen.= 7	19573	41275	39015	38175	39347
Bacillariophyceae	,				
Navicula	6495	950	1227	1590	4491
Fragilaria	12795	4667	6575	5780	8325
Synedra	1350				956
Gyrosigma	3880 .	7310	6660	2476	2097
Eunotia		4876	6870	2470	1656
Achnanthes	790	936	1092	1277	577
Cynobella	975	1646			788
Nitzschia	1485	950	2185	4322	1561
Tabellaria					317
Pinnularia	8865	7800	2480	4895	6522
Total genera = 10	35420	37685	27089	25810	27290
Euglenophycae					
Euglena		13123	29267	25986	13736
Phacus	850				903
Total gen. = 2	850	13123	29267	25986	14639
Grand Total					

Tab. 4: Abundance of Phytoplankton unit/l in P-4 (Pond-4).

Name of the Genera.	May- '94	June	July	Aug.	Sept.	Oct.	Nov.	Dec.
Chlorophyceae								
Chlorella			-	604	1732	1975	5890	6780
Actinastrum	-	-	-	-	-	2681	3906	4181
Crucigenia .	1018	950	855	889	1276	1236	1390	1780
Ankistrodesmus	6732	8310	2243	1509	8501	4361	7310	8333
Sclenastrum	-	-	-	-	-	3681	3894	4754
Scenedesmus	1333	1810	1897	1776	1605	3110	4140	1308
Padiastrum			-	-	-	1468	-	-
Staurastrum	-	-	· -	1830	-	-	-	-
Closterium	832		-	-	-	-	-	1127
Cosmarium	-	-	-	-	-	678	1814	998
Micrasterias	1236	_		-	-	-	-	-
Total gen= 11	11151	11070	4995	6608	13114	19180	28344	29261
Cyanophyceae		_	-	-	-	-	-	-
Microcystis	14264	9197	5957	-	-	_	9425	4748
Spiriluna	-	-	-	667	2334	9667	9000	2834
Oscillatoria	31193	1332	43458	17854	17637	13658	28045	23952
Lyngbeya	1727	1923	1332	668	1285	-	-	-
Anabaena	4330	3838	4338	2660	1308	978	-	-

Tab. 4: Abundance of Phytoplankton unit/l in P-4 (Pond-4).

Name of the Genera.	Jan-'95	Feb.	March	April	Yearly mean
Chlorophyceae					
Chlorella	9332	11431	-	-	3145
Actinastrum	1792	666	981	885	1258
Crucigenia .	2636	2520	1930	1618	1508
Ankistrodesmus	2184	1462	1797	1578	4527
Sclenastrum	1506	956	636	333	1313
Scenedesmus	- 1	-	-	691	1473
Padiastrum	-	8978	850	675	998
Staurastrum	-	1446	732	333	362
Closterium	936	1004	813	2115	569
Cosmarium	676	333	966	1276	562
Micrasterias	2681	-	-	966	407
Total gen= 11	21743	28797	8705	10470	16122
Cyanophyceae	-,	-	-	-	-
Microcystis	5526	4034	5957	12792	5992
Spiriluna	1676	7334	-	-	2793
Oscillatoria	36260	7580	9785	8847	9967
Lyngbeya	-	2327	2729	1732	1144
Anabaena	-	-	8310	7898	2805

Tab. 4: Abundance of Phytolankton unit/l in P-4 (Pond-4).

Name of the Genera.	May-94.	June	July	Aug.	Sept.	Oct.	Nov.	Dec.
Glaeotricha	-	•	-		***************************************	879	1804	2228
Cylindrospermum	-	-	-	2393				1032
Nodularia	961	-	-					813
Total gen.=8	52475	16290	55085	24242	22564	25182	48274	35607
Baciliariophyceae								
Rhopalodia	980	-	-				5612	6760
Navicula	1895	-	-		6776	6500	5525	13468
Pinnularia	2650	1628	1923	2122	1800	2346	8676	11398
Synedra	6895	11365	-				17114	21480
Gyrosigma	650	813	333	1669	1318			
Eunotia	950	1150	780	666				7876
Rhizosolenia	-	-	-				2406	1612
Melosira	1340	2185	1790	1827	980			486
Achnanthes	-	-	-					
Total gen.= 9	15360	17141	4826	6284	10874	8846	39333	63080
Euglenophyceae								
Euglena	39977	41923	25900	18953	39749			
Phacus	660	-	-			950		
Total gen. =2	40637	41923	25900	18953	39749	950		

Continued

Tab. 4: Abundance of Phytolankton unit/l in P-4 (Pond-4).

Name of the Genera.	Janu.	Feb.	March	April	Y. mean
Glaeotricha	827	1310	1004		671
Cylindrospermum	978	1013	669	723	567
Nodularia	666	333	1826	1346	495
Total gen.=8	45933	23931	30280	33338	34334
Bacillariophyceae					
Rhopalodia	4895	1736	2145		1844
Navicula	14770			2335	4273
Pinnularia	11340	17118	950	1156	5259
Synedra	11333	14680			6906
Gyrosigma		922	1105	776	632
Eunotia	9310	11400	12140	850	3760
Rhizosolenia				666	390
Melosira	1318	2766			1058
Achnanthes			878	1610	207
Total gen.= 9	52966	48622	17218	7393	24365
Euglenophyceae					
Euglena		48865	14884	12774	20688
Phacus		1255	956	1410	436
Total gen. =2		50120	15840	14184	

Tab.1a: Seasonal Abundance of different genera of Phytoplankton (unit/l) in pond-1.

Name of the Gen.	Summer	monsoon	Post -monsoon	Winter
Chlorophyceae	March - May	June -Aug.	SeptNov.	DecFeb.
Chlorella	979	3102	5238	1000
Sclenastrum	1467	1378	468	633
Ankistrodesmus	5829	8933	4307	4265
Crucigenia	939	799	715	924
Scenedesmum	1318	1156	1010	1495
Pediastrum	1098	1465	718	509
Actinastrum	968	1599	818	1336
Closterium	905	898	794	748
Cosmarium	761	605	709	1275
Staurastrum	1035	564	214	148
Ulothrix	297	95	133	112
Spirogyara	383	162		673
Total Gen = 12	15979	20756	15124	13118
Cyanophyceae		20,30	15127	
Microcystis	10483	4593	6311	10130
Merismopidia	1193	1943	1136	929
Spirulina	1123	1151	1130	3166
Oscillatoria	2249	5018	2707	2380
Anabaena	2194	3592	1192	994
Anabaenopsis	292	3392	614	95
	292		146	310
Cylindrospermum	16411	16207		18004
Total gen.=7	16411	16297	12106	18004
Bacillariophyceae		(20	0.50	2276
Brebissonia	110	630	852	2276
Fragilaria	110	1467	1214	824
Eunotia	1796	1140	1754	1187
Gyrosigma	2657	825	387	752
Navicula	2415	2519	4965	2030
Pinnularia	979	639	1995	1552
Gomphonema	317	536	139	
Tebellaria	917		1661	273
Cymbella	496	670		
Nitzschia	94	355	1476	600
Total gen. = 10	9781	8782	14443	9494
Euglenophyceae				
Euglena	21302	34677	24332	15610
Phacus	662	278		533
Trachelomonas	871	530		272
Total gene.= 3	22835	35485	24332	16415
Dinophyceae				
Ceratium	1038	905	78	130
Total gen. =1	1038	905	78	130
T. S. Abun of Phyto	66044	82225	66123	57161

Tab. 2a: Seasonal Abundance of different genera of Phytoplankton (unit/l) in pond-2.

Genera of Phytoplankton	Summer	Monsoon	Post- monsoon	Writer
General	March-May	June-Aug.	SeptNov.	DecFeb.
Chlorophyceae				
Ankistrodesmuz	14260	7537	2035	
Scenedesmus	1591	4113	4092	4810
Pediastrum	1308	633	594	2096
Crucigenia	2558	1760	2401	6498
Gloeotaenium	122		238	757
Actinastrum	605	9001	29905	6103
Closterium	6790	972	1412	19470
Cosmarium	3904	2475	615	2129
Micrasterias	1468		885	
Total gen. = 9	9 Gen	7 Gen	9 Gen	7 Gen
T. abun. of Chloro.	32606	26491	42177	41863
Cyanophyceae				
Microcystis	6702	8047	1007	5779
Anabaenopsis	1370	659		1669
Anabaena	3355	3340	1237	1524
Spirulina	334	849		1668
Oscillatoria	10303	8870	13913	16234
Lyngbeya	1028	675	122	608
Cylindrospermum	827	2084		274
Total gen. = 7	7 Gen	7 Gen	4 Gen	7 Gen
T. abun. of Cyano.	23919	24524	16279	27754
Bacillariophycae				
Fragilaria	5860	4931	8296	6688
Brebissonia	494	3726	6627	464
Gomphonema	965		3604	4238
Navicula	4317	3361	9873	10643
Pinnularia	5578	4272	10272	4934
Cymbella	259	2076	11130	1458
Total gen. = 6	6 Gen	5 Gen	6 Gen	6 Gen
T. abun. of Bacillario.	17473	18366	49802	28425
Euglenophyceae				
Euglena	31071	22629	33817	5582
Phacus	2387	748	275	510
Trachelomonas	427		464	1122
Total gen. = 3	3 Gen	2 Gen	3 Gen	3 Gen
T. abun. of Eugleno.	33885	23377	34556	7214
Dinophyceae				
Ceratium			2691	11085
Total gen. = 1	0	0	1	1
T. abun. of Dino.			2691	11085
T. S. Abun. of Phyto.	107883	92858	145523	116341

Tab. 3a: Seasonal Abundance of different genera of phytoplankton (unit/1) in pond-3.

Genera	Summer	Monsoon	Post- monsoon	Writer
3611075	March-May	June-Aug.	SeptNov.	DecFeb.
Chlorophyceae		<u>-</u>	· · · · · · · · · · · · · · · · · · ·	
Eudorina	8712			521
Scenedesmus	1567	7493	4174	
Pediastrum	1452	2206		
Chlorella	1063	1518		2296
Tetraedron	394			836
Sclenastrum	222			1563
Ankistrodesmus	1025	4668	1495	
Cruscigenia	1541	975	1194	
Staurastrum	813			539
Euastrum	1411	364		
Cosmarium	, 1503	708	1478	
Total gen. = 11	11	7	4	5
r.S. abun. of Chloro.	19703	17932	8341	5755
Cyanophyceae				
Microcystis	4152	7258	3051	3828
Spirulina	1046	379		512
Oscillatoria	35255	24364	27465	29913
Anabaenopsis	600		6618	1359
Nodularia	317		2094	
Arthrospira	4292		283	334
Cylindrospermum	720	1528	1683	
Total gen. = 7	7	4	6	5
T.S. abun. of Cyano	26382	33529	41194	35946
Bacillariophyceae	20002	00020	1,110	
Navicula	2209	2280	7082	6394
Fragilaria	7554	9223	4598	11926
Synedra	1001	1 22	2713	1112
Gyrosigma	3045	1612		3730
Eunotia	3630	594	770	1626
Achnanthes	1116	614		574
Cymbella	950		885	1315
Vitzchia	2169	2797		1276
l'abellaria	317		539	412
Pinnularia	3540	3265	10618	8663
Total gen. : 10	9	7	7	10
r. S. abun. of Bacillario	24530	20385	27205	37028
Euglenophyceae				
Euglena	22681		27886	4375
hacus	1100	900	1042	600
Total gen. = 2	2	1	2.	2
T. S. Phyto.	23781	900	28928	4975
Grand Total	114396	72746	105668	83704

7ab. 4a: Seasonal Abundance of different genera of phytoplankton (unit/1) in pond-4.

Genera of	Summer	Monsoon	Post-Monsoon	Writer
different classes	March-May	June-Aug.	Septnov.	DecFeb
Chlorophyceae				
Chlorella		202	3199	9182
Actinastrum	622		2196	2213
Crucigenia	1522	898	1301	2312
Ankistrodesmus	3369	4021	6724	3993
Sclenastrum	111		2525	2406
Scenedesmus	675	1828	2952	436
Pediastrum	508		490	2993
Staurastrum	355	610		482
Closterium	1254			1023
Cosmarium	747		831	669
Micrasterias	734			894
Total gen. = 11	10	5	8	11
T. S. abun. Chloro	9897	7559	20218	26603
Cyanophyceae	***************************************			
Microcystis	11004	5052	3142	4769
Spirulina		224	7001	3948
Oscillatoria	16608	20882	19780	22598
Lyngbeya	2063	1308	429	776
Anabaena	6846	3612	762	
Gloeotrichia	335		894	1455
Cylindrospermum	464	798		1008
Nodularia	1378			604
Total gen. = 8	7	6	6	7
T. S. abun. Cyano.	38698	31876	32008	35158
Bacillariophyceae				-
Rhopalodia	1042		1871	4462
Navicula	1410		6267	9113
Pinnularia	1585	1891	4274	13286
Synedra	2298	3788	5705	15831
Gyrosigma	844	938	439	308
Eunotia	4647	865		9529
Rhizosolenia	222		802	9529
Melosira	447	1934	327	538
Achnanthes	829			2523
lotal gen. = 9	9	5	7	9
T. S. abun. Bacillario.	13324	9416	19685	55590
Euglenophyceae				
Euglena	22545	28925	13250	16288
Phacus	1009		317	418
Total gen. = 2	2	1	2	2
T. S.abun. Eugleno	23554	28925	13567	16706
T. S. abun. of Phyto	85473	77776	85478	134057

Tab.1b: Seasonal abundance of different groups of phytoplankton in Pond-1 (unit/1). (P.C. composition in parenthesis).

Classes of	Summer	Monsoon	Post -monsoon	Winter
Phytoplankton	March-May	June-Aug.	SepNov.	DecFeb.
Chlorophyceae	17979	20756	15124	13118
	(27.2)	(25.2)	(22.9)	(22.9)
Cyanophyceae	16411	16297	12106	18004
	(24.8)	(19.8)	(18.3)	(31.5)
Bacillariophyceae	9781	8782	14443	9494
	(14.8)	(10.7)	(21.8)	(16.6)
Euglenophyceae	22835	35485	24332	16415
	(34.6)	(43.2)	(36.8)	(28.7)
Dinophyceae	1038	905	78	130
	(1.57)	(0.1)	(0.1)	(0.2)
T. S. abun. of Phyto.	66044	82225	66123	57161

Tab.2b: Seasonal abundance of different groups of phytoplankton in Pond -2 (unit/1). (P.C. composition in parenthesis).

Classes of	Summer	Monsoon	Post-monsoon	Writer	
Phytoplankton	March-	June-	SeptNov.	DecFeb.	
	May	Aug.	_		
Chlorophyceae	32606	26491	42177	41863	
	(30.2)	(28.5)	(28.9)	(35.9)	
Cyanophyceae	23919	24524	16279	27754	
	(22.2)	(26.4)	(11.2)	(23.6)	
Bacillariophyceae	17473	18366	49802	28425	
	(16.2)	(19.7)	(34.2)	(24.4)	
Euglenophyceae	33885	23377	34556	7214	
	(31.4)	(25.2)	(23.7)	(6.2)	
Dinophyceae	-	-	2691	11085	
			(1.8)	(9.5)	
T. S. abun. of Phyto.	107883	92858	145523	116341	

Tab.3b: Seasonal abundance of different groups of phytoplankton in Pond -3 (unit/1). (P.C. composition in parenthesis).

Classes of	Summer	Monsoon	Post -monsoon	Writer
Phytoplankton	March - May	June - Aug.	Sept Nov	Dec Feb.
Chlorophyceac	19703	17932	8341	5755
,	(17.2)	(24.7)	(7.9)	(6.9)
Cyanophyceae	46382	33529	41194	35446
	(40.5)	(46.1)	(38.9)	(42.3)
Bacillariophyceae	24530	20385	27205	37208
	(21.4)	(28.0)	(25.7)	(44.4)
Euglenophyceae	23781	900	28928	4975
	(20.8)	(1.2)	(27.3)	(5.9)
Dinophyceae	-	-	-	-
T. S. abun. of Phyto.	114396	72746	105668	83704

Tab.4b: Seasonal abundance of different groups of phytoplankton in Pond -4 (unit/1). (P.C. composition in parenthesis).

Classes of	Summer	Monsoon	Post-Monsoon	Writer
Phytoplankton	March-May	June-Aug.	SeptNov	DecFeb
Chlorophyceae	9897	7559	20218	26603
	(11.5)	(9.7)	(23.6)	$(19.8)_{-}$
Cyanophyceae	38698	31876	32008	35158
	(45.2)	(40.9)	(37.4)	(26.2)
Bacillariophyceae	13324	9416	19685	55590
	(15.5)	(7.3)	(20.0)	(41.5)
Englenophyceae	23554	28925	13567	16706
	(27.5)	(37.2)	(15.8)	(12.5)
Dinophyceae	-	-	-	-
T. S. abun. of Phyto.	85473	77776	85478	134057

Tab.5: Total Number of genera of phytoplankton belonging to different groups of algae in the study ponds (% composition in parenthesis).

Group/Classes	Pond-1	Pond-2	Pond- 3	Pond- 4
Chlorophyceae	12	9	11	11
	(36.3)	(34.6)	(36.7)	(36.7)
Cyanophyceae	7	7	7	8
	(21.2)	(26.9)	(23.3)	(26.6)
Bacillariophyceae	10	6	10	9
	(30.3)	(23)	(33.3)	(30)
Euglenophyceae	3	3	2	2
	(9)	(11.5)	(6.6)	(6.6)
Dinophyceae	. 1	1	-	-
	(3)	(3.8)		
T. Phyto. genera	33	26	30	30

Tab.6: Yearly abundance of different groups of phytoplankton in four ponds (units/1) and their % composition (in parenthesis).

Phytoplankton groups	Pond-1	Pond-2	Pond-3	Pond-4
Chlorophyceae	16242	35858	13098	16122
	(23.75)	(30.84)	(13.87)	(16.8)
Cyanophyceae	16159	24086	39347	34334
	(23.63)	(20.72)	(41.7)	(35.78)
Bacillariophyceae	10557	28089	27290	24365
-	(15.44)	(24.17)	(28.9)	(25.38)
Euglenophyceae	24868	24757	14639	21124
	(36.37)	(21.29)	(15.5)	(20.01)
Dinophyceae	537	3444	-	-
	(0.78)	(2.96)		
Total Plankton	68363	116243	94374	95945

Tab.7. Seasonal abundance (unit-1) and percentage composition (in parenthesis) of phyto-zooplankton in four study ponds (p₁, p₂, p₃ & p₄).

Season	Plankton	Pond	Pond	Pond	Pond
		1	2	3	4
Summer	Phyto plankton	66044	107883	114396	85473
		(41.8)	(22.3)	(32.8)	(16.7)
March-May	Zoo Plankton	91923	375451	234049	424278
		(58.2)	(77.7)	(67.2)	(83.3)
Monsoon .	Phyto plankton	82225	92858	72746	77776
		(41.7)	(26.6)	(25.9)	(38.8)
June-Aug.	Zoo Plankton	114981	255918	208150	122378
_		(58.3)	(73.4)	(74.1)	(61.2)
Post-monsoon	Phyto plankton	66123	145523	105668	85478
		(17.4)	(35.5)	(43.6)	(26.8)
SeptNov	Zoo Plankton	313183	264824	136820	233468
•		(82.6)	(64.5)	(56.4)	(73.2)
Winter	Phyto plankton	57161	116341	83704	134057
		(9.5)	(20.8)	(37.9)	(55.4)
Dec-Feb.	Zoo Plankton	541917	442213	136957	107953
		(90.5)	(79.2)	(62.1)	(44.6)

Tab.8. Comparative seasonal abundance of phyto-zooplankton in the study ponds (P₁, p₂, p₃ & p₄) unit/1.

Season	Plankton	Pond	Pond	Pond	Pond
	,	1	2	3	4
Summer	Phytoplankton	66044	107883	114396	85473
March-May	Zooplankton	91923	375451	234049	424278
Monsoon	Phytoplankton	82225	92858	72746	77776
June-Aug.	Zooplankton	114981	255918	208150	122378
Post-monsoon	Phytoplankton	66123	145523	105668	85478
SeptNov	Zooplankton	313183	264824	136820	233468
Winter	Phytoplankton	57161	116341	83704	134057
Dec-Feb.	Zooplankton	541917	442213	136820	107953

	,		av_d	sdd	tds	econ	ph	co2	co3	hco3	th	cl	do
	et 		1.10	.20	489.50	753.00	7.70	21.50	.00	30.50	209.0	112.00	6.10
1	34.8	31.80	1.10	.20	494.30	751.00	8.30	9.70	.00	23.50	295.0	121.00	5.20
2	29.1	29.90	1.10	.20	484.30	745.00	8.30	10.10	.00	21.00	287.0	103.00	6.30
3	32.1	31.10	1.20	.20	454.70	700.00	8.30	2.80	17.20	39.00	200.0	143.00	6.80
4	33.6	31.50	.90	.10	353.00	543.00	8,50	26.70	.00	114.1	209.0	102.00	6.90
5	30.3	30.40	,90	.20	456.00	702.00	8.70	23.50	.00	165.3	267.0	109.00	6.30
6	27.6	27.70 48.10	1.20	.20	488.00	752.00	8.60	13.00	.00	135.0	285.0	99.00	5.70
7	23.1	19.30	1.00	.20	450.50	693.00	8.60	27.30	.00	184.0	386.0	87.00	5.40
8	20.2	19.60	1.00	.30	440.40	678.00	8.60	15.70	2.80	231.0	467.3	107.00	7.60
9	17.5	23.90	1.00	.20	484.30	745.00	8.60	33.20	.00	301.2	440.0	107.00	4.00
10	23.8	29.00	1.00	.20	444.00	683.00	8.70	34.70	.00	346.0	429.0	122.00	5.40
11	32.4			.20	420.20	647.00	8.70	28.00	.00	403.0	538.0	131.00	5.40
12	38.3	30.50	1	1	I	ı	1	1	1	1	1		

c:\spsswin\a2.sav

	o2	bod	cod	nh3	po4	eh	rh2	gpp	npp	tzp	tpp
1	82.00	6.90	15.30	.10	.10	.36	27.8	.20	.10	456596.0	92154.00
2	68.00	7.60	18.80	.10	.10	.29	26.6	.20	10	120309.0	107062.0
3	86.00	5.40	11.90	.10	.10	.29	26.6	.30	.04	48770.00	80894.00
4	92,00	12.00	22.50	.10	.20	.29	26.6	.30	.02	180686.0	65568.00
5	92.00	3.40	8.40	.10	.10	.27	26.3	.20	.04	172147.0	41360.00
6	81.00	3.10	7.20	.10	.20	.25	26.0	.30	.03	117931.0	74868.00
7	69.00	4.90	10.60	.11	.10	.25	25.8	.20	11	699633.0	81201.00
8	60.00	5.60	12.20	.10	.10	.26	26.1	.20	14	101347.0	44360.00
9	86.00	7.00	15.20	.13	.70	.26	26.1	.40	.10	1303536	56847.00
10	45.00	3.20	9.80	.10	.20	.26	26.1	.30	.12	231114.0	72967.00
11	70.00	3.10	8.50	.10	.70	.25	26.0	.40	.14	52059.00	111365.0
12	73.00	2.70	7.40	.00	.20	.25	26.0	.30	.08	18302.00	57703.00
13											

- 1

TOB-18.

- - Correlation Coefficients - -

TA	AV_D	BOD	Cr	CO2	CO3
1.0000 .2659 .0067 .6634* 0191 .1620 0062 0471 1276 .3837 0364 0559 6301* .3461 .3406 3651 2228	.2659 1.0000 .5875* .41016490* .4474 .57090717 .5382 .31531809310714572792 .039138002447	.0067 .5875* 1.0000 .4024 8027** .8002** .9735** .3958 .2897 .4996 0927 6333* .3384 2513 .3859 5285 0745	.6634* .4101 .4024 1.00002661 .6289* .4048 .05300173 .1346 .3474 .08464194 .3580 .27341216 .1969	01916490*8027**2661 1.000057547620**471234483334 .1635 .7438**2782 .40335050 .3949 .2478	.1620 .4474 .8002** .6289* 5754 1.0000 .7010* .3807 0146 .1441 .1898 2791 .1082 .0176 .4159 1593 .0603
2228 .3969 4338 1237 2021 .2866 6053* .3220	2447 .2401 .2132 .5357 1465 .2927 .0884 .5940*	0745 .4632 .2842 .2970 4465 .0467 .2322	.1969 .1458 .0702 0095 .0177 .2947 2239 .1283	.2478262223133532 .5115095924453451	.0603 .1263 .1204 0205 3127 1561 .0452 .0204
	1.0000 .2659 .0067 .6634* 0191 .1620 0062 0471 1276 .3837 0364 0559 6301* .3461 .3406 3651 2228 .3969 4338 1237 2021 .2866 6053*	1.0000 .2659 .2659 1.0000 .0067 .5875* .6634* .4101 01916490* .1620 .4474 0062 .5709 04710717 1276 .5382 .3837 .3153 03641809 05593107 6301*1457 .34612792 .3406 .0391 36513800 22282447 .3969 .2401 4338 .2132 1237 .5357 20211465 .2866 .2927 6053* .0884	1.0000	1.0000	1.0000

	COD	DO	ECON	EH	GPP	HCO3
AT	-,0062	0471	1276	.3837	0364	0559
	.5709	0717	.5382	.3153	1809	3107
AV_D BOD	.9735**	.3958	.2897	.4996	-,0927	6333*
CL	.4048	.0530	0173	.1346	.3474	.0846
-						
CO2	7620**	4712	3448	3334	.1635	.7438**
CO3	.7010*	.3807	0146	.1441	.1898	2791
COD	1.0000	.2702	.3551	.5396	1170	6203*
DO	.2702	1.0000	4067	. 15 68	.1871	3699
ECON	.3551	4067	1.0000	.3250	0656	2906
EH	.5396	.1568	.3250	1.0000	3802	6610*
GPP	1170	.1871	0656	3802	1.0000	.5386
HCO3	6203*	3699	2906	6610*	.5386	1.0000
NH3_N	.3412	.3081	.2461	.1562	.0193	4880
NPP	2429	.0932	1932	.1073	.6770*	.4239
02	.2649	.9126**	4032	.3115	.0951	4644
PH	5651	1599	3586	9950**	.4201	.7051*
PO4	0678	.2445	1542	3386	.8656**	.5301
RH2	.5068	.1512	.2852	.9939**	3318	6049*
SDD	.3025	.1570	.4747	0672	.5657	.1892
TDS	.3698	4138	.9990**	.3310	0789	3029
TH	3851	3612	0625	5606	.5467	.8727**
TPP	.1569	3463	.6188*	.2777	.1277	1406
TZP	.2214	.4793	.0948	.0176	.2199	0161
WT	0382	0310	.2061	.0758	3703	2662
	0362	0310	.2001	.0756	.5,05	2002

[&]quot; \cdot " is printed if a coefficient cannot be computed

	инз_11	NPP	02	PH	PO4	RH2
- 100	6301*	.3461	.3406	3651	2228	.3969
ΛT	1457	2792	,0391	3800	2447	.2401
VA D	.3384	2513	.3859	5285	0745	.4632
BOD	4194	.3580	.2734	1216	.1969	.1458
CP CP	2782	. 4033	5050	.3949	.2478	2622
go2 go3	.1082	.01.76	.4159	1593	.0603	.1263
CCD	.3112	2429	.2649	5651	0678	.5068
DO	.3081	.0932	.9126**	1599	.2445	.1512
ECON	.2461	1932	4032	3586	1542	.2852
EH	.1562	. 1073	.31.15	9950**	3386	.9939**
GPP	.0193	.6770*	.0951	.4201	.8656**	3318
нсоз	4880	.4239	4644	.7051*	.5301	6049*
NH3 N	1.0000	1389	.1023	1827	.2092	.1270
NPF	1389	1.0000	.1610	0481	.5456	.1705
02	1.023	.1610	1.0000	3146	.0633	.3029
PH	1827	0481	3146	1.0000	.3740	9779**
FO4	.2092	.5456	.0633	.3740	1.0000	2953
RH2	.1270	. 1705	.3029	9779**	2953	1.0000
SDD	.2051	. 1355	0917	.0743	.5738	0569
TDS	.2460	2078	4064	3645	1608	.2911
TH	4325	. 2837	5161	.5982*	.5284	5117
TPP	.1537	.1218	1841	2913	.1799	.2580
TZP	.4821	.0823	.1712	0401	.4380	0057
WT	0971	2543	.1504	1538	3614	0135
		Corre	elation Coef	riolents	- -	
	SDD	TDS	ТН	TPP	TZP	WT
AΤ	4338	1237	-,2021	.2866	6053*	.3220
AV D	.2132	.5357	1465	.2927	.0884	.5940*
BOD	.2842	.2970	4465	.0467	.2322	.0121
CL	.0702	0095	.0177	.2947	2239	.1283
CO2						
CO3	2313	3532	.5115 ·	0959	2445	3451
	.1204	0205	.5115 · 3127	0959 1561	2445 .0452	3451 .0204
COD	.1204 .3025	0205 .3698	.5115 · 3127 3851	0959 1561 .1569	2445 .0452 .2214	3451 .0204 0382
COD DO	.1204 .3025 .1570	0205 .3698 4138	.5115 ·312738513612	0959 1561 .1569 3463	2445 .0452 .2214 .4793	3451 .0204 0382 0310
ECON DO COD	.1204 .3025 .1570 .4747	0205 .3698 4138 .9990**	.5115 - 3127 3851 3612 0625	0959 1561 .1569 3463 .6188*	2445 .0452 .2214 .4793 .0948	3451 .0204 0382 0310 .2061
EH DO COD	.1204 .3025 .1570 .4747 0672	0205 .3698 4138 .9990** .3310	.5115 · .31273851361206255606	0959 1561 .1569 3463 .6188* .2777	2445 .0452 .2214 .4793 .0948 .0176	3451 .0204 0382 0310 .2061 .0758
COD ECON EH COD	.1204 .3025 .1570 .4747 0672 .5657	0205 .3698 4138 .9990** .3310 0789	.5115 · .31273851361206255606 .5467	0959 1561 .1569 3463 .6188* .2777	2445 .0452 .2214 .4793 .0948 .0176 .2199	3451 .0204 0382 0310 .2061 .0758 3703
COD FCON EH GPP HCO3	.1204 .3025 .1570 .4747 0672 .5657 .1892	0205 .3698 4138 .9990** .3310 0789 3029	.5115 · .31273851361206255606 .5467 .8727**	09591561 .15693463 .6188* .2777 .12771406	2445 .0452 .2214 .4793 .0948 .0176 .2199 0161	3451 .0204 0382 0310 .2061 .0758 3703 2662
HE3_N ECON ECON ECON ECON	.1204 .3025 .1570 .4747 0672 .5657 .1892 .2051	0205 .3698 4138 .9990** .3310 0789 3029 .2460	.511531273851361206255606 .5467 .8727**4325	09591561 .15693463 .6188* .2777 .12771406 .1537	2445 .0452 .2214 .4793 .0948 .0176 .2199 0161	3451 .0204 0382 0310 .2061 .0758 3703 2662 0971
NED HCO3 HCO3 NH3 N NED HCO3	.1204 .3025 .1570 .4747 0672 .5657 .1892 .2051	0205 .3698 4138 .9990** .3310 0789 3029 .2460 2078	.511531273851361206255606 .5467 .8727**4325 .2837	09591561 .15693463 .6188* .2777 .12771406 .1537 .1218	2445 .0452 .2214 .4793 .0948 .0176 .2199 0161 .4821	3451 .0204 0382 0310 .2061 .0758 3703 2662 0971 2543
COD LO ECON EH GPP HCO3 IH3_N NFP	.1204 .3025 .1570 .4747 0672 .5657 .1892 .2051 .1355	0205 .3698 4138 .9990** .3310 0789 3029 .2460 2078 4064	.511531273851361206255606 .5467 .8727**4325 .28375161	09591561 .15693463 .6188* .2777 .12771406 .1537 .12181841	2445 .0452 .2214 .4793 .0948 .0176 .2199 0161 .4821 .0823 .1712	3451 .0204 0382 0310 .2061 .0758 3703 2662 0971 2543 .1504
COD LO ECON EH GPP HCO3 IH3_N NFP O2 PH	.1204 .3025 .1570 .4747 0672 .5657 .1892 .2051 .1355 0917	0205 .3698 4138 .9990** .3310 0789 3029 .2460 2078 4064 3645	.511531273851361206255606 .5467 .8727**4325 .28375161 .5982*	09591561 .15693463 .6188* .2777 .12771406 .1537 .121818412913	2445 .0452 .2214 .4793 .0948 .0176 .2199 0161 .4821 .0823 .1712 0401	3451 .0204 0382 0310 .2061 .0758 3703 2662 0971 2543 .1504 1538
COD ECON ECON EH GPP HCO3 IH3_N NFP O2 PH FO4	.1204 .3025 .1570 .4747 0672 .5657 .1892 .2051 .1355 0917 .0743	0205 .3698 4138 .9990** .3310 0789 3029 .2460 2078 4064 3645 1608	.511531273851361206255606 .5467 .8727**4325 .28375161 .5982* .5284	09591561 .15693463 .6188* .2777 .12771406 .1537 .121818412913 .1799	2445 .0452 .2214 .4793 .0948 .0176 .2199 0161 .4821 .0823 .1712 0401 .4380	3451 .0204 0382 0310 .2061 .0758 3703 2662 0971 2543 .1504 1538 3614
COD ECON EH GPP HCO3 IH3_N NFP O2 PH FO4 RH2	.1204 .3025 .1570 .4747 0672 .5657 .1892 .2051 .1355 0917 .0743 .5738 0569	0205 .3698 4138 .9990** .3310 0789 3029 .2460 2078 4064 3645 1608 .2911	.511531273851361206255606 .5467 .8727**4325 .28375161 .5982* .52845117	09591561 .15693463 .6188* .2777 .12771406 .1537 .121818412913 .1799 .2580	2445 .0452 .2214 .4793 .0948 .0176 .2199 0161 .4821 .0823 .1712 0401 .4380 0057	3451 .0204 0382 0310 .2061 .0758 3703 2662 0971 2543 .1504 1538 3614 0135
COD DO ECON EH GPP HCO3 IH3_N NFP O2 PH FO4 RH2 SDD	.1204 .3025 .1570 .4747 0672 .5657 .1892 .2051 .1355 0917 .0743 .5738 0569 1.0000	0205 .3698 4138 .9990** .3310 0789 3029 .2460 2078 4064 3645 1608 .2911 .4670	.511531273851361206255606 .5467 .8727**4325 .28375161 .5982* .52845117	09591561 .15693463 .6188* .2777 .12771406 .1537 .121818412913 .1799 .2580 .1477	2445 .0452 .2214 .4793 .0948 .0176 .2199 0161 .4821 .0823 .1712 0401 .4380 0057 .6458*	3451 .0204 0382 0310 .2061 .0758 3703 2662 0971 2543 .1504 1538 3614 0135 3125
COD DO ECON EH GPP HCO3 IH3_N NFP O2 PH FO4 RH2 SDD TDS	.1204 .3025 .1570 .4747 0672 .5657 .1892 .2051 .1355 0917 .0743 .5738 0569 1.0000 .4670	0205 .3698 4138 .9990** .3310 0789 3029 .2460 2078 4064 3645 1608 .2911 .4670 1.0000	.511531273851361206255606 .5467 .8727**4325 .28375161 .5982* .52845117 .48440668	09591561 .15693463 .6188* .2777 .12771406 .1537 .121818412913 .1799 .2580 .1477 .6328*	2445 .0452 .2214 .4793 .0948 .0176 .2199 0161 .4821 .0823 .1712 0401 .4380 0057 .6458*	3451 .0204 0382 0310 .2061 .0758 3703 2662 0971 2543 .1504 1538 3614 0135 3125 .2004
COD DO ECON EH GPP HCO3 NH3_N NPP O2 PH FO4 RH2 SDD TDS TH	.1204 .3025 .1570 .4747 0672 .5657 .1892 .2051 .1355 0917 .0743 .5738 0569 1.0000 .4670 .4844	0205 .3698 4138 .9990** .3310 0789 3029 .2460 2078 4064 3645 1608 .2911 .4670 1.0000 0668	.511531273851361206255606 .5467 .8727**4325 .28375161 .5982* .52845117 .48440668 1.0000	09591561 .15693463 .6188* .2777 .12771406 .1537 .121818412913 .1799 .2580 .1477 .6328*1000	2445 .0452 .2214 .4793 .0948 .0176 .2199 0161 .4821 .0823 .1712 0401 .4380 0057 .6458* .0842 .1263	3451 .0204 0382 0310 .2061 .0758 3703 2662 0971 2543 .1504 1538 3614 0135 3125 .2004 4097
COD DO ECON EH GPP HCO3 NH3_N NFP O2 PH FO4 RH2 SDD TDS TH TEF	.1204 .3025 .1570 .4747 0672 .5657 .1892 .2051 .1355 0917 .0743 .5738 0569 1.0000 .4670 .4844 .1477	0205 .3698 4138 .9990** .3310 0789 3029 .2460 2078 4064 3645 1608 .2911 .4670 1.0000 0668 .6328*	.511531273851361206255606 .5467 .8727**4325 .28375161 .5982* .52845117 .48440668 1.00001000	09591561 .15693463 .6188* .2777 .12771406 .1537 .121818412913 .1799 .2580 .1477 .6328*1000 1.0000	2445 .0452 .2214 .4793 .0948 .0176 .2199 0161 .4821 .0823 .1712 0401 .4380 0057 .6458* .0842 .1263 1402	3451 .0204 0382 0310 .2061 .0758 3703 2662 0971 2543 .1504 1538 3614 0135 3125 .2004
COD DO ECON EH GPP HCO3 NH3_N NPP O2 PH FO4 RH2 SDD TDS TH	.1204 .3025 .1570 .4747 0672 .5657 .1892 .2051 .1355 0917 .0743 .5738 0569 1.0000 .4670 .4844	0205 .3698 4138 .9990** .3310 0789 3029 .2460 2078 4064 3645 1608 .2911 .4670 1.0000 0668	.511531273851361206255606 .5467 .8727**4325 .28375161 .5982* .52845117 .48440668 1.0000	09591561 .15693463 .6188* .2777 .12771406 .1537 .121818412913 .1799 .2580 .1477 .6328*1000	2445 .0452 .2214 .4793 .0948 .0176 .2199 0161 .4821 .0823 .1712 0401 .4380 0057 .6458* .0842 .1263	3451 .0204 0382 0310 .2061 .0758 3703 2662 0971 2543 .1504 1538 3614 0135 3125 .2004 4097 .3257

[&]quot; . " is printed if a coefficient cannot be computed

	Watt	AV DEPT	SEC D D	TDS	ECOND	Ph	FCO2	CO3alk
Air T	32.8	1.7	0.2	404.3	622	7.8	14.7	0
34.8	29.7	1.6	0.2	418	643	8.2	8.2	0
28.6	31	1.5	0.2	455	700	8.4	8.9	0
32	32.1	1.8	0.2	383	589	8.2	8.6	4.8
33.6	31	1,5	0.2	344	529	8.4	19	0.9
30.3	27.4	1.4	0.2	499	767	8.7	8.3	2.7
27.6	23.1	1.4	0.2	435	669	8.7	11.3	2.8
23.1	18.4	1.4	0.4	482	741	8.7	32.5	0
20.2	18.5	1.6	0.2	440.4	677.5	8.8	4.9	7.3
17.5	24	1.4	0.1	406.9	709	8.9	13.1	10.8
23.8	29.2	1.4	0.3	436.2	671	8.9	12.02	5.3
32.6 38.7	31.8	1.8	0.2	408.9	629	8.8	4.3	15.2
30.7								

O3 alk	Th	Chloride	Do	%Sato2	Bods	COD	NH3	PO4
21.2	285.5	138.7	6	83	3.2	7.3	0.1	0.1
18.8	307	121.6	6.1	80	2.3	6.7	0.1	0.1
18.7	227.5	191.7	6.3	85.3	5.8	14.5	0.1	0.1
41.3	226.2	133.7	7.6	105	9.6	20.7	0.1	0.1
109	214.2	136.9	8.4	113	3.9	8.7	0.1	0.1
160.5	317.7	129.8	8.1	104	8.7	12.8	0.1	0.02
141.3	311	146.7	6.8	81	3.6	8.2	0.11	0.1
171.4	414.6	104.6	5	53.2	3.7	10.1	0.12	0.1
195.3	487	121.7	7.7	86.5	3.4	8.7	0.1	0.1
290.1	464.8	120.6	7.1	85.5	2.9	7.3	0.1	0.03
329.7	465.2	103.6	6.8	90	2.5	7.1	0.1	0.1
356.7	609.2	136.6	8.4	114.4	3.7	8.4	0.1	0.1

H my	rH2	GPPmg	NPPmg	Tphyt	T-Zoo
0.35	27.67	0.3	-0.03	154817	523934
0.31	27.09	0.3	0.1	86425	197565
0.28	26.46	0.4	-0.1	102144	291268
0.31	27.09	0.4	0.1	85595	278907
0.28	26.46	0.3	0.01	147148	155918
0.25	26.02	0.3	0.01	122285	183888
0.25	26.02	0.3	0.21	167096	452816
0.25	26.02	0.1	-0.16	112705	264092
0.24	25.88	0.4	0.05	121479	945002
0.23	25.73	0.3	0.15	123803	140551
0.22	25. 39	0.2	-0.07	88257	178785
0.24	25.88	0.3	0.02	83682	423637

- - Correlation Coefficients - -

	AIR_T	AV_DEPT	BODS	CHLORID	CO3ALK	COD
AIR_T AV_DEPT BODS CHLORID CO3ALK COD DO ECO2 ECOND EH_MV GPPMG HCO3_ALK NH3 NPPMG PH PO4 RH2 SATO2_PR SEC_D_D T_ZOO TDS TH TPHYT	AIR_T 1.0000 .5714 .1934 .3178 .1448 .1942 .252404383555 .3900 .16950434511115814152 .2009 .3629 .5940*158231474025 .09973499	AV_DEPT .5714 1.0000 .2053 .1619 .2938 .3055 .267222856317* .5387 .477019903974 .10355107 .4219 .5645 .43252294 .35474790 .26673452	BODS .1934 .2053 1.0000 .30110971 .9143** .337913300293 .1842 .399531741432 .007216893044 .1980 .395110041640 .159420571751	CHLORID .3178 .1619 .3011 1.00002033 .3684 .078906642620 .3248 .6249*48472585 .00983407 .1347 .3074 .22033984 .066606202813 .1636	.1448 .2938 0971 2033 1.0000 1253 .5578 3426 .0291 5535 .1500 .7942** 2905 .3222 .5842* 2262 5178 .4349 3608 .1873 1629 .3148 3148	COD .1942 .3055 .9143** .36841253 1.0000 .156713121191 .2440 .430839510614066523020217 .2563 .248000261181 .045522113137
TTAW	.9427**	.5195	.2482	.4082	0480	.2347

	DO	ECO2	ECOND	EH_MV	GPPMG	HCO3_ALK
	0504	0420	2555	2000	1.005	0434
AIR_T	.2524	0438	3555	.3900	.1695	0434
AV_DEPT	.2672	2285	6317*	. 5387	.4770	1990
BODS	.3379	1330	0293	.1842	.3995	3174
CHLORID	.0789	0664	2620	. 32 48	.6249*	4847
COBALK	.5578	3426	.0291	~. 5535	.1500	.7942**
COD	.1567	1312	1191	.2440	.4308	3951
DO	1.0000	.1341	3516	2615	.4827	.3427
ECO2	.1341	1.0000	4756	.0892	2799	1243
ECOND	3516	4756	1.0000	5812*	4196	.3953
EH MV	2615	.0892	5812*	1.0000	.2998	8457**
GPPMG	.4827	2799	4196	.2998	1.0000	3669
HCO3 ALK	.3427	1243	.3953	8457**	3669	1.0000
инз	5898*	.1747	.2800	1963	6860*	.0249
NPPMG	.3696	2109	3080	0082	.4416	.0069
PĦ	.2699	1209	.5698	9964**	2802	.8440**
PO4	2673	.1481	4441	.3196	.0000	2555
RH2	2497	.0571	5898*	.9964**	.3189	8411**
SATO2 PR	.9200**	.1416	4728	.0019	.5038	.1739
SEC D D	5197	.1804	.3531	1457	7426**	.0767
T_ZOO	.0733	3235	1044	.0193	.3766	0022
TDS	~.3737	4769	.8704**	3719	3087	.1111
TH	.1034	2530	.1806	3191	2523	. 4503
ТРНҮТ	0270	.3850	1838	.1368	0165	2110
TTAW	.2641	.0531	4663	.5525	.3209	2795

[&]quot; • " is printed if a coefficient cannot be computed

70b-28

1000		Corre	lation Coef	ficients		
	инз	NPPMG	PH	PO4	RH2	SATO2_PR
AIR_T	5111	~.1581	4152	.2009	.3629	.5940*
AV_DEPT	3974	.1035	5107	.4219	.5645	. 4325
BODS	1432	.0072	1689	3044	.1980	.3951
CHLORID	2585	.0098	3407	.1347	.3074	.2203
CHROKE	2905	.3222	.5842*	2262	5178	. 4349
COD	0614	0665	2302	0217	.2563	.2480
po	5898*	.3696	.2699	2673	2497	.9200**
ECO2	.1747	2109	1 20 9	.1481	.0571	.1416
ECOND	.2800	3080	.5698	4441	5898*	4728
EH MV	1963	0082	9964**	.3196	.9964**	.0019
GPPMG	6860*	.4416	2802	.0000	.3189	.5038
HCO3_ALK	.0249	.0069	.8440**	2555	8411**	.1739
NH3	1.0000	2513	.2029	.1874	1898	7120**
NPPMG	2513	1.0000	.0573	2244	.0408	.2598
PH	.2029	.0573	1.0000	3425	9857**	0056
PO4	.1874	2244	3425	1.0000	.2958	1449
RH2	1898	.0408	9857**	.2958	1.0000	0002
SATO2_PR	7120**	. 2598	0056	1449	0002	1.0000
SEC_D_D	.7132**	~.7016*	.1172	.4111	1740	5155
T Z.00	0180	.0965	0048	.3517	.0350	1020
TDS	.4201	3607	.3724	3318	3698	5001
TH	.1587	3278	.3115	.4199	3233	.0439
TPHYT	.2228	. 25 42	1507	1090	.1208	1388
WATT	6309*	~.0408	5770*	.1442	.5247	.6180*
		Corre	lation Coef	ficients		
	SEC_D_D	T_Z00	TDS	тн	ТРНҮТ	WATT
AIR T	1582	3147	4025	.0997	3499	.9427**
AV DEPT	2294	.3547	~.4790	.2667	3452	.5195
BODS	1004	1640	.1594	2057	1751	.2482
CHLORID	3984	.0666	0620	2813	.1636	.4082
					2442	0.400

	SEC_D_D	T_ZOO	TDS	ТН	трнүт	WATT
AIR_T AV_DEPT BODS CHLORID CO3ALK COD DO ECO2 ECOND EH_MV GPPMG HCO3_ALK NH3 NPPMG PH	1582 2294 1004 3984 3608 0026 5197 .1804 .3531 1457 7426** .0767 .7132** 7016*	T_ZOO 3147 .35471640 .0666 .18731181 .073332351044 .0193 .376600220180 .09650048	40254790 .159406201629 .045537374769 .8704**37193087 .1111 .42013607 .3724	.0997 .2667 2057 2813 .3148 2211 .1034 2530 .1806 3191 2523 .4503 .1587 3278 .3115	349934521751 .1636314831370270 .38501838 .136801652110 .2228 .25421507	.9427** .5195 .2482 .40820480 .2347 .2641 .05314663 .5525 .320927956309*04085770*
PO4 RH2 SATO2_PR SEC_D_D T_Z© TDS TH TPHYT WATT	.4111 1740 5155 1.0000 0589 .4251 .5047 1888 3137	.3517 .0350 1020 0589 1.0000 .0782 .4625 .2319 3750	3318 3698 5001 .4251 .0782 1.0000 .2904 0996 4969	.41993233 .0439 .5047 .4625 .2904 1.000032931418	1090 .1208 1388 1888 .2319 0996 3293 1.0000 2428	.1442 .5247 .6180* 3137 3750 4969 1418 2428 1.0000

[&]quot; . " is printed if a coefficient cannot be computed

a:\aziz2.sav

		•					
wat_t	av_dep	sec_d_d	tds	econd	ph	eco2	co3all
32.40	2.20	.20	397.20	611.00	8.20	1.30	4
29.50	2.10	.20	412.10	634.00	8.50	2.00	1
31.10	2.10	.30	419.30	645.00	8.40	2.80	1
31.90	2.30	.20	381.60	587.00	8.40	4.30	1
29.20	1.70	.20	387.70	597.00	8.60	6.30	3
29.00	2.00	.20	389.40	599.00	98.60	2.90	1
22.90	1.70	.20	448.50	690. 0 0	8.60	3.80	3
19.00	1.70	.20	447.90	689.00	8.60	4.00	4
20.10	1.90	.10	450.10	693.00	8.70	4.50	8
24.30	1.80	.30	441.70	680.00	8.80	6.40	1
29.30	1.80	.2ძ	445.90	686.00	8.70	14.60	1
32.50	2.00	.20	423.20	651.00	8.90	1.00	2
	32.40 29.50 31.10 31.90 29.20 29.00 22.90 19.00 20.10 24.30 29.30	32.40 2.20 29.50 2.10 31.10 2.10 31.90 2.30 29.20 1.70 29.00 2.00 22.90 1.70 19.00 1.70 20.10 1.90 24.30 1.80 29.30 1.80	32.40 2.20 .20 29.50 2.10 .20 31.10 2.10 .30 31.90 2.30 .20 29.20 1.70 .20 29.00 2.00 .20 22.90 1.70 .20 19.00 1.70 .20 20.10 1.90 .10 24.30 1.80 .30 29.30 1.80 .26	32.40 2.20 .20 397.20 29.50 2.10 .20 412.10 31.10 2.10 .30 419.30 31.90 2.30 .20 381.60 29.20 1.70 .20 387.70 29.00 2.00 .20 389.40 22.90 1.70 .20 448.50 19.00 1.70 .20 447.90 20.10 1.90 .10 450.10 24.30 1.80 .30 441.70 29.30 1.80 .2d 445.90	32.40 2.20 .20 397.20 611.00 29.50 2.10 .20 412.10 634.00 31.10 2.10 .30 419.30 645.00 31.90 2.30 .20 381.60 587.00 29.20 1.70 .20 387.70 597.00 29.00 2.00 .20 389.40 599.00 22.90 1.70 .20 448.50 690.00 19.00 1.70 .20 447.90 689.00 20.10 1.90 .10 450.10 693.00 24.30 1.80 .30 441.70 680.00 29.30 1.80 .2d 445.90 686.00	32.40 2.20 .20 397.20 611.00 8.20 29.50 2.10 .20 412.10 634.00 8.50 31.10 2.10 .30 419.30 645.00 8.40 31.90 2.30 .20 381.60 587.00 8.40 29.20 1.70 .20 387.70 597.00 8.60 29.00 2.00 .20 389.40 599.00 98.60 22.90 1.70 .20 448.50 690.00 8.60 19.00 1.70 .20 447.90 689.00 8.60 20.10 1.90 .10 450.10 693.00 8.70 24.30 1.80 .30 441.70 680.00 8.80 29.30 1.80 .20 445.90 686.00 8.70	32.40 2.20 .20 397.20 611.00 8.20 1.30 29.50 2.10 .20 412.10 634.00 8.50 2.00 31.10 2.10 .30 419.30 645.00 8.40 2.80 31.90 2.30 .20 381.60 587.00 8.40 4.30 29.20 1.70 .20 387.70 597.00 8.60 8.30 29.00 2.00 .20 389.40 599.00 98.60 2.90 22.90 1.70 .20 448.50 690.00 8.60 3.80 19.00 1.70 .20 447.90 689.00 8.60 4.00 20.10 1.90 .10 450.10 693.00 8.70 4.50 24.30 1.80 .30 441.70 680.00 8.80 6.40 29.30 1.80 .20 445.90 686.00 8.70 14.60

P-93

a:\aziz2.sav

ı	air t	wat_t	av_dep	sec_d_d	tds	econd	ph	eco2	co3alk
	air_t 33.80	32.40	2.20	.20	397.20	611.00	8.20	1.30	4.80
1	30.40	29.50	2.10	.20	412.10	634.00	8.50	2.00	1.80
2	30.40	31.10	2.10	.30	419.30	645.00	8.40	2.80	1.30
3	32.60	31.90	2.30	.20	381.60	587.00	8.40	4.30	1.20
4	28.50	29.20	1.70	.20	387.70	597.00	8.80	6.30	3.20
5	27.40	29.00	2.00	.20	389.40	599.00	98,60	2.90	1.70
6	22.80	22.90	1.70	.20	448.50	690.00	8.60	3.80	3.30
8	19 80	19.00	1.70	.20	447.90	689.00	8.60	4.00	4.30
9	19.00	20.10	1.90	.10	450.10	693.00	8.70	4.50	8.00
10	24.30	24.30	1.80	.30	441.70	680.00	8.80	6.40	1.40
11	32.60	29.30	1.80	.2ď	445.90	686.00	8.70	14.60	1.80
12	35.10	32.50	2.00	.20	423.20	651.00	8.90	1.00	2.70
12						057,00	0.00		

a:\aziz2.sav

	hco3_alk	th	chloride	do	sato2_pe	bods	cod	nh3	po4
	25.60	240.20	132.40	7.70	105.40	9.00	19.60	.10	01
1					101,90	13.60	28.70	.10	.03
2	15.10	243.00	142.00	7.70				.10	.01
3	16.70	228.70	103.40	7.30	98.90	13.00	25.80		
4	81.30	246.80	113.60	7.00	95.50	9.40	22.30	.10	.03
5	113.60	238.70	107.50	9.60	126.00	8.70	14.50	.10	.01
6	134.80	247.50	132.10	6.80	89.10	12.80	27.70	.10	.01
 7	114.20	271.70	121.60	5.90	70.00	7.40	16.50	.11	.02
8	156.40	434.30	188.70	9.10	101.20	7.60	18.90	.13	.03
9	218.80	424.00	201.80	8.20	93.00	3.40	9.30	.10	.01
10	283.70	466.20	178.00	5.70	69.20	3.20	8.70	.10	.01
11	298 20	500.00	140.70	7.10	93.40	11.60	26.30	.10	.03
12	145.20	608.80	156.70	8.30	114.20	10.60	20.80	.10	.01

a:\aziz2.sav

	eh_mv	rh2	gpp_mg	npp_mg	tphyt	t_zoo	t_plankt
1	.31	27.09	.40	.03	550865.0	493846.0	1043911
2	.27	25.71	.30	.03	154842.0	134422.0	289264.0
3	.28	26.46	.40	.10	308555.0	386944.0	699643.0
4	.28	26.46	.30	01	75986,00	98937.00	174923.0
5	.26	26.17	.30	.02	513447.0	200732.0	714179.0
6	.26	26.17	.30	.11	143227.0	495292.0	638519.0
7	.26	26.17	.20	.10	277906.0	450040.0	427946.0
8	.26	26.17	.30	.10	386860.0	130745.0	518605.0
9	.25	25.02	.30	.10	137376.0	483978.0	321334.0
10	.23	25.52	.30	20	52232.00	84973.00	137205.0
11	.25	26.02	.30	.10	65528.00	143484.0	209012.0
12	.23	25.73	.20	.03	145126.0	65613.00	210739.0

	AIR T	AV DEP	BODS	CHLORIDE	COBALK	COD
MIR T NV THEP SOFS CHLORIDE COSALK COD	AIR T 1.0000 .6156* .6324* .5886* .5252 .5804*0035	.6156* 1.0000 .4744 .32782170 .46401311	.6324* .4744 1.0000 .5482 5606 .9832**	5886* 3278 5482 1.0000 .5689 5011 .1233	5252 2170 5606 5689 1.0000 5566	.5804* .4640 .9832** .50115566 1.00000318
DO BCO2 BCOND EH MV CPE M' HCO3 ALK HH3 NPP MC PH CO4 EH2 SATO2 PE	0232 5791* .2955 .1688 .0164 5633 0927 0390 .0212 .4828	4511 6090* .5813* .4390 .3678 4794 0517 .0847 .0114 .3773 .1743	1226 3990 .3300 .1716 .2781 1719 .4454 .3359 .2834 .4092 .2768	.0396 .6509* 5062 1995 .5101 .3663 1473 1051 .0080 6740* 2221	1525 .3500 .0577 .0076 .0579 .2252 .3146 1994 2430 2842 .1296	0397 3828 .3598 .1811 .2999 0987 .4585 .3605 .4130 .4165
SEC D D T PLANKT T ZOO TOS TH TENYT WAT T	.2417 .0850 2589 5768* 0604 0064 .9661**	.0499 .1178 .0903 6068* 3578 1210 .6778*	.1896 .1597 0117 3946 2787 0579 .6434*	3497 .3752 1328 .6495* .6692* 2846 6952*	7341** .2363 .4622 .3478 .1534 .33045465	.1381 .1150 0382 3783 2828 1054 .5770*

[&]quot; . " is printed if a coefficient cannot be computed

10 Dec 98 STSS for MS WINDOWS Release 6.0

-- Correlation Coetticients --

	DO	ECO2	ECOND	EH_MV	GPP_MG	HCO3_ALK
ATR_T AV_DEP BODS CHLORIDE COSALK COD DO ECOX ECOND EH_MV GFT_MG HCO3_ALK NHS NFF_MG PH FO4 BUS SATO2_PE SEC_D_D T_FLANKT T_ZOO	00351311 .0111 .1253 .42960318 1.000012391764 .0950 .10360049 .2698 .321519850081 .0010 .9839** .4145 .31361571	0232 4511 .1226 .0396 1525 0397 1239 1.0000 .3344 3140 .0292 .3205 0614 0201 1364 .3431 1707 1717 .0102 3517 2792	ECOND 5791*6090* .3990 .6509* .350038281764 .3344 1.00005183 .3148 .4497 .4269 .08343662 .140054124973 .064340180465 1.0000**	EH_MV .2955 .5813* .33905062 .0577 .3598 .086031405183 1.0000 .6825*8402*0310 .1066 .7909** .2256 .0133 .7058* .45015163	GFP_MG .1688 .4380 .1716 -1995 .0076 .1811 .103602923148 .6825* 1.00005934*1698 .000000521562 .4735 .1884 .2928 .5968* .30593149	HCO3_ALK .01643678 .2781 .5101 .057929990049 .3205 .44978402** .5934* 1.0000039920910507151654990513 .10355820*4235 .4490
TDS TH TPHYT WAT T	1776 .0719 .5395 0235	.3331 .3133 3069 1310	1.0000** .6159* 2553 7190**	5163 7594** .5760 .3655	3149 4682 .3813 .2551	.9397** 4788 1215

^{*} Signif. LE .05 ** Signif. LE .01 (2 tailed)

[&]quot; . " is printed it a coefficient cannot be computed

10 Dec 98 STSS for MS WINDOWS Release 6.0

1000

	инз	NFP_MG	PH	FO4	RH2	SATO2_PE
	-,5633	0927	0390	.0212	_4828	_4437
ATP T	4794	0517	.0847	.0114	,3773	.1743
AV_DEP	.1719	.4454	.3359	.2834	.4092	.2768
eons.	.3663	- 1473	.1051	.0080	67401	2221
CHLORIDE	.2252	.3146	1994	2430	2842	.1296
COBATK	0987	.4585	.3605	.4130	.4165	.2213
COD	.2698	.3215	1985	0081	.0010	.8839**
DO	0614	0201	1364	.3431	1707	1717
ECO2	. 4269	.0834	- 3662	_1400	5412	4873
ECOND	0309	.2761	0310	.1065	.7909**	.2256
EH WA	.1698	.0000	.0052	.1562	.4735	.1884
GET_MG	0399	2091	0507	1516	5499	0513
HCO3_ALK	1.0000	.2702	1190	.4244	.0877	0785
NH3	.2702	1.0000	.2423	.1701	.1669	.2343
NPP MG	1180	.2423	1.0000	2455	.0618	1457
PH POA	.4244	.1701	2455	1.0000	.0668	0840
PO4 BH2	.0877	.1669	.0618	.0668	1.0000	_2210
SATC2_PE	0785	.2343	1457	0840	.2210	1.0000
SEC D_D	.0663	.4910	.0515	.1372	.3033	.2347
T PLANKT	.0692	.3378	.2092	4142	-6846 ¹	.3666
T ZOO	1218	.4993	.4015	4771	.2190	1994
TDS	.4288	.9861	3653	.1427	5381	4878
TH	.1473	1687	2277	.0053	5478	0475
TEHYT	.2975	.2324	1701	2536	.5996*	.5062
WA'T_T	6526*	0891	-0896	1200	.5121	.4439

Signif. LE .05 ** Signif. LE .01 (2 tailed)

[&]quot;. " is printed it a coefficient cannot be computed

	SEC_D_D	T_FLANKT	T_Z00	TDS	TH	TPHYT
- 17	.2417	.0850	2589	5768*	0604	0064
ATP T	.0499	.1178	.0903	6068*	3578	1210
AV_DEP	.1896	.1597	.0117	.3946	.2787	.0579
BODS	3497	3752	1328	.64954	.66921	2846
CHLORIDE	7341**	.2363	.4622	.3478	.1534	.3304
СОЗУРК	.1381	.1150	0382	3783	2028	1054
COD	4145	.3136	1571	1776	.0719	.5395
DO 2	.0102	3517	2792	.3331	.3133	3069
econii Ecos	0643	~_401R	0465	1.0000**	.6159*	2553
EII_IIV	0133	.7059*	.4501	5163	7594**	.5760
Grr_MG	.2928	.5968*	.3059	.3149	.4682	.3813
HCO3 ALK	1035	5820*	- 4235	.4490	.9397**	4788
кни	0663	.0692	1219	.4288	.1478	.2975
NPP MG	4910	.3378	.4993	.0861	1687	.2324
FH	0515 .	.2092	.4015	3653	2277	1701
PO4	1372	4142	4771	.1427	.0053	2536
RH2	.3033	- 6846*	.2190	5381	5478	.5996*
SATO2_FE	2347	.3666	1994	4878	0475	.5062
SEC D D	1.0000	.0422	.2709	.0633	.0998	.0112
T PLANKT	.0422	1.0000	.6359*	4013	57781	.869111
TZOO	2709	.6359*	1.0000	0465	4790	.3416
TDS	0633	4013	0465	1.0000	.6150*	2555
TH	0998	5778+	4790	.6158*	1.0000	4291
TPHYT	0112	.8691**	.3416	2555	4291	1.0000
WAT_T	.2840	.1941	- 1229	7173**	2425	-0389

Signif. LE .05 ** Signif. LE .01 (2 tailed)

[&]quot;. " is printed it a coefficient cannot be computed

```
WAT_T
             .9661**
AIR T
             .6778*
AV DEP
             .6434+
BODS
CHLORIDE
            -.6952<sup>1</sup>
            -.5465
созурк
             .5770*
COP
            -.0235
DO
            -.1310
ECO2
            -.7190**
ECOMP
             .3655
EH MV
            .2551
GTT MG
            -.1215
HCOS ALK
            -.6526*
инз
            -.0891
NPP MG
             .0896
PH
            -.1200
FO4
             .5121
p112
             .4439
SATO2_PE
             .2840
SEC D_D
             .1941
T FLANKT
TZOO
            -.1229
            -.7173**
TDS
            -.2425
TH
             .0389
TPHYT
            1 0000
мат т
                                                    (2 tailed)
                              Signif. LE .01
* Signif. LE .05
```

Signif. LE .05 "" Signif. LE .01 (2 Carre

[&]quot; . " is printed if a coefficient cannot be computed

					=00ND	Dh	FCO2	CO3alk
Air T	Wat t	AV DEPT	SEC D D	TDS	ECOND	Ph		
-	32.5	1.6	0.2	362.7	558	8.4	2.3	4.9
33.8	_					8.5		2.7
30.4	29.7	1.6	0.2	382.5	589			2.3
-	30.9	1.8	0.2	403.3	621	8.1		
30.2	31.6	2	0.2	354.9	546	8.4		5. 8
32.6				_	585	8.6		4.8
28.5	29.6	1.4	0.3	380.3				1.8
27.4	28.7	1.4	0.2	370.8	571	8.6	4	
-	23	1 .5	0.2	404.6	623	8 .5	4.3	2.5
22.8	19.1	1.7	0.1	470.6	724	8.6		5. 3
19.3						8.7	5.4	4.3
19	20.3	1.4	0.2	397.5	612		J. 4	
24.4	24.4	1.4	0.2	451.4	695	8.9		7.7
-	29.4	1.6	0.2	378.6	583	8.8		6.6
32.6								6.6
35.1	32.6	1.9	0.2	351.3	541	8.9		0.0

HCO3 alk 21.3 16.3 15.2 78 113.8 130.4 102 133 174 251	Th 268 272 197.8 221 216 232 228.3 340.3 217.3 377.2 409	Chloride 98.6 132 137 109.2 113.6 142 137 121 98.6 93.4 110.6	Do 7.1 6.3 6.4 6.7 10.6 7.9 6.5 7.9 7.9 8.2 7.7	%Sato2 97.4 83.7 86.5 91 140 101.2 77.5 88 90 100 101	Bods 12.6 16.8 13.2 14.7 12.4 12.5 16.8 12.7 13.6 9.3 8.6	COD 25.8 37.2 26.8 31.2 26.2 27.3 33.9 22.8 24.8 20.7 17.3	NH3 0.1 0.1 0.1 0.1 0.1 0.1 0.1 0.1 0.1	PO4 0.4 0.3 0.7 1 0.1 0.2 0.2 0.1 0.1 0.2 0.02
251	409	110.6	7.7	101	8.6	17.3	0.1	0.02
336.6	469	121.6	8.2	113	13	25.9	0.1	0.2

EH mv 0.28 0.27 0.32 0.28 0.26 0.25 0.27 0.26	rH2 26.46 26.31 27.23 26.46 26.17 26.02 26.31 26.17	GPPmg 0.3 0.3 0.4 0.4 0.4 0.3 0.3	NPPmg 0.12 -0.04 -0.07 0.03 0.09 0.05 0.09	Tphyt 69637 86424 90806 56087 86301 54158 115951	T-Zoo 626306 67525 199468 100145 182921 356296 161200
	26.46	0.4	0.03	56087	100145
	26.17	0.4	0.09	86301	182921
	26.02	0.3	0.05	54158	356296
	26.31	0.3	0.09	115951	161200
	26.17	0.3	0.13	127948	116635
0.25	26.02	0.3	-0.02	120642	99110
0.23	25.73	0.2	-0 .16	151470	109471
0.24	25.88	0.2	-0.17	72043	3706673
0.23	25.75	0.3	-0 .05	65385	284857

Correlation Coefficients - -

	AIR T	AV_DEPT	BODS	CHLORIDE	CO3ALK	COD
AIR_T AV_DEFT BODS	1.0000 .5270 0970	.5270 1.0000 .2062	0970 .2062 1.0000 1965	1150 .0827 1965 1.0000	.1641 .1961 6371* 1288	.0803 .1641 .9440**
CHLORIDE CO3ALK COD	1150 .1641 .0803 1186	.0827 .1961 .1641 4173	6371* 9440** 4770	1288 3052 .1663	1.0000 6315* .3765	6315* 1.0000 4698
DO ECOND EH_MV FCO2	3938 .1382 9793*	0935 .3469 8411	5125 .4594 .3434	.7693** .0999 .1263	.3269 5810* 2768	6590* .4374 .0653 .4869
GPPMG HCO3_ALK NH3	.1552 0149 2022	.3896 0389 1624 0057	.5206 5626 5059 .0065	.2155 .1908 .0871 .7209**	4237 .6715* .5054 0470	5950* 6059* 1546
NPPMG PH PO4	2394 1317 .4077 .1487	3044 .6790* .3602	4936 .3328 .4144	1166 .0311 .1330	.6489* 1236 5675	4845 .4017 .3918
RH2 SATO2_PR SEC_D_D T ZOO	.3032 .3558 .2999	1962 3097 0149	- 4402 - 0261 - 5814*	.1208 0541 .0315	.3586 0553 .3186	3515 .1328 5514
TDS TH TPHYT	7629** .2484 2218	2959 .2095 4390 .3574	2033 5221 0948 .2161	.0312 0869 .3460 4721	.1268 .7145** 4434 1407	3046 5162 0098 .4353
WATT	.8499++	. 33/9	. 2 101	. 4121	1 1 10 /	. 4303

		24.00			anmea	WG02 11 W
	DO	ECOND	EH_MV	FCO2	GPPMG	HCO3_ALK
AIR T	1186	3938	.1382	9793*	.1552	0149
AV DEPT	4173	0935	.3469	8411	.3896	0389
BODS	4770	5125	.4594	.3434	.5206	5626
CHLORIDE	.1663	.7693**	.0999	.1263	.2155	.1908
CO3ALK	.3765	.3269	5810*	2768	4237	.6715*
COD	4698	6590*	.4374	.0653	.4869	5950*
DO	1.0000	.3367	5140	.3587	.0211	. 4435
ECOND	.3367	1.0000	1215	.7872	0754	.3521
EH MV	5140	1215	1.0000	7793	.6934*	8581**
FCO2	.3587	.7872	7793	1.0000	.7273	.9613*
GPPMG	.0211	0754	.6934*	.7273	1.0000	5715
HCO3 ALK	.4435	.3521	8581**	.9613*	5715	1.0000
NH3	.2029	.3072	7324**	7273	7596**	.8650**
NPPMG	.2980	. 6965 *	.1333	8794	.3610	0800
PH	.5146	.1460	9887**	. 9054	7028*	.8985**
PO4	5236	1180	.6794*	9909**	.6124*	5061
RH2	5030	0924	.9953**	7824	.6838*	8148**
SATO2_PR	.9045**	.1366	3734	4062	.1575	.3577
SEC_D_D	.4866	1860	.0000	7273	.3189	0400
T_ZCO	.0233	.0495	2625	9610*	5119	. 3445
TDS	.0557	. 4308	0752	. 7848	3113	.0566
TH	.1725	.0837	6918*	9799*	6660*	.7988**
ТРНҮТ	.1004	.1026	1974	.1003	1259	.0189
TTAW	2359	7858**	.2497	9402	.2276	2933

^{* -} Signif. LE .05 ** - Signif. LE .01 (2-tailed)

[&]quot; . " is printed if a coefficient cannot be computed

- - Correlation Coefficients - -

	AIR_T	AV_DEPT	BODS	CHLORIDE	CO3ALK	COD
AIR_T AV_DEFT RODS CHLORIDE CO3ALK COD DO ECOND EH_MV FCO2 GPPMG HCO3_ALK NH3 NPPMG PH PO4 RH2 SATO2_PR SEC_D_D T_ZOO TDS TH TPHYT	1.0000 .5270 0970 1150 .1641 .0803 1186 3938 .1382 9793* .1552 0149 2022 2394 1317 .4077 .1487 .3032 .3558 .2999 7629** .2484 2218	.5270 1.0000 .2062 .0827 .1961 .164141730935 .34698411 .38960389162400573044 .6790* .36021962309701492959 .20954390	0970 .2062 1.0000 1965 6371* .9440** 4770 5125 .4594 .3434 .5206 5626 5059 .0065 4936 .3328 .4144 4402 0261 5814* 2033 5221 0948	1150 .0827 1965 1.0000 1288 3052 .1663 .7693** .0999 .1263 .2155 .1908 .0871 .7209** 1166 .0311 .1330 .1208 0541 .0315 .0312 0869 .3460	.1641 .1961 6371* 1288 1.0000 6315* .3765 .3269 5810* 2768 4237 .6715* .5054 0470 .6489* 1236 5675 .3586 0553 .3186 .1268 .7145** 4434	.0803 .1641 .9440** 3052 6315* 1.0000 4698 6590* .4374 .0653 .4869 5950* 6059* 1546 4845 .4017 .3918 3515 .1328 5514 3046 5162 0098
$W\Lambda TT$.8499**	.3574	.2161	4721	1407	.4353

	DO	ECOND	eh_mv	FCO2	GPPMG	HCO3_ALK
AIR T	1186	3938	.1382	9793*	.1552	0149
AV DEPT	4173	0935	.3469	~.8411	.3896	0389
BODS	4770	5125	.4594	.3434	.5206	5626
CHLORIDE	.1663	.7693**	.0999	.1263	.2155	.1908
CO3ALK	.3765	.3269	~.5810*	2768	4237	.6715*
COD	4698	6590*	.4374	.0653	.4869	5950*
DO	1.0000	.3367	5140	.3587	.0211	.4435
ECOND	.3367	1.0000	1215	.7872	0754	.3521
EH MV	5140	1215	1.0000	~.7793	.6934*	8581**
FCO2	.3587	.7872	7793	1.0000	.7273	.9613*
GPPMG	.0211	0754	.6934*	.7273	1.0000	5715
HCO3 ALK	.4435	.3521	8581**	.9613*	5715	1.0000
NH3	.2029	.3072	7324**	7273	7596**	.8650**
NPPMG	.2980	.6965*	.1333	8794	.3610	0800
PH	.5146	.1460	9887**	.9054	7028*	.8985**
FO4	5236	1180	.6794*	9909**	.6124*	 50 6 1
RH2	5030	0924	.9953**	7824	.6838*	8148**
SNTO2_PR	.9045**	.1366	3734	4062	.1575	.3577
SEC_D_D	.4866	1860	.0000	7273	.3189	0400
T_ZCO	.0233	.0495	2625	9610*	5119	.3445·
TDS:	.0557	.4308	0752	.7848	3113	.0566
ТН	.1725	.0837	6918*	9799*	6660*	.7988**
ТРНҮТ	.1004	.1026	1974	.1003	1259	.0189
WATT	~.2359	7858**	.2497	9402	.2276	2933

[&]quot; \cdot " is printed if a coefficient cannot be computed

Tas-48

-- Correlation Coefficients --

	инз	NPFMG	PH	PO4	RH2	SATO2_PR
	•••			4077	.1487	.3032
**************************************	-,2022	2394	1317	. 4077	.3602	1962
AIR_T	-,1624	0057	3044	.6790*	.4144	4402
AV DEPT	5059	.0065	1936	.3328	.1330	.1208
BODS	.0871	.7209**	1166	.0311	5675	.3586
CHLORIDE	.5054	0470	.6189*	1236		3515
COSVIK	6059*	1546	4845	.4017	.3918	.9045**
COD	.2029	.2980	.5146	5236	5030	.1366
po	.3072	.6965*	.1460	1180	0924	3734
ECOND	7324**	.1333	9887**	.6794*	.9953**	4062
EH_MV	7273	8794	.9054	9909**	7824	.1575
FCO2	7596**	.3610	7028*	.6124*	.6838*	
GPPMG	.8650**	0800	.8985**	5061	8148**	.3577
HCO3_VTK	1.0000	1913	.7927**	6698*	6845*	.0430
NH3	1913	1.0000	1774	.0354	.1105	.1938
NPPMG	.7927**	1774	1.0000	6773*	9756**	.3749
FH	6698*	.0354	6773*	1.0000	.6722*	2942
PO4	6845*	.1105	9756**	.6722*	1.0000	3585
RH2	.0430	.1938	.3749	2942	3585	1.0000
SATO2_FR	2874	0552	.0000	.0000	.0000	.6737*
SEC_D_D		2510	.2852	3003	2475	.0847
T_Z00	.4012	.0394	.1178	3134	0600	2855
TDS	.2899	2778	.7453**	-,4131	~.6564*	.1889
TH	.7953**	.1621	.0816	1962	1955	.0170
TYHTT	0462	5246	2595	.3856	.2456	.1537
TYPAW	3963	5240	12.550	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,		
Correlation Coefficients						
	SEC_D_D	T_Z00	TDS	ТН	ТРНҮТ	WATT
AIR T	.3558	.2999	7629**	.2484	2218	.8499**
AV DEFT	3097	0149	2959	.2095	4390	.3574
BODS	0261	5814*	2033	5221	0948	.2161
CHLORIDE	0541	.0315	.0312	0869	.3460	4721
COSALK	0553	.3186	.1268	.7145**	4434	1407
COD	.1328	5514	3046	5162	0098	. 4353
DO	.4866	.0233	.0557	.1725	.1004	2359
ECOND	1860	.0495	.4308	.0837	.1026	7858**
EH MV	.0000	2625	0752	~,6918*	1974	.2497
FCO2	7273	9610*	.7848	9799*	.1003	9402
GPPMG	.3189	5119	3113	6660*	1259	.2276
HCO3 ALK	0400	.3445	.0566	.7988**	.0189	2933
инз	2874	.4012	.2899	.7953**	~.0462	3963
NPPMG	0552	2510	.0394	2778	.1621	5246
PH	.0000	.2852	.1178	.7453**	.0816	2595
PO4	.0000	3003	3134	4131	1962	.3856
RH2	.0000	2475	0600	6564*	1955	.2456
SATO2 PR	.6737*	.0847	2855	.1889	.0170	.1537
SEC D D	1.0000	.0075	5254	2959	0672	.3817
T_ZOO	.0075	1.0000		.4298	1256	.0695
TDS			1539	.0974	.0103	6787*
T'H	5254	1539	1.0000			
ТРНҮТ	2959	.4298	.0974	1.0000	1912	.0221
WATT	0672	1256	.0103	1912	1.0000	1466
	.3817	.0695	6787*	.0221	1466	1.0000

Rajshahi Chare 16. ... Dodenne ot 16. ... D. 75. Down 12: 12:12:199

[&]quot; . " is printed if a coefficient cannot be computed